

HETEROCYCLES, Vol. 76, No. 2, 2008, pp. 1633 - 1645. © The Japan Institute of Heterocyclic Chemistry
 Received, 13th May, 2008, Accepted, 9th June, 2008, Published online, 12th June, 2008. COM-08-S(N)121

**CATALYTIC ASYMMETRIC SYNTHESIS OF (–)-RITODRINE
 HYDROCHLORIDE VIA SILYL ENOL ETHER AMINATION USING
 DIRHODIUM(II) TETRAKIS [TETRAFLUOROPHTHALOYL-(S)-TERT-
 LEUCINATE]**

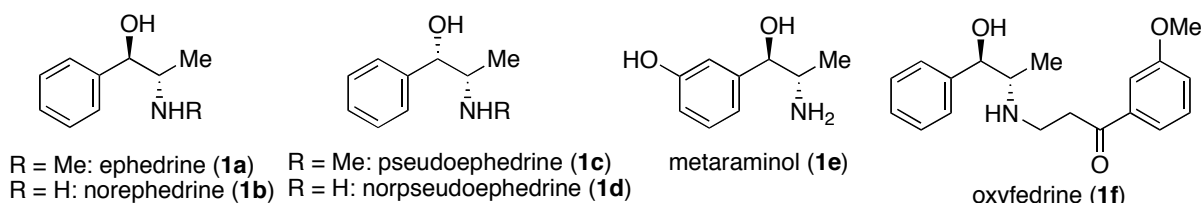
**Masahiko Tanaka, Seiichi Nakamura, Masahiro Anada,
 and Shunichi Hashimoto***

Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812,
 Japan

e-mail: hsmat@pharm.hokudai.ac.jp

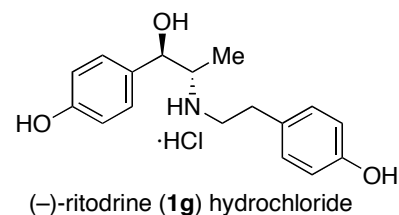
Abstract – A catalytic asymmetric synthesis of (–)-ritodrine hydrochloride was achieved, incorporating an enantioselective amination of (*Z*)-silyl enol ether derived from 4-benzyloxypropiophenone with [(2-nitrophenylsulfonyl)imino]-phenyliodinane (NsN=IPh) and a chelation-controlled reduction of the ketone carbonyl group with $Zn(BH_4)_2$ as the key steps. The use of dirhodium(II) tetrakis[tetrafluorophthaloyl-(*S*)-*tert*-leucinate] as a catalyst produced the targeted α -amino ketone in 94% yield with 91% ee.

2-Amino-1-aryl-1-propanols (**1**) are important and versatile intermediates in the synthesis of a wide variety of natural products and pharmaceuticals, such as ephedrine (**1a**),¹ norephedrine (**1b**),² pseudoephedrine (**1c**),³ norpseudoephedrine (**1d**),⁴ the α_1 -adrenergic receptor agonist metaraminol (**1e**),⁵ and the antianginal agent oxyfedrine (**1f**).⁶ Further interest stems from their use as chiral auxiliaries and ligands.⁷ Because of their importance, several strategies have been developed to achieve the catalytic asymmetric synthesis of **1**, including asymmetric nitroaldol (Henry) reaction,^{8,9} dynamic kinetic resolution of racemic α -amino ketones,¹⁰ and asymmetric reduction of α -oxoketoxime ethers.¹¹

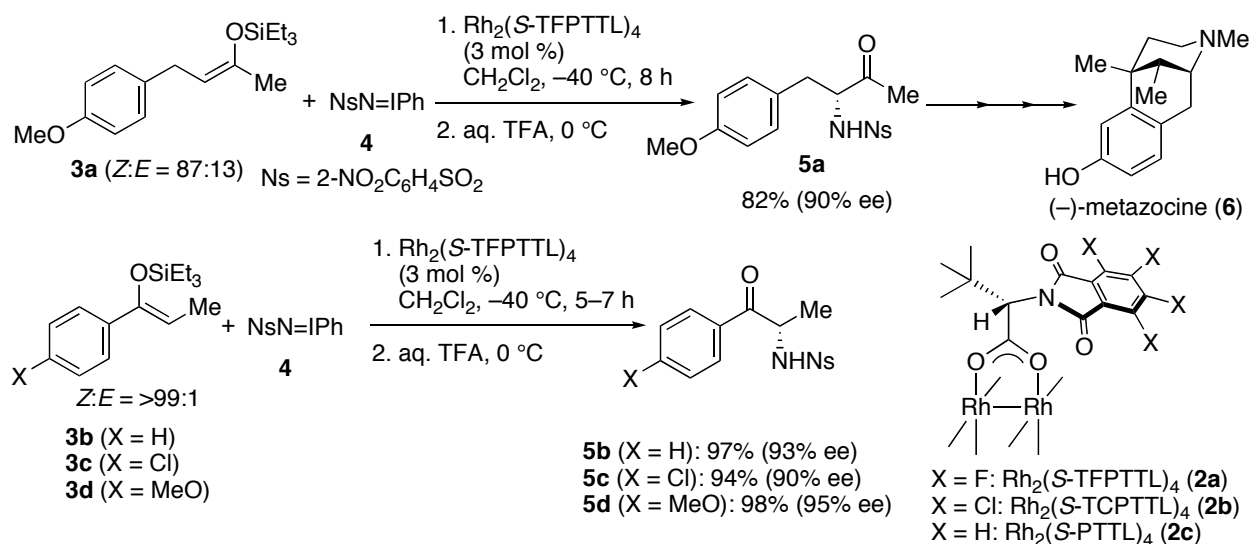


† Dedicated to Professor Ryoji Noyori on the occasion of his 70th birthday.

Ritodrine, (\pm) -(1*R**,2*S**)-1-(4-hydroxyphenyl)-2-[*N*-[2-(4-hydroxyphenyl)ethyl]amino]-1-propanol, is a selective β_2 -adrenergic receptor agonist, and its hydrochloride is currently one of the most commonly used drugs for the treatment of preterm labor.¹² Ritodrine binds to β_2 -receptors located on the outer surface of the myometrium, triggering the conversion of adenosine 5'-triphosphate (ATP) to cyclic adenosine 3',5'-monophosphate (cAMP), which leads to relaxation of myometrial tissues. Although ritodrine is used as a racemate, it has been disclosed that the (1*R*,2*S*)-(-)-enantiomer (**1g**) is primarily responsible for the pharmacological effects.¹³ Very recently, Shibasaki and co-workers described the first highly efficient, catalytic asymmetric synthesis of (-)-ritodrine (**1g**) hydrochloride by developing an *anti*- and enantio-selective nitroaldol reaction catalyzed by a heterobimetallic Pd/La/Schiff base complex.¹⁴



