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**A RING-OPENING CROSS-METATHESIS REACTION OF  
N-TRIALKYLSILYL 2-AZABICYCLO[2.2.1]HEPT-5-EN-3-ONE WITH  
ALLYLTRIMETHYLSILANE**

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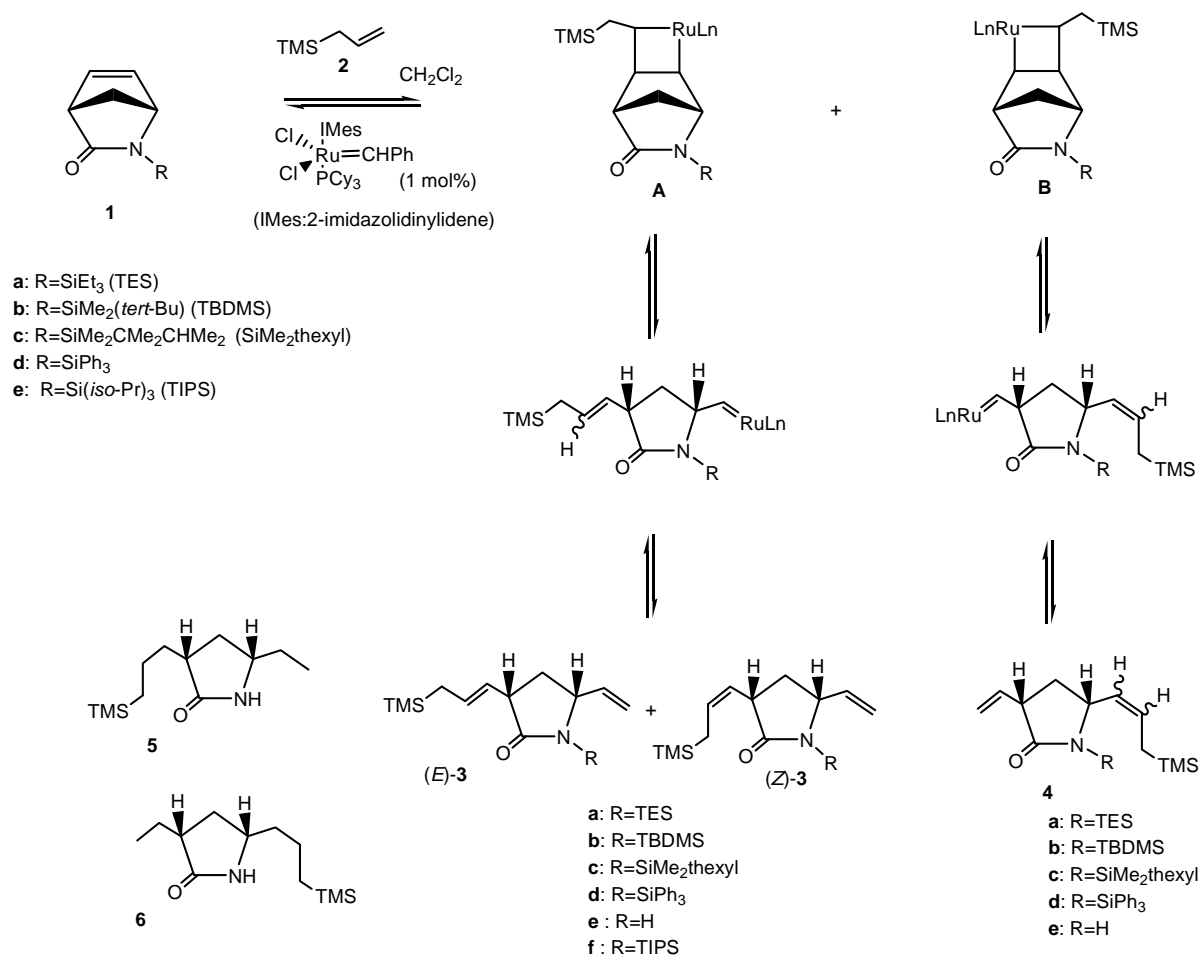
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**Abstract** – Subjection of *N*-trialkylsilyl-2-azabicyclo[2.2.1]hept-5-en-3-one (**1**) to a ring-opening cross-metathesis reaction with allyltrimethylsilane in the presence of Grubbs' catalyst was found to allow the predominate formation of pyrrolidine (**3**) over pyrrolidine (**4**).

