

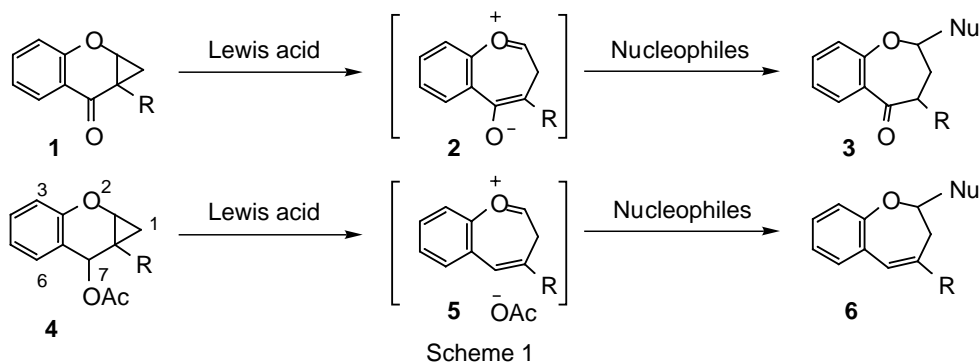
**LEWIS ACID-MEDIATED RING EXPANSION REACTION OF BENZO[*b*]CYCLOPROPA[*e*]PYRAN-7-OL ACETATES: FACILE SYNTHESIS OF 2-ALKYL SUBSTITUTED 2,3-DIHYDRO-1-BENZOXEPINS**

Yoshiaki Sugita,\* Hiroki Hosoya, and Ichiro Yokoe

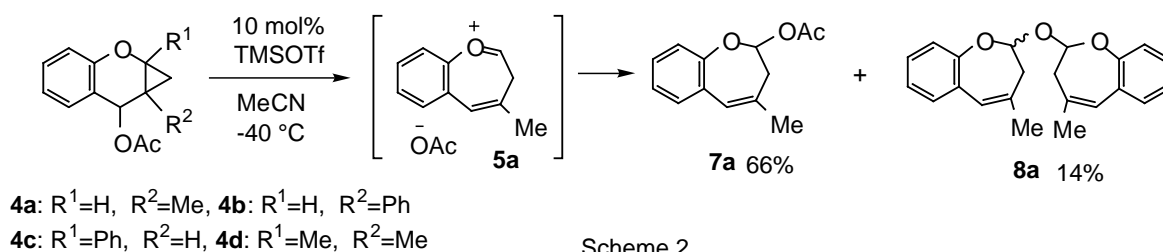
Faculty of Pharmaceutical Sciences, Josai University, Sakado,  
Saitama 350-0295, Japan

**Abstract** - In the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf), benzo[*b*]cyclopropa[*e*]pyran-7-ol acetates easily reacted with silyl enol ethers to give 2-alkyl substituted 2,3-dihydro-1-benzoxepin derivatives in good yields.

Cyclopropanes having an electron-withdrawing or donating group are susceptible to ring opening reactions.<sup>1</sup> We have recently reported that benzocyclopropapyranone (**1**) was transformed into a ring-opened 1,3-zwitterion (**2**) in the presence of a Lewis acid, and that **2** reacted with silyl enol ethers to give the ring expanding product (**3**).<sup>2</sup> During the course of our study to find further applications of benzocyclopropapyrans, we examined the reaction of 7-acetoxybenzo[*b*]cyclopropa[*e*]pyran (**4**) with nucleophiles. We expected that **4** was transformed into a cyclic oxonium ion intermediate *via* removal of the acetoxy group by the action of a Lewis acid, and that the intermediate may react with nucleophiles to provide several 2-alkyl substituted 2,3-dihydro-1-benzoxepins (**6**) as the ring-expanded products (Scheme 1).<sup>3</sup> Although there are many examples of the synthesis of 2,3-dihydro-1-benzoxepins,<sup>4</sup> only a few examples are reported for the construction of those having an alkyl group at the 2-position on the ring, and no systematic study has been reported.<sup>5</sup> We now report the synthesis of 2-alkyl substituted 2,3-dihydro-1-benzoxepins by the Lewis acid-promoted reaction of 7-acetoxybenzo[*b*]cyclopropa[*e*]pyran derivatives (**4**) with silyl enol ethers *via* a cyclic oxonium ion.<sup>6</sup> As the reactions of **4**, only a few solvolytic ring opening reactions have been reported.<sup>7</sup> To our knowledge, there has been no publication concerning the carbon-carbon bond forming reactions of **4** under the Lewis acid-promoted conditions.