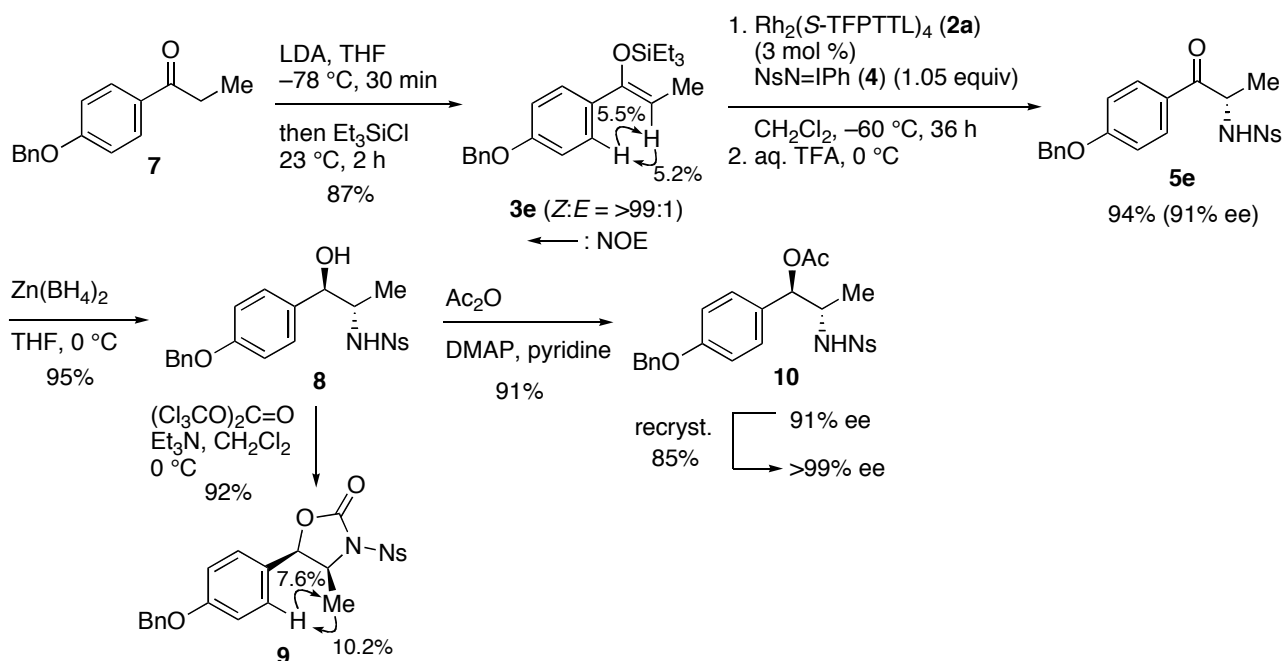
We have recently documented the utility of $\text{Rh}_2(\text{S-TFPTTL})_4$ (**2a**) and $\text{Rh}_2(\text{S-TCPTTL})_4$ (**2b**),¹⁵ characterized by the substitution of fluorine or chlorine atoms for four hydrogen atoms on the phthalimido group in the parent dirhodium(II) complex, $[\text{Rh}_2(\text{S-PTTL})_4]$ (**2c**),^{16,17} in the catalysis of enantioselective nitrene transfer reactions¹⁸ such as C–H amination^{19,20} and olefin aziridination.²¹ More recently, we have reported that $\text{Rh}_2(\text{S-TFPTTL})_4$ -catalyzed enantioselective amination of silyl enol ethers with [(2-nitrophenylsulfonyl)imino]phenyliodinane (NsN=IPh) (**4**) provides *N*-(2-nitrophenylsulfonyl)- α -amino ketones with enantioselectivities of up to 95% ee,²² the effectiveness of which has been demonstrated by an asymmetric formal synthesis of (-)-metazocine (**6**), a benzomorphan analgesic (Scheme 1). In this process, the use of NsN=IPh (**4**) as a nitrene precursor is not only crucial for high levels of enantioselection but also synthetically advantageous since the alkylation of resultant *N*-



Scheme 1

monosubstituted Ns-amides and deprotection proceed under mild conditions.²³ In order to further broaden the utility of this catalytic protocol, we envisioned that stereocontrolled reduction of the 1-phenyl-2-[N-(2-nitrophenylsulfonyl)amino]propan-1-one derivatives thus obtained would provide facile access to optically active 2-amino-1-aryl-1-propanols (**1**).^{24,25} Based on this scenario, we have now addressed the catalytic asymmetric synthesis of (–)-ritodrine (**1g**) hydrochloride.

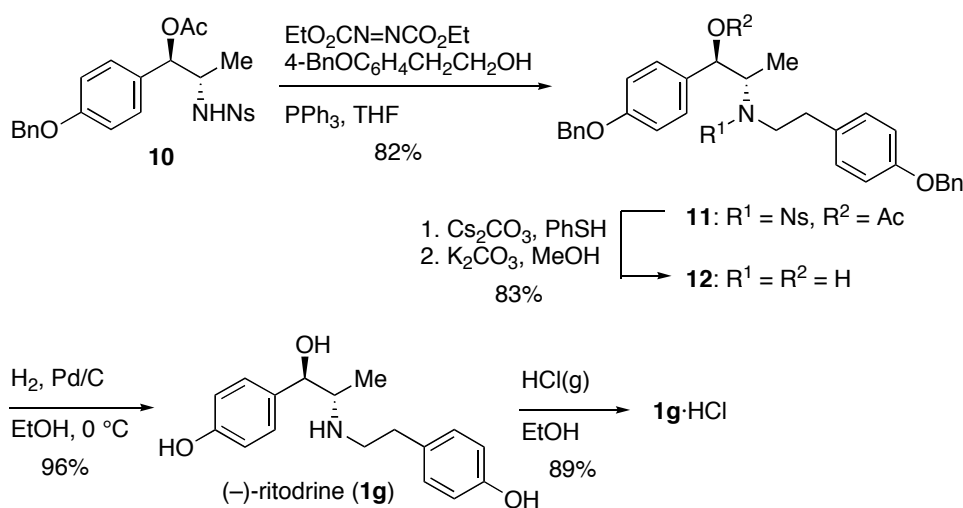
Toward this end, we selected (*Z*)-1-(4-benzyloxyphenyl)-1-triethylsiloxy-1-propene (**3e**) as a substrate on the basis of our previous finding that only (*Z*)-silyl enol ethers are responsible for the formation of α -amino ketones.^{22,26} (*Z*)-Silyl enol ether (**3e**) was readily prepared from 4-benzyloxypropiophenone (**7**)²⁷ by treatment with LDA at $-78\text{ }^{\circ}\text{C}$ followed by the addition of Et_3SiCl (Scheme 2).²⁸ The (*Z*)-stereochemistry of **3e** was established by the ^1H NOE between C2–H and *ortho* protons on the benzene ring. The reaction of **3e** with $\text{NsN}=\text{IPh}$ (**4**) in CH_2Cl_2 in the presence of 3 mol % of $\text{Rh}_2(\text{S-TFPTTL})_4$ proceeded at $-40\text{ }^{\circ}\text{C}$ to completion in 5 h, and, after treatment with 90% aqueous trifluoroacetic acid, gave α -amino ketone (**5e**), $[\alpha]_{\text{D}}^{23} +4.65^{\circ}$ (c 1.27, CHCl_3), in 92% yield. The enantioselectivity of this reaction was determined to be 82% ee by HPLC analysis (Daicel Chiralpak IA column). The preferred absolute configuration of **5e** was assigned as *S* by its conversion to **1g**·HCl (*vide infra*). Gratifyingly, the enantioselectivity was further enhanced up to 91% ee by lowering the reaction temperature to $-60\text{ }^{\circ}\text{C}$ without affecting product yield, although much longer reaction times (36 h) were necessary to reach completion. Chelation-controlled reduction of the ketone carbonyl group in **5e** with $\text{Zn}(\text{BH}_4)_2$ led to the exclusive formation of the desired *anti*-amino alcohol (**8**) in 95% yield.^{25b,c,29} The *anti*-stereochemistry of **8** was established by its transformation to the 1,3-oxazolidin-2-one (**9**),³⁰ which



Scheme 2

showed an NOE correlation between the C4-methyl group and *ortho* protons on the benzene ring. No signs of the *syn*-amino alcohol could be detected in the crude reaction mixture by NMR spectroscopy.³¹ Acetylation of **8** under standard conditions provided amino acetate (**10**) in 93% yield as a white solid. Fortunately, we found that a single recrystallization of **10** from MeOH gave crystals with 3% ee {11%, mp 130.5–131.5 °C, $[\alpha]_D^{23} -1.46^\circ$ (*c* 1.12, CHCl₃)}, while the mother liquor contained **10** with >99% ee {84%, mp 50.0–51.0 °C, $[\alpha]_D^{23} -41.2^\circ$ (*c* 1.05, CHCl₃)}.