The use of 2-azabicyclo[2.2.1]hept-5-en-3-one (ABH) (**1**;R=H) has proven to be highly successful in the preparation of a number of carbocyclic nucleosides,<sup>1</sup> and we have previously reported the first preparation of 2',3'-methano-, 2',3'-epimino- and 2',3'-oxirane-fused carbocyclic nucleosides based on ABH.<sup>2</sup> In spite of its attractive chemical lability, the development of the synthetic potential of ABH still remains challenging. Among only a few hitherto known reports of ruthenium-catalyzed metathesis reactions of ABH,<sup>3</sup> we have re-examined the known metathesis reaction of *N*-Boc-ABH (**1**;R=Boc) with allyltrimethylsilane (**2**) in the presence of Grubbs' catalyst in order to gain more detailed knowledge of the exemplified high regioselectivity.<sup>3a</sup> However, as a result, we observed insufficiency in regioselectivity in the ring-opening of ABH (**1**), leading to a pair of regioisomeric ring-opening products (through **A** and **B** to **3** and **4**, respectively), which has become a crucial issue to be overcome<sup>4</sup> (Scheme 1). The transannular participation of the nitrogen and carbon at the 6 position in ABH (**1**;R=H) has been recognized during bromination of **1** (R=H), leading to dibromide by way of addition-rearrangement.<sup>5</sup> Thus, we reasoned that this nitrogen participation would be essential to achieve regioselectivity successfully. First, therefore, we began by establishing the effects of *N*-substituents in ABH on the regioselectivity of the metathesis reaction. In our previous report, we observed low regioselectivity of the reaction of *N*-acyl ABH with **2**,<sup>4a,b</sup> but *N*-trialkylsilyl group in ABH (**1**) was eventually found to be

significantly effective in promoting the metathesis reaction in a regioselective manner.<sup>4c</sup> Encouraged by these results, we turned our attention to the isolation of metathesis products (**3,4**), and their conversion to pyrrolizines (**8,9**).

After the reaction of **1a** with **2** (1.2 equiv.) was successfully and cleanly carried out using Grubbs' second generation catalyst (1 mol%) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 2h under argon atmosphere, the reaction mixture was concentrated and directly subjected to separation by medium pressure liquid chromatography (MPLC) with hexane-AcOEt, allowing the isolation of (*Z*)-**3a** and (*E*)-**3a** as major products and a minor amount of **4a** as a *E/Z* mixture<sup>6</sup> (Scheme 1). Catalytic hydrogenation (4 atm of H<sub>2</sub>, 10% Pd on C in THF) of (*E*)-**3a** and (*Z*)-**3a**, accompanied by spontaneous *N*-desilylation during the reduction, provided **5** as a sole product. Otherwise, **6** was the sole product of catalytic hydrogenation of an inseparable *E/Z* mixture of **4a**.



Scheme 1

Similarly, formation of **3b,c** predominated over **4b,c** in the reaction of *N*-trialkylsilyl ABH (**1b,c**) with **2**. These products (**3a-c** and **4a-c**) were somewhat less stable, accompanied by decomposition to some

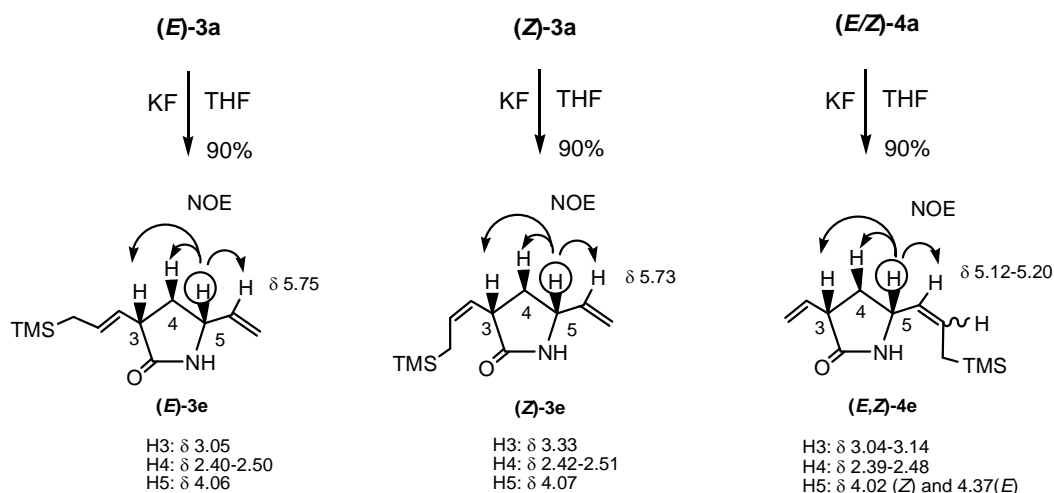
extent during the separation and handling, but enough to be isolated and characterized. However, probably due to their intrinsic instability, only (*E*)-**3f** was isolated from the reaction mixture of *N*-triisopropylsilyl ABH (**1e**) with **2** in low yield (Table 1). On the other hand, when *N*-triphenylsilyl ABH (**1d**) was exposed to the reaction with **2**, it appeared that *E/Z* mixture of **4d** was more likely to be obtained than (*Z*)-**3d** and (*E*)-**3d**.

