

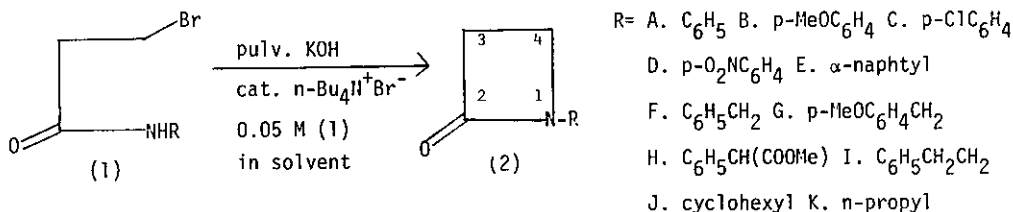
A CONVENIENT SYNTHESIS OF MONOCYCLIC β -LACTAMS

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Abstract: The intramolecular N-alkylation of β -bromopropionamide (1) under phase transfer conditions gave monocyclic β -lactams (2) in high yields.

Synthesis of 2-azetidinones, the basic structure unit of the β -lactam antibiotic, is attractive to organic chemists as a synthetic target. Formation of these four-membered rings has been approached from nearly every conceivable way.¹ In continuation of our work on N-alkylation of lactams under phase transfer conditions,² we now report a facile synthesis of β -lactams by the formation of N-C₄ bond, which mimics the proposed biosynthesis,³ by cyclization of β -bromopropionamides (1), readily available from coupling of β -bromopropionylchloride with amines,⁴ in solid-liquid system.⁵



A typical procedure for the formation of β -lactams is as follows. To a suspension of pulverized KOH (5.5 mmol) and n-Bu₄N⁺Br⁻ (1 mmol) in 50 ml of dry CH₂Cl₂ at room temperature was added a solution containing N-3-bromopropionyl-2-phenylglycine methyl ester (1H) (5 mmol) in 50 ml of dry CH₂Cl₂ over 6 hr with stirring. After completion of the addition, the reaction mixture was stirred for 30 min. The precipitate was filtered off and then washed with CH₂Cl₂. After removal of the combined solvent, the residue was purified by column chromatography on silica gel (CH₂Cl₂-MeOH) (50:1) to give the desired β -lactam (2H) in 83 % yield. The spectral data [ν c=O (neat) 1735 and 1725 cm⁻¹; δ (CDCl₃) 2.83-3.27 (3H, m), 3.67 (1H, t, J=3.5 Hz), 3.83 (3H, s, COOMe), 5.67 (1H, s, CHCOOMe), 7.45 (5H, s, ArH); m/e 219 (M⁺)] and elemental analysis (molecular formula C₁₂H₁₃NO₃) supported this assignment.

Table

Compd. ^a	mp (°C) ^b	Yield (%)	IR ν c=O (cm ⁻¹) ^c
(2A)	79-81	94	1730
(2B)	104-5	92	1725
(2C) ^d	137-9	94	1730
(2D) ^d	162-4	81	1745
(2E) ^d	52-3	91	1740
(2F)	oil	86	1740
(2G) ^d	oil	85	1745
(2H)	oil	83	1735
(2I) ^e	oil	83 (50) ^f	1740
(2J) ^e	oil	63 (74) ^f	1730
(2K)	oil	67 (94) ^f	1745

a) All new compounds gave satisfactory elemental analyses.

b) The oil compounds were purified by chromatography, because the distillation resulted in partial decomposition of β -lactams.

c) (2A-E) (nujol), (2F-K) (neat).

d) The solvent (CH₂Cl₂: CH₃CN) (19:1) was used.

e) The acrylamides of by-products were obtained in 14% and 6% yields.

f) The THF was used as a solvent.

The results are summarized in Table. The cyclization of amides (1) to the β -lactams (2) was dependent on the concentration of the solution, the addition rate of amides to the base and the solvent. Both the high concentration over 0.05 M and the rapid addition of the amides resulted in low yields of β -lactams, along with the formation of substantial amounts of N-alkyl acrylamides. With regard to the solvent employed, the β -lactams (2A-I) could be prepared in good yields by using CH₂Cl₂, however, the use of THF for their cyclization caused complication. On the other hand, the use of THF was favoured over that of CH₂Cl₂ for cyclization of (1J) and (1K). In particular, this reaction proceeded at room temperature, the procedure is simple, straightforward, and easy to work up, the formation of by-products was scarcely caused, and the desired products were obtained in high yields. In conclusion, this procedure under phase transfer conditions is more convenient for N-alkyl monocyclic β -lactams syntheses.^{1,6,7,8} In addition, the β -lactams thus readily obtained have high reactivities (Fries rearrangement,⁹

C-C bond formation at C₃-position,¹⁰ azidation at C₃-position,¹¹ oxidation at C₄-position,¹² and so on¹³). Therefore, they would be served as potential synthetic intermediates as well as those of natural products containing the β-lactam ring.¹¹

References and Notes

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