With enantioenriched amino acetate (**10**) (>99% ee) in hand, we proceeded to the elaboration of the target molecule (Scheme 3). *N*-Alkylation of **10** with 2-(4-benzyloxyphenyl)ethanol³² under Mitsunobu conditions gave the *N,N*-disubstituted amine derivative (**11**) in 82% yield.^{33,34} Removal of the Ns group under standard Fukuyama conditions²³ followed by removal of the acetyl group with K₂CO₃ in MeOH gave β-amino alcohol (**12**) in 83% yield. Hydrogenolysis of the benzyl ethers over 10% Pd/C in EtOH furnished (–)-ritodrine (**1g**), $[\alpha]_D^{21} -5.33^\circ$ (*c* 1.43, EtOH), in 96% yield, which upon treatment with HCl in EtOH afforded (–)-ritodrine (**1g**) hydrochloride^{13,14} in 89% yield. The synthetic material (**1g**·HCl) had an optical rotation, $[\alpha]_D^{23} -13.6^\circ$ (*c* 0.30, EtOH), in good agreement with the literature value {lit.,¹³ $[\alpha]_D^{20} -13.2^\circ$ (*c* 0.24, EtOH)}. Thus, the preferred absolute configuration of α-amino ketone (**5e**) was established as *S*.



Scheme 3

In summary, we have achieved the catalytic asymmetric synthesis of (–)-ritodrine hydrochloride in 10 steps and with 35% overall yield from 4-benzyloxypropiophenone. The present protocol provides ready access to its novel analogues for biological and pharmacological investigations.

EXPERIMENTAL

General. Melting points were determined on a Büchi 535 digital melting point apparatus and are uncorrected. IR spectra were recorded on a JASCO FT/IR-4100 spectrometer and absorbance bands are

reported in wavenumber (cm^{-1}). ^1H NMR spectra were recorded on JEOL JNM-ECX 400 (400 MHz) spectrometer or JEOL JNM-ECA 500 (500 MHz) spectrometer. Chemical shifts are reported relative to the internal standard (tetramethylsilane; δ_{H} 0.00, CDCl_3 ; δ_{H} 7.26 or $\text{DMSO}-d_6$; δ_{H} 2.50). Data are presented as follows: chemical shift (δ , ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad), coupling constant and integration. ^{13}C NMR spectra were recorded on JEOL JNM-ECX 400 (100 MHz) spectrometer or JEOL JNM-ECA 500 (125 MHz) spectrometer. The following internal references were used (CDCl_3 ; δ 77.0, CD_3OD ; δ 49.8 or $\text{DMSO}-d_6$; δ 39.7). Optical rotations were measured on a JASCO P-1030 digital polarimeter at the sodium D line (589 nm). EI-MS spectra were obtained on a JEOL JMS-FABmate spectrometer, operating with ionization energy of 70 eV. FAB-MS spectra were obtained on a JEOL JMS-HX 110 spectrometer. Column chromatography was carried out on Kanto silica gel 60 N (63–210 mesh) or Wakogel[®] C-200 (75–150 mesh). Analytical thin layer chromatography (TLC) was carried out on Merck Kieselgel 60 F₂₅₄ plates with visualization by UV light, anisaldehyde stain solution or phosphomolybdic acid stain solution. Analytical high performance liquid chromatography (HPLC) was performed on a JASCO PU-1580 intelligent HPLC pump with JASCO UV-1575 intelligent UV/vis detector. Detection was performed at 254 nm. Chiralpak AD-H (0.46 cm \times 25 cm) or Chiralpak IA column (0.46 cm \times 25 cm) from Daicel were used. Retention times (t_{R}) and peak ratios were determined with JASCO-Borwin analysis system.

All non-aqueous reactions were carried out in flame-dried glassware under an argon atmosphere unless otherwise noted. Reagents and solvents were purified by standard means. Dehydrated stabilizer-free CH_2Cl_2 and THF were purchased from Kanto Chemical Co., Inc. Diisopropylamine and *N,N*-dimethylformamide (DMF) were distilled from calcium hydride prior to use. $\text{Rh}_2(\text{S-TFPTTL})_4 \cdot 2\text{EtOAc}$ (**2**) was prepared from *L-tert*-leucine according to literature procedure.^{15,17} [*N*-(2-nitrophenylsulfonyl)imino]phenyliodinane (**4**) was prepared from the corresponding sulfonamide according to literature procedure.³⁵ Diethyl azodicarboxylate (2.2 M in toluene) was purchased from Tokyo Chemical Industry Co., Ltd. Cesium carbonate was purchased from Wako Pure Chemical Industries, Ltd.

(Z)-1-(4-Benzyloxyphenyl)-1-triethylsiloxy-1-propene (3e). BuLi (1.6 M in hexane, 6.8 mL, 11.0 mmol) was added to a solution of diisopropylamine (1.31 g, 1.82 mL, 12.0 mmol) in THF (20 mL) at -78 °C. After stirring at this temperature for 0.5 h, 4-benzyloxypropiophenone²⁷ (2.40 g, 10.0 mmol) in THF (15 mL) was added to the mixture over a period of 1 h. The solution was stirred for 0.5 h and chlorotriethylsilane (1.85 mL, 11.0 mmol) was added. The mixture was allowed to warm to rt over 0.5 h and stirred for 1 h. The mixture was diluted with 5% triethylamine in hexane (60 mL), and the mixture was washed with saturated aqueous NaHCO_3 (3 \times 30 mL) and brine (2 \times 30 mL), and dried over

anhydrous Na₂SO₄. Filtration and evaporation *in vacuo* furnished the crude product, which was purified by Kugelrohr distillation (3 mmHg, 180 °C) to provide **3e** (2.81 g, 81% yield, *Z/E* = >99:1) as a colorless oil; *R_f* = 0.48 (19:1 hexane/EtOAc); IR (neat) ν : 2954, 1651, 1606, 1508, 1240, 1065, 1008, 835 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.61 (q, *J* = 8.2 Hz, 6H, CH₃CH₂Si), 0.93 (t, *J* = 8.2 Hz, 9H, CH₃CH₂Si), 1.73 (d, *J* = 6.9 Hz, 3H, CH₃CH=C), 5.06 (s, 2H, PhCH₂O), 5.10 (q, *J* = 6.9 Hz, 1H, CH₃CH=C), 6.88–6.92 (m, 2H, *ortho* proton of 4-benzyloxyphenyl group), 7.31–7.45 (m, 7H, *Ar*); ¹³C NMR (125 MHz, CDCl₃) δ 5.34 (CH₂), 6.70 (CH₃), 11.47 (CH₃), 69.95 (CH₂), 103.73 (CH), 114.18 (CH), 126.62 (CH), 127.50 (CH), 127.92 (CH), 128.53 (CH), 132.68 (C), 136.96 (C), 149.96 (C), 158.17 (C); HRMS (EI) C₂₂H₃₀O₂Si (M⁺) 354.2015, found 354.2027.