Table 1 Isolation of **3** and **4**\*

<b>1</b>	Isolated yield (%) of <b>3</b>	Isolated yield (%) of <b>4</b>
<b>1a</b>	46% [( <i>E</i> )- <b>3a</b> ], 14% [( <i>Z</i> )- <b>3a</b> ]	10% ( <b>4a</b> )
<b>1b</b>	55% [( <i>E</i> )- <b>3b</b> ], 15% [( <i>Z</i> )- <b>3b</b> ]	8% ( <b>4b</b> )
<b>1c</b>	56% [( <i>E</i> )- <b>3c</b> ], 17% [( <i>Z</i> )- <b>3c</b> ]	8% ( <b>4c</b> )
<b>1d</b>	20% [( <i>E</i> )- <b>3d</b> ], 5% [( <i>Z</i> )- <b>3d</b> ]	41% ( <b>4d</b> )
<b>1e</b>	10% [( <i>E</i> )- <b>3f</b> ]	-----

\*Yields (%) based on the corresponding **1**

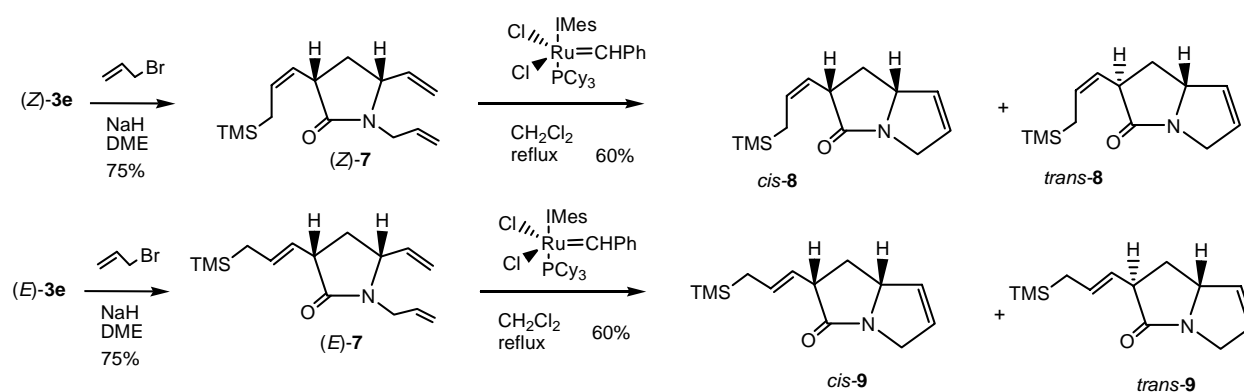
Removal of *N*-trialkylsilyl group from (*E*)-**3a**, (*Z*)-**3a**, and *E/Z* mixture of **4a** was achieved easily with KF in THF at room temperature to give (*E*)-**3e**, (*Z*)-**3e**, and *E/Z* mixture of **4e**, respectively. *cis*-Configuration at the 3- and 5-positions in **3e** and **4e** was firmly established based on NOE experiments, in which irradiation of 5-proton gave NOE enhancement at 3-proton (Scheme 2).