7-Acetoxybenzo[*b*]cyclopropano[*e*]pyran (**4**) was synthesized by the NaBH<sub>4</sub> reduction of the corresponding benzocyclopropapyranone derivatives,<sup>7,8</sup> prepared from chromones and dimethyloxosulfonium methylide, followed by the treatment of the resulting alcohols with Ac<sub>2</sub>O/Et<sub>3</sub>N in the presence of a catalytic amount of DMAP. First, the reactivity of 7-acetoxy-1,1a,7,7a-tetrahydro-7a-methylbenzo[*b*]cyclopropano[*e*]pyran (**4a**) by the action of a Lewis acid was examined. A solution of a catalytic amount of TMSOTf in MeCN was added to a solution of **4a** in MeCN at -40 °C to give the 2-acetoxy-2,3-dihydro-4-methyl-1-benzoxepin (**7a**) and the dimeric compound (**8a**) in 66% and 14% yields, respectively (Scheme 2). A similar tendency was observed with TiCl<sub>4</sub> as the Lewis acid. It was found that **4a** is very unstable under acidic conditions and was converted into **7a** even when **4a** was processed by silica gel column chromatography. From this result, we considered that **4a** is the equivalent of a cyclic oxonium ion (**5a**) under the Lewis acid-promoted conditions.



We next examined the carbon-carbon bond-forming reactions of the cyclopropane (**4**) with silyl enol ethers as the nucleophile. The reaction of **4a** with silyl enol ether (**9**) was chosen as the model. When the reaction was performed in the presence of TiCl<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -40 °C, however, **7a** was obtained as the major product and the desired adduct was obtained in low yield (Table 1, Entry 1). In order to improve of the yield of **6b**, we examined the various reaction conditions such as the addition order of the reagents, solvent, and Lewis acid. As a Lewis acid, TMSOTf worked nicely as compared to a typical Lewis acid such as TiCl<sub>4</sub> or SnCl<sub>4</sub> (Entries 4 and 5). The use of MeCN as a polar solvent increased the

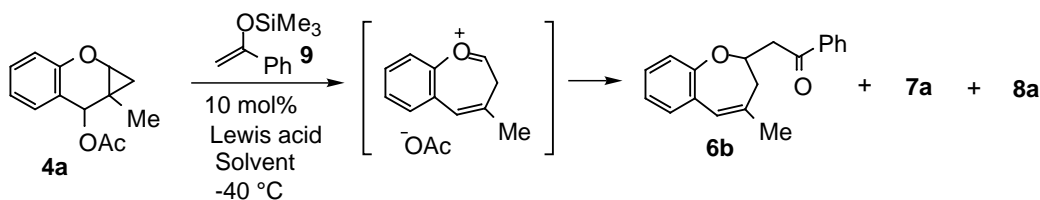


Table 1. Effects of Lewis Acid and Solvent

Entry	Lewis acid	Solvent	Products (%) <sup>a</sup>		
			<b>6b</b>	<b>7a</b>	<b>8a</b>
1	TiCl <sub>4</sub>	CH <sub>2</sub> Cl <sub>2</sub>	2	30	6
2	SnCl <sub>4</sub>	CH <sub>2</sub> Cl <sub>2</sub>	15	58	0
3	SnCl <sub>4</sub>	MeCN	60	27	0
4	TMSOTf	CH <sub>2</sub> Cl <sub>2</sub>	66	trace	7
5	TMSOTf	MeCN	99 (66) <sup>b</sup>	0 (31) <sup>b</sup>	0

<sup>a</sup> Isolated yield. <sup>b</sup> To a solution of **4a** and **9** in MeCN was added a solution of TMSOTf in MeCN.