(S)-1-(4-Benzyloxyphenyl)-2-[(2-nitrophenylsulfonyl)amino]-1-propanone (5e). [*N*-(2-nitrophenylsulfonyl)imino]phenyliodinane (**4**) (806.4 mg, 2.0 mmol, 1.05 equiv) was added in one portion to a solution of **3e** (673.6 mg, 1.90 mmol, *Z/E* = >99:1) and Rh₂(*S*-TFPTTL)₄·2EtOAc (**2a**) (90.4 mg, 0.057 mmol, 3 mol %) in CH₂Cl₂ (15 mL) at –60 °C. After stirring at this temperature for 36 h, the reaction mixture was treated with 90% aqueous TFA (*ca.* 0.5 mL) followed by stirring for 0.5 h at rt. The mixture was partitioned between EtOAc (50 mL) and pH 7.0 phosphate buffer (15 mL). The separated organic layer was washed with water (2 × 20 mL) and brine (2 × 20 mL), and dried over anhydrous Na₂SO₄. Filtration and evaporation furnished the crude product, which was purified by column chromatography (silica gel, 3:1 hexane/EtOAc). Rh₂(*S*-TFPTTL)₄ (**2a**) {75.0 mg, 83%, *R_f* = 0.71 (2:1 hexane/EtOAc)} was eluted first through the column. The fractions containing the product were collected and concentrated to provide **5e** (786.1 mg, 94% yield) as a colorless viscous oil; *R_f* = 0.23 (2:1 hexane/EtOAc); [α]_D²³ +5.33° (*c* 1.51, CHCl₃) for 91% ee; IR (neat) ν : 3330, 1679, 1544, 1378, 1359, 1222, 1174, 972 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.49 (d, *J* = 7.2 Hz, 3H, CH₃CHN), 5.13 (s, 2H, PhCH₂O), 5.17 (dq, *J* = 8.2, 7.2 Hz, 1H, CH₃CHN), 6.65 (d, *J* = 8.2 Hz, 1H, NH), 6.98–7.02 (m, 2H, *Ar*), 7.33–7.43 (m, 5H, *Ar*), 7.56 (dt, *J* = 1.4, 7.7 Hz, 1H, *Ar*), 7.64 (dt, *J* = 1.4, 7.7 Hz, 1H, *Ar*), 7.79–7.83 (m, 2H, *Ar*), 7.86 (dd, *J* = 1.4, 7.7 Hz, 1H, *Ar*), 7.99 (dd, *J* = 1.4, 7.7 Hz, 1H, *Ar*); ¹³C NMR (125 MHz, CDCl₃) δ 21.16 (CH₃), 54.12 (CH), 70.19 (CH₂), 114.95 (CH), 125.49 (CH), 126.28 (C), 127.43 (CH), 128.32 (CH), 128.68 (CH), 130.01 (CH), 130.91 (CH), 132.71 (CH), 133.49 (CH), 134.33 (C), 135.75 (C), 147.56 (C), 163.41 (C), 195.63 (C=O); HRMS (FAB) calcd for C₂₂H₂₁N₂O₆S (M+H⁺) 441.1120, found 441.1126. The enantiomeric excess of **5e** was determined to be 91% by HPLC with a Daicel Chiralpak IA column (3:1 hexane/*i*-PrOH, 1.0 mL/min): *t_R* (major) = 30.5 min for (*S*)-enantiomer; *t_R* (minor) = 33.9 min for (*R*)-enantiomer.

(1R,2S)-1-(4-Benzyloxyphenyl)-2-[(2-nitrophenylsulfonyl)amino]-1-propanol (8). Zn(BH₄)₂³⁶ (0.64 M in THF, 3.2 mL, 2.06 mmol) was added to a solution of **5e** (603.9 mg, 1.37 mmol) in THF (15 mL) at

0 °C. After stirring at 0 °C for 0.5 h, the reaction was quenched with 10% aqueous HCl (5 mL) and the whole was extracted with EtOAc (40 mL). The organic layer was washed with water (2 × 10 mL) and brine (2 × 10 mL), and dried over anhydrous Na₂SO₄. Filtration and concentration *in vacuo* furnished the crude product, which was purified by column chromatography (silica gel, 2:1 hexane/EtOAc) to give **8** (576 mg, 95%) as colorless needles; *R_f* = 0.53 (1:1 hexane/EtOAc); mp 54.0–56.0 °C (2:1 hexane/EtOAc); [α]_D²³ –57.6° (*c* 1.14, MeOH) for 91% ee; IR (KBr) *ν*: 3528, 3321, 1540, 1356, 1332, 1227, 1054 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.01 (d, *J* = 6.9 Hz, 3H, CH₃CHN), 2.16 (br-s, 1H, OH), 3.78 (ddq, *J* = 3.9, 8.3, 6.9 Hz, 1H, CH₃CHN), 4.73 (d, *J* = 3.9 Hz, 1H, CHOH), 5.04 (s, 2H, PhCH₂O), 5.51 (d, *J* = 8.3 Hz, 1H, NH), 6.88–6.91 (m, 2H, *Ar*), 7.16–7.19 (m, 2H, *Ar*), 7.34 (m, 1H, *Ar*), 7.38–7.44 (m, 4H, *Ar*), 7.70 (dt, *J* = 1.7, 7.4 Hz, 1H, *Ar*), 7.73 (dt, *J* = 1.7, 7.4 Hz, 1H, *Ar*), 7.84 (m, 1H, *Ar*), 8.14 (m, 1H, *Ar*); ¹³C NMR (125 MHz, CDCl₃) δ 15.61 (CH₃), 55.98 (CH), 69.97 (CH₂), 75.65 (CH), 114.71 (CH), 125.41 (CH), 127.32 (CH), 127.45 (CH), 128.04 (CH), 128.60 (CH), 130.63 (CH), 132.15 (C), 132.89 (CH), 133.36 (CH), 134.80 (C), 136.75 (C), 147.61 (C), 158.47 (C); HRMS (FAB) C₂₂H₂₂N₂O₆Na (M+Na)⁺ 465.1096, found 465.1091.

(4S,5R)-5-(4-Benzyloxyphenyl)-4-methyl-3-(2-nitrophenylsulfonyl)-1,3-oxazolidin-2-one (9).