Scheme 2

Further conversion of **3e** to pyrrolizines (**8,9**) was performed as follows. *N*-Allylation of (*E*)-**3e** and (*Z*)-**3e** by treatment with NaH and allyl bromide gave (*E*)-**7** and (*Z*)-**7**, respectively. Ring-closing metathesis reaction of (*Z*)-**7** and (*E*)-**7** was successfully carried out in the presence of Grubbs' second generation catalyst in CH<sub>2</sub>Cl<sub>2</sub> under reflux. Under these conditions, the ring-closing reaction accompanied the

isomerization at the C3 in (*Z*)- and (*E*)-**7**, producing **8**<sup>7</sup> and **9** as a pair of *cis:trans*=1:1 isomers, respectively (Scheme 3).<sup>8</sup>



Scheme 3

In conclusion, the regioselectivity in the ruthenium-catalyzed ring-opening cross-metathesis reaction of **1** with allyltrimethylsilane was markedly improved by the introduction of a trialkylsilyl group into the nitrogen of ABH (**1**), though the mechanistic features that control the regioselectivity are not obvious. The metathesis products (**3,4**) were successfully isolated by MPLC, and moreover, transformation of **3a** allowed an access to pyrrolizines (**8,9**). Investigations including further improvement in the regioselectivity and mechanistic insight are in progress.

## ACKNOWLEDGEMENTS

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## REFERENCES AND NOTES

- (a) L. A. Agrofoglio and S. R. Challand, “Acyclic, Carbocyclic and L-Nucleosides,” Kluwer Academic Publisher, Dordrecht, 1998; (b) M. Ferrero and Y. Goto, *Chem. Rev.*, 2000, **100**, 4319; (c) E. Ichikawa and K. Kato, *Curr. Med. Chem.*, 2001, **84**, 385; L. A. Agrofoglio, I. Gillaizeau, and Y. Sato, *Chem. Rev.*, 2003, **103**, 1875.
- (a) N. Katagiri, Y. Yamatoya, and M. Ishikura, *Tetrahedron Lett.*, 1999, **40**, 9069; (b) M. Ishikura, A. Murakami, and N. Katagiri, *Heterocycles*, 2002, **58**, 317; (c) M. Ishikura, A. Murakami, and N. Katagiri, *Org. Biomol. Chem.*, 2003, **1**, 451; (d) M. Ishikura, K. Matsumoto, and A. Murakami, *Heterocycles*, 2004, **64**, 241.
- (a) M. F. Schneider, N. Lucas, J. Velder, and S. Blechert, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 257; (b) O. Arjona, A. G. Csaky, R. Medel, and J. Plumet, *J. Org. Chem.*, 2002, **67**, 1380.