yield of the desired adduct compared with the use of  $\text{CH}_2\text{Cl}_2$  (Entries 4 and 5). It was also found that the order of the addition of the reagents dramatically influenced the yield. While the desired adduct (**6b**) was produced in quantitative yield by the slow addition of a dilute solution of **4a** in MeCN to a mixture of **9** and TMSOTf in MeCN, the decrease in the yield of **6b** from 99% to 66% was observed and **7a** was obtained in 31% yield when TMSOTf was added to a solution of **4a** and **9** in MeCN (Entry 5).<sup>9</sup> Several examples of the TMSOTf-catalyzed ring expansion reactions were examined and the results are summarized in Table 2. As expected, the reaction of **4a** with several silyl enol ethers smoothly proceeded to give the corresponding benzoxepins in high yields (Entries 1~6). As for the cyclopropanes, **4b** as well as **4a** effectively reacted with silyl enol ethers to give the benzoxepins in good yields (Entries 7~12). Under the same reaction conditions, the cyclopropane (**4c**) having a phenyl group at the 1a-position also gave the 2,2-disubstituted benzoxepins (**6m** and **6n**) in moderate yields (Entries 13 and 14). Furthermore, the dialkyl substituted cyclopropane (**4d**) reacted with silyl enol ethers to give the 2,2,4-trialkyl-2,3-dihydro-1-benzoxepins in good yields (Entries 15 and 16).

Table 2. Reaction of 7-Acetoxy-1,1a,7,7a-tetrahydrobenzo[*b*]cyclopropano[*e*]pyrans (**4**) with Silyl Enol Ethers<sup>a</sup>

Entry	<b>4</b>	Silyl enol ether	Product	Yield (%) <sup>b</sup>	Entry	<b>4</b>	Silyl enol ether	Product	Yield (%) <sup>b</sup>
1	<b>4a</b>			81	9	<b>4b</b>			97 <sup>c</sup>
2	<b>4a</b>			99	10	<b>4b</b>			82 <sup>c</sup>
3	<b>4a</b>			81 <sup>c</sup>	11	<b>4b</b>			83 <sup>c</sup>
4	<b>4a</b>			91 <sup>c</sup>	12	<b>4b</b>			85 <sup>c</sup>
5	<b>4a</b>			89 <sup>c</sup>	13	<b>4c</b>			70
6	<b>4a</b>			80 <sup>c</sup>	14	<b>4c</b>			36
7	<b>4b</b>			68	15	<b>4d</b>			52
8	<b>4b</b>			85	16	<b>4d</b>			70

<sup>a</sup> All reactions were carried out in dry MeCN at  $-40\text{ }^\circ\text{C}$  in the presence of 10 mol% of TMSOTf. <sup>b</sup> Isolated yield. <sup>c</sup> Two diastereomers were formed in the ratio of 1 : 1.

In summary, we have demonstrated that the TMSOTf-catalyzed ring opening addition reactions of **4** with silyl enol ethers smoothly proceeded to afford the corresponding 2,3-dihydro-1-benzoxepins in good yields. We are now investigating the Lewis acid-mediated ring expansion reactions of **4** with various nucleophiles, and the results will be reported in due course.

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9. General procedure for the reaction of **4** with silyl enol ethers: This is exemplified by the reaction of **4a** with silyl enol ether (**9**) using TMSOTf as a Lewis acid. To a stirred solution of silyl enol ether (**9**) (192 mg, 1 mmol) and TMSOTf (11 mg, 0.05 mmol) in MeCN (4 mL) was dropwise added a solution of **4a** (109 mg, 0.5 mmol) in MeCN (1 mL) over a 30 min period at -40 °C under an argon atmosphere. After being stirred for 30 min, the reaction was quenched at the same temperature by adding saturated aqueous NaHCO<sub>3</sub> (2 mL). The mixture was vigorously stirred for 10 min and allowed to warm to rt. The mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (20 mL x 3), the combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>, and the solvent was evaporated under reduced pressure. The residue was purified by silica gel column chromatography (eluent: hexane-AcOEt = 10 : 1) to give the benzoxepin (**6b**) (138 mg, 99%).