Triphosgene (42.1 mg, 0.14 mmol) was added at 0 °C to a solution of **8** (62.8 mg, 0.14 mmol, 91% ee) and Et₃N (71.8 mg, 0.71 mmol) in CH₂Cl₂ (3 mL). After stirring at 0 °C for 0.5 h, the reaction was quenched with crushed ice. The mixture was partitioned between EtOAc (20 mL) and water (3 mL). The organic layer was washed with brine (2 × 3 mL) and dried over anhydrous Na₂SO₄. Filtration and evaporation *in vacuo* furnished the crude product, which was purified by column chromatography (silica gel, 2:1 hexane/EtOAc) to provide **9** (60.5 mg, 92%) as colorless needles; *R_f* = 0.34 (2:1 hexane/EtOAc); mp 107.0–108.5 °C; [α]_D²³ +235.1° (*c* 1.14, CHCl₃); IR (KBr) *ν*: 2924, 1779, 1543, 1516, 1373, 1348, 1253, 1178, 969 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.16 (d, *J* = 6.9 Hz, 3H, C4-CH₃), 4.82 (dq, *J* = 6.9, 7.4 Hz, 1H, C4-H), 5.08 (s, 2H, PhCH₂O), 5.82 (d, *J* = 7.4 Hz, 1H, C5-H), 7.03 (d, *J* = 8.6 Hz, 2H, *Ar*), 7.25 (d, *J* = 8.6 Hz, 2H, *Ar*), 7.33–7.44 (m, 5H, *Ar*), 7.80–7.86 (m, 3H, *Ar*), 8.48 (m, 1H, *Ar*); ¹³C NMR (125 MHz, CDCl₃) δ 17.36 (CH₃), 58.77 (CH), 70.08 (CH₂), 80.39 (CH), 115.10 (CH), 124.59 (CH), 124.66 (C), 127.28 (CH), 127.49 (CH), 128.12 (CH), 128.63 (CH), 130.67 (CH), 132.25 (CH), 134.80 (C), 135.40 (C), 136.49 (C), 147.94 (C), 151.18 (C), 159.33 (C=O); HRMS (EI) C₂₃H₂₀N₂O₇S (M⁺) 468.0991, found 468.1002.

(1R,2S)-1-(4-Benzyloxyphenyl)-2-[(2-nitrophenylsulfonyl)amino]-1-propyl acetate (10). Acetic anhydride (195 mg, 1.91 mmol) was added to a solution of **8** (564.5 mg, 1.28 mmol) and 4-(*N,N*-dimethylamino)pyridine (15.9 mg, 0.13 mmol) in pyridine (2 mL) at 0 °C. After stirring at rt for

0.5 h, the reaction was quenched with crushed ice. The mixture was partitioned between EtOAc (25 mL) and H₂O (5 mL). The organic layer was washed with 1 N HCl (2 × 5 mL) and brine (2 × 5 mL), and dried over anhydrous Na₂SO₄. Filtration and concentration *in vacuo* furnished the crude product, which was purified by column chromatography (silica gel, 3:1 hexane/EtOAc) to give **10** (576.7 mg, 93%) as a white solid; $R_f = 0.38$ (3:2 hexane/EtOAc); mp 51.0–55.0 °C; $[\alpha]_D^{23} -35.9^\circ$ (c 1.06, CHCl₃) for 91% ee; IR (KBr) ν : 3290, 2979, 1719, 1540, 1365, 1336, 1251, 1163, 1032 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.10 (d, $J = 6.9$ Hz, 3H, CH₃CHN), 2.02 (s, 3H, CH₃CO), 3.97 (ddq, $J = 4.6, 8.6, 6.9$ Hz, 1H, CH₃CHN), 5.01 (s, 2H, PhCH₂O), 5.40 (d, $J = 8.6$ Hz, 1H, NH), 5.54 (d, $J = 4.6$ Hz, 1H, CHOAc), 6.87 (d, $J = 8.6$ Hz, 2H, *Ar*), 7.13 (d, $J = 8.6$ Hz, 2H, *Ar*), 7.32–7.42 (m, 5H, *Ar*), 7.70 (m, 2H, *Ar*), 7.82 (dd, $J = 1.1, 7.4$ Hz, 1H, *Ar*), 8.10 (dd, $J = 1.1, 7.4$ Hz, 1H, *Ar*); ¹³C NMR (125 MHz, CDCl₃) δ 17.22 (CH₃), 20.87 (CH₃), 53.86 (CH), 69.89 (CH₂), 76.43 (CH), 114.73 (CH), 125.43 (CH), 127.40 (CH), 127.89 (CH), 127.95 (C), 128.02 (CH), 128.57 (CH), 130.37 (CH), 132.99 (CH), 133.40 (CH), 134.76 (C), 136.61 (C), 147.48 (C), 158.72 (C), 169.82 (C=O); HRMS (EI) C₂₄H₂₄N₂O₇S (M⁺) 484.1304, found 484.1299; Anal. Calcd for C₂₄H₂₄N₂O₇S: C, 59.49; H, 4.99; N, 5.78; S, 6.62. Found: C, 59.31; H, 4.95; N, 5.72; S, 6.60.

10 (419.0 mg, 91% ee) was dissolved in 4.5 mL of boiling MeOH and stored at rt overnight. Colorless needles of **10** (47.0 mg, 11%, 3% ee) {mp 130.5–131.5 °C, $[\alpha]_D^{23} -1.46^\circ$ (c 1.12, CHCl₃)} were removed by filtration, and the filtrate was concentrated *in vacuo* to afford **10** (350 mg, 84%, >99% ee) as colorless needles; mp 50.0–51.0 °C; $[\alpha]_D^{23} -41.2^\circ$ (c 1.05, CHCl₃).

The homochirality of **10** was established by comparison of retention time in HPLC (Chiralpak, AD-H, 1:1 hexane/*i*-PrOH, 1.0 mL/min) with a racemic sample: t_R (minor) = 8.45 min for *ent*-**10**, t_R (major) = 10.53 min for **10**.

(1R,2S)-1-(4-Benzyloxyphenyl)-2-[N-[2-(4-benzyloxyphenyl)ethyl]-N-(2-nitrophenylsulfonyl)amino]-1-propyl acetate (11). Diethyl azodicarboxylate (2.2 M in toluene, 0.75 mL, 1.65 mmol) was added to a solution of **10** (400 mg, 0.826 mmol, >99% ee), 2-(4-benzyloxyphenyl)ethan-1-ol³² (377.1 mg, 1.65 mmol), and triphenylphosphine (433.3 mg, 1.65 mmol) in THF (4 mL) at rt, and the mixture was stirred for 3 h. The mixture was concentrated *in vacuo* and the residue was purified by column chromatography (silica gel, 9:1 toluene/Et₂O) to afford **11** (468.4 mg, 82%) as a colorless viscous oil; $R_f = 0.36$ (2:1 hexane/EtOAc); $[\alpha]_D^{23} -30.4^\circ$ (c 1.07, CHCl₃); IR (CHCl₃) ν : 3031, 2942, 1744, 1544, 1372, 1349, 1233, 1175, 1024 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 1.35 (d, $J = 6.9$ Hz, 3H, CH₃CHN), 2.07 (s, 3H, CH₃CO), 2.79 (dt, $J = 5.4, 11.5$ Hz, 1H, ArCH₂CH₂N), 2.91 (dt, $J = 5.4, 11.8$ Hz, 1H, ArCH₂CH₂N), 3.44 (ddd, $J = 5.4, 11.5, 15.1$ Hz, 1H, ArCH₂CH₂N), 3.50 (ddd, $J = 5.4, 11.8, 15.1$ Hz, 1H, ArCH₂CH₂N), 4.33 (dq, $J = 5.7, 6.9$ Hz, 1 H, CH₃CHN), 5.01 (s, 2H, PhCH₂O), 5.05 (s, 2H, PhCH₂O), 5.83 (d, $J = 5.7$ Hz, 1H, ArCHOAc), 6.80–6.83 (m, 2H, *Ar*), 6.91–6.94 (m, 2H, *Ar*), 7.08–7.11 (m, 2H, *Ar*), 7.13–7.16 (m, 2H, *Ar*),