4. (a) M. Ishikura, M. Saijo, and A. Hino, *Heterocycles*, 2002, **57**, 241; (b) M. Ishikura, M. Saijo, and A. Hino, *Heterocycles*, 2003, **59**, 573; (c) M. Ishikura, M. Hasunuma, and M. Saijo, *Heterocycles*, 2004, **63**, 5.
5. C. F. Palmer, K. P. Parry, S. M. Roberts, and V. Sik, *J. Chem. Soc., Perkin Trans. I*, 1992, 1021.
6. **(E)-3b**:  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.00 (s, 9H), 0.20 (s, 3H), 0.24 (s, 3H), 0.93 (s, 9H), 1.45 (d, 2H,  $J=8.0$  Hz), 1.69 (ddd, 1H,  $J=4.6, 6.3, 13.2$  Hz), 2.47 (ddd, 1H,  $J=8.0, 9.8, 13.2$  Hz), 2.96-3.03 (m, 1H), 4.07 (dt, 1H,  $J=4.6, 8.0$  Hz), 5.03 (d, 1H,  $J=10.3$  Hz), 5.08 (d, 1H,  $J=17.2$  Hz), 5.31 (dd, 1H,  $J=6.3, 14.8$  Hz), 5.54 (dtd, 1H,  $J=14.8, 8.0, 1.5$  Hz), 5.82 (ddd, 1H,  $J=8.0, 10.3, 17.2$  Hz).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : -4.2, -4.0, -1.8, 19.1, 23.0, 27.3, 36.0, 46.3, 60.8, 114.8, 126.3, 129.0, 142.8, 183.7. HR-MS  $m/z$ : Calcd for  $\text{C}_{18}\text{H}_{35}\text{NOSi}_2$ : 337.2254. Found: 337.2257. **(Z)-3b**:  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.00 (s, 9H), 0.20 (s, 3H), 0.26 (s, 3H), 0.94 (s, 9H), 1.46 (ddd, 1H,  $J=1.5, 7.4, 13.7$  Hz), 1.58 (ddd, 1H,  $J=5.1, 6.8, 13.2$  Hz), 1.67 (ddd, 1H,  $J=1.1, 9.7, 13.7$  Hz), 2.47 (ddd, 1H,  $J=8.0, 9.8, 13.2$  Hz), 3.27 (ddd, 1H,  $J=1.1, 6.8, 9.1$  Hz), 4.07-4.13 (m, 1H), 5.05 (d, 1H,  $J=10.3$  Hz), 5.10 (d, 1H,  $J=17.2$  Hz), 5.27 (dd, 1H,  $J=9.1, 10.6$  Hz), 5.57 (ddt, 1H,  $J=1.1, 7.4, 10.6$  Hz), 5.81 (ddd, 1H,  $J=7.5, 10.3, 17.2$  Hz).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : -4.1, -3.9, -1.7, 19.1, 21.0, 27.3, 36.7, 41.8, 60.7, 115.0, 125.4, 129.1, 142.6, 183.9. HR-MS  $m/z$ : Calcd for  $\text{C}_{18}\text{H}_{35}\text{NOSi}_2$ : 337.2254. Found: 337.2268. **4b**:  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.00 (s, 9H), 0.20 & 0.22 & 0.23 (three s, 6H), 0.93 & 0.94 (two s, 9H), 1.35 (ddd, 0.8H,  $J=1.5, 6.3, 13.7$  Hz), 1.43 (d, 1.2H,  $J=8.0$  Hz), 1.61-1.75 (m, 1H), 2.44 (ddd, 0.6H,  $J=8.1, 9.8, 13.2$  Hz), 2.42-2.49 (m, 0.4H), 3.01-3.09 (m, 1H), 4.05 (dt, 0.6H,  $J=5.1, 8.0$  Hz), 4.36-4.42 (m, 0.4H), 5.13-5.19 (m, 2H), 5.20-5.32 (m, 1H), 5.36-5.50 (m, 1H), 5.87-5.94 (m, 1H).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : -4.3, -4.2, -4.1, -3.9, -1.8, -1.6, 19.0, 22.5, 25.8, 27.3, 27.4, 47.3, 54.3, 60.5, 116.5, 116.6, 126.3, 127.9, 132.2, 132.8, 136.2, 136.3, 182.6. HR-MS  $m/z$ : Calcd for  $\text{C}_{18}\text{H}_{35}\text{NOSi}_2$ : 337.2254. Found: 337.2268.
7. **cis-8**: IR (neat): 1698  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.00 (s, 9H), 1.47-1.63 (m, 3H), 2.55 (ddd, 1H,  $J=13.7, 7.4, 6.3$  Hz), 3.68 (dd, 1H,  $J=3.5, 15.5$  Hz), 3.63-3.69 (m, 1H), 4.40 (dd, 1H,  $J=4.0, 15.5$  Hz), 4.53-4.59 (m, 1H), 5.23 (dd, 1H,  $J=9.7, 10.9$  Hz), 5.64 (dt, 1H,  $J=8.0, 10.9$  Hz), 5.84-5.90 (m, 2H).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : -1.8, 18.9, 38.3, 43.2, 50.0, 65.1, 123.7, 128.3, 130.2, 130.3, 178.2. HR-MS  $m/z$ : Calcd for  $\text{C}_{13}\text{H}_{21}\text{NOSi}$ : 235.1392; Found: 235.1400. **trans-8**: IR (neat): 1700  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.00 (s, 9H), 1.57 (ddd, 1H,  $J=1.1, 8.6, 13.7$  Hz), 1.70 (ddd, 1H,  $J=1.1, 1.9, 13.7$  Hz), 2.05 (dt, 1H,  $J=8.6, 12.1$  Hz), 2.14 (dd, 1H,  $J=6.3, 12.1$  Hz), 3.42 (dd, 1H,  $J=8.6, 9.2$  Hz), 3.64 (dd, 1H,  $J=3.5, 16.0$  Hz), 4.38 (dd, 1H,  $J=4.0, 16.0$  Hz), 4.72-4.78 (m, 1H), 5.45 (dd, 1H,  $J=9.2, 9.7$  Hz), 5.56 (q, 1H,  $J=9.7$  Hz), 5.83-5.89 (m, 2H).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : -1.7, 18.9, 36.8, 45.1, 49.9, 65.9, 123.3, 128.2, 129.5, 130.6, 179.2. HR-MS  $m/z$ : Calcd for  $\text{C}_{13}\text{H}_{21}\text{NOSi}$ : 235.1392; Found: 235.1390.
8. Although we have no clear account, this may involve the isomerization of less stable metallacycle

intermediate (A) to intermediate (B).