7.32–7.36 (m, 2H, *Ar*), 7.38–7.44 (m, 8H, *Ar*), 7.54 (dd, $J = 1.1, 7.4$ Hz, 1H, *Ar*), 7.56 (dd, $J = 1.1, 7.4$ Hz, 1H, *Ar*), 7.62 (dt, $J = 1.1, 7.4$ Hz, 1H, *Ar*), 7.85 (dd, $J = 1.1, 7.4$ Hz, 1H, *Ar*); ^{13}C NMR (125 MHz, CDCl_3) δ 14.53 (CH_3), 21.20 (CH_3), 36.82 (CH_2), 46.78 (CH_2), 57.68 (CH), 69.88 (CH_2), 69.96 (CH_2), 77.64 (CH), 114.59 (CH), 115.00 (CH), 124.23 (CH), 127.41 (CH), 127.43 (CH), 127.81 (CH), 127.92 (CH), 128.03 (CH), 128.54 (C), 128.58 (CH), 128.59 (CH), 129.95 (C), 130.66 (C), 130.86 (CH), 131.58 (CH), 133.46 (CH), 133.58 (C), 136.73 (C), 136.94 (C), 147.95 (C), 157.53 (C), 169.56 (C=O); HRMS (FAB) $\text{C}_{24}\text{H}_{24}\text{N}_2\text{O}_7\text{S}$ ($\text{M}+\text{Na}$) $^+$ 717.2247, found 717.2245.

(1R,2S)-1-(4-Benzyloxyphenyl)-2-[2-(4-benzyloxyphenyl)ethyl]amino-1-propanol (12). Cs_2CO_3 (628.0 mg, 1.93 mmol) was added to a solution of sulfonamide **11** (447.1 mg, 0.643 mmol) and PhSH (132 μL , 1.29 mmol) in DMF (0.3 mL) at rt. After stirring for 1 h, the mixture was poured into 10% aqueous NaOH (3 mL) and the whole was extracted with EtOAc (2×15 mL). The combined organic layers were washed with H_2O (2×4 mL) and brine (2×4 mL) and dried over anhydrous Na_2SO_4 . Filtration and evaporation *in vacuo* furnished the crude product, which was dissolved in MeOH (0.5 mL). K_2CO_3 (444.3 mg, 3.22 mmol) was added to the mixture at rt. After stirring for 2 h, the mixture was partitioned between EtOAc (5 mL) and H_2O (5 mL). The whole was extracted with EtOAc (2×15 mL) and the organic layer was washed with brine (2×7 mL) and dried over anhydrous Na_2SO_4 . Filtration and evaporation *in vacuo* furnished the crude product, which was purified by column chromatography (silica gel, 19:1 $\text{CHCl}_3/\text{MeOH}$) to afford **12** (249.6 mg, 83% for 2 steps) as a colorless solid; $R_f = 0.32$ (9:1 $\text{CHCl}_3/\text{MeOH}$); mp 52.0–53.0 $^\circ\text{C}$; $[\alpha]_{\text{D}}^{23} -8.1^\circ$ (c 1.16, MeOH); IR (KBr) ν : 3401, 3304, 3104, 1611, 1511, 1455, 1382, 1241, 1170, 1014 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 0.82 (d, $J = 6.3$ Hz, 3H, CH_3CHN), 2.34 (br-s, 1H, OH), 2.68–2.81 (m, 2H), 2.85–2.99 (m, 3H), 4.68 (d, $J = 3.6$ Hz, 1 H, ArCHOH), 5.04 (s, 2H, PhCH_2O), 5.05 (s, 2H, PhCH_2O), 6.91 (d, $J = 8.6$ Hz, 2H, *Ar*), 6.93 (d, $J = 8.6$ Hz, 2H, *Ar*), 7.11 (d, $J = 8.6$ Hz, 2H, *Ar*), 7.21 (d, $J = 8.6$ Hz, 2H, *Ar*), 7.30–7.44 (m, 10H, *Ar*); ^{13}C NMR (100 MHz, CD_3OD) δ 16.37 (CH_3), 35.98 (CH_2), 49.74 (CH_2), 50.65 (CH_2), 60.59 (CH), 71.73 (CH_2), 77.85 (CH), 116.61 (CH), 116.79 (CH), 129.36 (CH), 129.40 (CH), 129.58 (C), 129.60 (C), 130.25 (CH), 130.27 (CH), 131.31 (CH), 131.34 (CH), 133.29 (CH), 136.34 (C), 139.38 (C), 139.49 (C), 159.50 (C), 160.53 (C); HRMS (FAB) $\text{C}_{31}\text{H}_{34}\text{NO}_3$ ($\text{M}+\text{H}$) $^+$ 468.2539, found 468.2530.

(1R,2S)-1-(4-Hydroxyphenyl)-2-[N-[2-(4-hydroxyphenyl)ethyl]amino]-1-propanol [(–)-ritodrine] (1g). A solution of benzyl ether **12** (111.0 mg, 0.238 mmol) in EtOH (2 mL) was vigorously stirred with 10% Pd/C (20 mg) under 1 atm of hydrogen at rt for 3 h. The catalyst was removed by filtration through a Celite[®] pad, and the filtrate was evaporated *in vacuo*. The residue was purified by column chromatography (silica gel, 3:1 $\text{CHCl}_3/\text{MeOH}$) to afford **1g** (65.7 mg, 96%) as a colorless viscous oil; R_f

= 0.28 (3:1 CHCl₃/MeOH); $[\alpha]_D^{21} -5.33^\circ$ (*c* 1.43, EtOH); IR (neat) ν : 3282, 3022, 2927, 1614, 1596, 1516, 1451, 1238 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆) δ 0.85 (d, *J* = 6.3 Hz, 3H, CH₃CHN), 2.64-2.67 (m, 2H), 2.79-2.90 (m, 3H), 4.65 (d, *J* = 2.9 Hz, 1 H, CHOHar), 6.68 (d, *J* = 8.6 Hz, 2H, *Ar*), 6.71 (d, *J* = 8.6 Hz, 2H, *Ar*), 6.98 (d, *J* = 8.6 Hz, 2H, *Ar*), 7.10 (d, *J* = 8.6 Hz, 2H, *Ar*); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 12.91 (CH₃), 33.82 (CH₂), 47.78 (CH₂), 58.39 (CH), 72.27 (CH), 114.67 (CH), 115.16 (CH), 127.20 (CH), 129.32 (C), 129.44 (CH), 132.98 (C), 155.71 (C), 156.21 (C); HRMS (FAB) C₃₁H₃₄NO₃ (M+H)⁺ 288.1600, found 288.1610.

(-)-**Ritdorine·hydrochloride (1g·HCl)**.^{13,14} Saturated HCl in EtOH (0.5 mL) was added to a solution of **1g** (50.7 mg) in EtOH (0.5 mL). The solvent was evaporated under reduced pressure to afford (-)-**1g·HCl** (48.8 mg, 89%) as a colorless gum; $[\alpha]_D^{23} -13.6^\circ$ (*c* 0.30, EtOH) {lit.,¹³ $[\alpha]_D^{20} -13.2^\circ$ (*c* 0.24, EtOH)}; IR (neat) ν : 3250, 3019, 2977, 1614, 1596, 1516, 1446, 1223 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) δ 0.94 (d, *J* = 6.9 Hz, 3H, CH₃CHN), 2.82-2.96 (m, 2H), 3.13 (brs, 2H), 3.34 (1H, m), 5.03 (s, 1 H, ArOH), 5.96 (d, *J* = 4.0 Hz, 1H, CHOHar), 6.73 (d, *J* = 8.6 Hz, 2H, *Ar*), 6.75 (d, *J* = 8.6 Hz, 2H, *Ar*), 7.06 (d, *J* = 8.6 Hz, 2H, *Ar*), 7.16 (d, *J* = 8.6 Hz, 2H, *Ar*); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 9.37 (CH₃), 30.85 (CH₂), 46.13 (CH₂), 58.43 (CH), 69.29 (CH), 114.89 (CH), 115.39 (CH), 126.91 (CH), 127.57 (C), 129.60 (CH), 131.19 (C), 156.21 (C), 156.58 (C).

ACKNOWLEDGEMENTS

This research was supported in part by a Grant-in-Aid for Scientific Research on Priority Areas "Advanced Molecular Transformations of Carbon Resources" from the Ministry of Education, Culture, Sports, Science and Technology, Japan and by Grant-in-Aid from Innovation Plaza Hokkaido in Japan Science and Technology Agency. We thank S. Oka, M. Kiuchi and T. Hirose of the Center for Instrumental Analysis at Hokkaido University, for technical assistance in MS and elemental analysis.

REFERENCES AND NOTES

1. M. M. Weinberger, *Pediatr. Clin. North Am.*, 1975, **22**, 121.
2. (a) O. K. Haugeto, K. E. Schroder, and I. W. Mair, *J. Otolaryngol.*, 1981, **10**, 359. (b) I. Kanfer, J. M. Haigh, and R. Dowse, 'Analytical Profiles of Drug Substances,' Vol. 12, ed. by K. Florey, Academic Press, New York, 1983, pp. 357-383.
3. S. A. Benezra and J. W. McRae, 'Analytical Profiles of Drug Substances,' Vol. 8, ed. by K. Florey, Academic Press, New York, 1979, pp. 489-507.
4. (a) H. Hofmann, K. Opitz, and H.-J. Schnelle, *Arzneimittel-Forsch.*, 1955, **5**, 367. (b) F. Frosch, *Arzneimittel-Forsch.*, 1977, **27**, 665.

5. A. Cession-Fossion, *Archiv. Int. Pharmacodyn. Ther.*, 1968, **172**, 421.
6. K. Thiele, U. Schimassek, and A. von Schlichtegroll, *Arzneimittel-Forsch.*, 1966, **16**, 1064.
7. For reviews, see: (a) D. J. Ager, I. Prakash, and D. R. Schaad, *Chem. Rev.*, 1996, **96**, 835. (b) A. G. Myers and M. G. Charrest, 'Handbook of Reagents for Organic Synthesis: Chiral Reagents for Asymmetric Synthesis,' ed. by L. A. Paquette, Wiley, Chichester, 2003, pp. 485–496.
8. For reviews, see: (a) M. Shibasaki and H. Gröger, 'Comprehensive Asymmetric Catalysis,' vol. III, eds. by E. N. Jacobsen, A. Pfaltz, and H. Yamamoto, Springer, Berlin, 1999, pp. 1075–1090. (b) M. Shibasaki, H. Gröger, and M. Kanai, 'Comprehensive Asymmetric Catalysis, Supplement 1,' eds. by E. N. Jacobsen, A. Pfaltz, and H. Yamamoto, Springer, Heidelberg, 2004, pp. 131–133. (c) C. Palomo, M. Oiarbide, and A. Mielgo, *Angew. Chem. Int. Ed.*, 2004, **43**, 5442. (d) J. Boruwa, N. Gogoi, P. P. Saikia, and N. C. Barua, *Tetrahedron: Asymmetry*, 2006, **17**, 3315. (e) C. Palomo, M. Oiarbide, and A. Laso, *Eur. J. Org. Chem.*, 2007, 2561.
9. For selected examples, see: (a) H. Sasai, T. Suzuki, S. Arai, T. Arai, and M. Shibasaki, *J. Am. Chem. Soc.*, 1992, **114**, 4418. (b) H. Sasai, T. Tokunaga, S. Watanabe, T. Suzuki, N. Itoh, and M. Shibasaki, *J. Org. Chem.*, 1995, **60**, 7388. (c) T. Arai, Y. M. A. Yamada, N. Yamamoto, H. Sasai, and M. Shibasaki, *Chem. Eur. J.*, 1996, **2**, 1368. (d) T. Ooi, K. Doda, and K. Maruoka, *J. Am. Chem. Soc.*, 2003, **125**, 2054. (e) T. Risgaard, K. V. Gothelf, and K. A. Jørgensen, *Org. Biomol. Chem.*, 2003, **1**, 153. (f) D. Uraguchi, S. Sasaki, and T. Ooi, *J. Am. Chem. Soc.*, 2007, **129**, 12392. (g) T. Nitabaru, N. Kumagai, and M. Shibasaki, *Tetrahedron Lett.*, 2008, **49**, 272.
10. (a) N. Arai, H. Ooka, K. Azuma, T. Yabuuchi, N. Kurono, T. Inoue, and T. Ohkuma, *Org. Lett.*, 2007, **9**, 939, see also: (b) T. Nishi, M. Kitamura, T. Ohkuma, and R. Noyori, *Tetrahedron Lett.*, 1988, **29**, 6327. (c) K. Matsumura, S. Hashiguchi, T. Ikariya, and R. Noyori, *J. Am. Chem. Soc.*, 1997, **119**, 8738.
11. M. Masui and T. Shioiri, *Tetrahedron Lett.*, 1998, **39**, 5195.
12. (a) V. Claassen, J. Van Dijk, and H. D. Moed, US Patent 3,410,944, 1968. (b) J. Van Dijk and H. D. Moed, *Recl. Trav. Chim. Pays-Bas*, 1973, **92**, 1281. (c) B. E. Travis and J. M. McCullough, *Pharmacotherapy*, 1993, **13**, 28.
13. N. Yamazaki, Y. Fukuda, Y. Shibasaki, T. Niizato, I. Kosugi, and S. Yoshioka, US Patent 5,449,694, 1995.
14. S. Handa, K. Nagawa, Y. Sohtome, S. Matsunaga, and M. Shibasaki, *Angew. Chem. Int. Ed.*, 2008, **47**, 3230.
15. H. Tsutsui, Y. Yamaguchi, S. Kitagaki, S. Nakamura, M. Anada, and S. Hashimoto, *Tetrahedron: Asymmetry*, 2003, **14**, 817.
16. (a) N. Watanabe, T. Ogawa, Y. Ohtake, S. Ikegami, and S. Hashimoto, *Synlett*, 1996, 85. (b) M.

- Anada and S. Hashimoto, *Tetrahedron Lett.*, 1998, **39**, 79. (c) M. Anada and S. Hashimoto, *Tetrahedron Lett.*, 1998, **39**, 9063. (d) S. Kitagaki, M. Yasugahira, M. Anada, M. Nakajima, and S. Hashimoto, *Tetrahedron Lett.*, 2000, **41**, 5931. (e) T. Takahashi, H. Tsutsui, M. Tamura, S. Kitagaki, M. Nakajima, and S. Hashimoto, *Chem. Commun.*, 2001, 1604. (f) S. Kitagaki, Y. Yanamoto, H. Tsutsui, M. Anada, M. Nakajima, and S. Hashimoto, *Tetrahedron Lett.*, 2001, **42**, 6361. (g) H. Tsutsui, M. Matsuura, K. Makino, S. Nakamura, M. Nakajima, S. Kitagaki, and S. Hashimoto, *Isr. J. Chem.*, 2001, **41**, 283. (h) H. Saito, H. Oishi, S. Kitagaki, S. Nakamura, M. Anada, and S. Hashimoto, *Org. Lett.*, 2002, **4**, 3887. (i) K. Minami, H. Saito, H. Tsutsui, H. Nambu, M. Anada, and S. Hashimoto, *Adv. Synth. Catal.*, 2005, **347**, 1483. (j) Y. Natori, M. Anada, S. Nakamura, H. Nambu, and S. Hashimoto, *Heterocycles*, 2006, **70**, 635.
17. For a practical synthesis of $\text{Rh}_2(\text{S-PTTL})_4$, see: H. Tsutsui, T. Abe, S. Nakamura, M. Anada, and S. Hashimoto, *Chem. Pharm. Bull.*, 2005, **53**, 1366.
18. For recent reviews on nitrene transfer reactions catalyzed by transition metal complexes, see: (a) P. Müller, 'Advances in Catalytic Processes,' vol. II, ed. by M. P. Doyle, JAI Press, Greenwich, 1997, pp. 113–151. (b) E. N. Jacobsen, 'Comprehensive Asymmetric Catalysis,' vol. II, eds. by E. N. Jacobsen, A. Pfaltz, and H. Yamamoto, Springer, Berlin, 1999, pp. 607–618. (c) T. Katsuki, 'Catalytic Asymmetric Synthesis,' ed. by I. Ojima, Wiley-VCH, New York, 2000, pp. 287–325. (d) P. Dauban and R. H. Dodd, *Synlett*, 2003, 1571. (e) P. Müller and C. Fruit, *Chem. Rev.*, 2003, **103**, 2905. (f) M. P. Doyle, 'Reactive Intermediate Chemistry,' eds. by R. A. Moss, M. S. Platz, and M. Jones, Jr., John Wiley & Sons, Hoboken, 2004, pp. 561–592. (g) T. Katsuki, *Chem. Lett.*, 2005, **34**, 1304. (h) C. G. Espino and J. Du Bois, 'Modern Rhodium-Catalyzed Organic Reactions,' ed. by P. A. Evans, Wiley-VCH, Weinheim, 2005, pp. 379–416. (i) H. M. L. Davies and M. S. Long, *Angew. Chem. Int. Ed.*, 2005, **44**, 3518. (f) H. M. L. Davies, *Angew. Chem. Int. Ed.*, 2006, **45**, 6422.
19. (a) M. Yamawaki, H. Tsutsui, S. Kitagaki, M. Anada, and S. Hashimoto, *Tetrahedron Lett.*, 2002, **43**, 9561. (b) M. Yamawaki, S. Kitagaki, M. Anada, and S. Hashimoto, *Heterocycles*, 2006, **69**, 527.
20. Recently, Davies and Reddy reported highly enantioselective benzylic C–H aminations using dirhodium(II) tetrakis[*N*-tetrachlorophthaloyl-(*S*)-(1-adamantyl)glycinate], $[\text{Rh}_2(\text{S-TCPTAD})_4]$, as a catalyst, see: R. P. Reddy and H. M. L. Davies, *Org. Lett.*, 2006, **8**, 5013.
21. M. Yamawaki, M. Tanaka, T. Abe, M. Anada, and S. Hashimoto, *Heterocycles*, 2007, **72**, 709.
22. M. Anada, M. Tanaka, T. Washio, M. Yamawaki, T. Abe, and S. Hashimoto, *Org. Lett.*, 2007, **9**, 4559.
23. (a) T. Fukuyama, C.-K. Jow, and M. Cheung, *Tetrahedron Lett.*, 1995, **36**, 6373. For a review on the nitrophenylsulfonamide chemistry, see: (b) T. Kan and T. Fukuyama, *Chem. Commun.*, 2004, 353.
24. (a) D. A. Evans and D. S. Johnson, *Org. Lett.*, 1999, **1**, 595. (b) Y. Yamashita, H. Ishitani, and S.

- Kobayashi, *Can. J. Chem.*, 2000, **78**, 666, see also: (c) N. Kumaragurubaran, K. Juhl, W. Zhuang, A. Bøgevig, and K. A. Jørgensen, *J. Am. Chem. Soc.*, 2002, **124**, 6254.
25. For examples of stereoselective reduction of α -amino ketones, see: (a) M. Fujita and T. Hiyama, *J. Org. Chem.*, 1988, **53**, 5415. (b) S. Sengupta, D. Das, and S. Mondal, *Synlett*, 2001, 1464. (c) R. V. Hoffman, N. Maslouh, and F. Cervantes-Lee, *J. Org. Chem.*, 2002, **67**, 1045. (d) D. S. Fraser, S. B. Park, and J. M. Chong, *Can. J. Chem.*, 2004, **82**, 87.
26. Very recently, we reported the enantioselective amination of silylketene acetals derived from methyl arylacetates with NsN=IPh catalyzed by $\text{Rh}_2(\text{S-TCPTTL})_4$, in which (*E*)-silylketene acetals were also found to be less reactive, see: M. Tanaka, Y. Kurosaki, T. Washio, M. Anada, and S. Hashimoto, *Tetrahedron Lett.*, 2007, **48**, 8799.
27. C. L. Viswanathan, M. M. Kodgule, and A. S. Chaudhari, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 3532.
28. R. Noyori, I. Nishida, and J. Sakata, *J. Am. Chem. Soc.*, 1983, **105**, 1598.
29. T. Oishi and T. Nakata, *Acc. Chem. Res.*, 1984, **17**, 338.
30. M. Tang and S. G. Pyne, *J. Org. Chem.*, 2003, **68**, 7818.
31. Reduction of **5e** with NaBH_4 in EtOH at 0 °C afforded a 1.5:1 mixture of *syn*- and *anti*-amino alcohols in 92% yield.
32. N. Lebegue, S. Gallet, N. Flouquet, P. Carato, B. Pfeiffer, P. Renard, S. Léonce, A. Pierré, P. Chavatte, P. Berthelot, *J. Med. Chem.*, 2005, **48**, 7363.
33. For examples of *N*-alkylation under Mitsunobu conditions, see: (a) J. R. Henry, L. R. Marcin, M. C. McIntosh, P. M. Scola, G. D. Harris, Jr., and S. M. Weinreb, *Tetrahedron Lett.*, 1989, **30**, 5709. (b) T. Tsunoda, J. Otsuka, Y. Yamamiya, and S. Itô, *Chem. Lett.*, 1994, **23**, 539.
34. Attempts at *N*-alkylation of amino alcohol (**8**) with 2-(4-benzyloxyphenyl)ethyl iodide (3 equiv.) and with Cs_2CO_3 (3 equiv.) in DMF at 23 °C met with little success (*ca.* 30% yield) due to the formation of substantial amounts (*ca.* 60% yield) of 4-benzyloxystyrene arising from the β -elimination.
35. (a) Y. Yamada, T. Yamamoto, and M. Okawara, *Chem. Lett.*, 1975, **4**, 361. (b) M. J. Södergren, D. A. Alonso, and P. G. Andersson, *Tetrahedron: Asymmetry*, 1997, **8**, 3563.
36. W. J. Gensler, F. Johnson, and A. D. B. Sloan, *J. Am. Chem. Soc.*, 1960, **82**, 6074.