

STRUCTURE OF PELIRINE AND CHEMICAL CONVERSION OF GARDNERINE
TO 11-METHOXY-16-EPIAFFININE

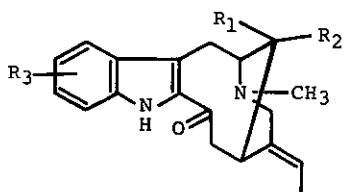
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Abstract—The structure of pelirine, isolated from roots
of *Rauwolfia perakensis* was shown to be 10-methoxyepi-
affinine by comparison with the spectral data of 11-
methoxyepiaffinine which was synthesized from the known
alkaloid gardnerine. The proposed structure has been
confirmed by X-ray crystallography.

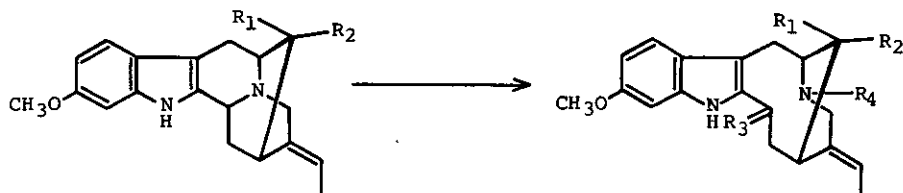
Pelirine was isolated from the roots of *Rauwolfia perakensis*¹ and was assumed to be
one of the methoxy substituted affinine derivatives belonging to be the 2-acyl
indole alkaloid.^{2,3,4}

The location of the methoxy group on the aromatic ring and the configuration of
the hydroxymethyl group on C-16 are still uncertain.



$R_1 = \text{CH}_2\text{OH}, R_2 = R_3 = \text{H}:$	1 affinine
$R_1 = R_3 = \text{H}, R_2 = \text{CH}_2\text{OH}:$	2 16-epiaffinine
$R_1 = \text{H}, R_2 = \text{CH}_2\text{OH},$ $R_3 = \text{OCH}_3 \text{ on C-11}$	3
$R_1 = \text{H}, R_2 = \text{CH}_2\text{OH},$ $R_3 = \text{OCH}_3 \text{ on C-10}$	4 pelirine

In this paper we report the partial synthesis of 11-methoxy-16-epiaffinine **3** from
gardnerine **5** and that the structure of pelirine was determined to be 10-methoxy-
16-epiaffinine **4** from comparison of the spectral data with those of **3** and finally



R ₁ = CH ₂ OH, R ₂ = H: 5	R ₁ = H, R ₂ = CH ₂ OAc,	R ₁ = H, R ₂ = CH ₂ OSi(CH ₃) ₂ t-Bu,
R ₁ = CHO, R ₂ = H: 6	R ₃ = OCH ₃ , H, R ₄ = CN: 10	R ₃ = O, R ₄ = H: 13
R ₁ = H, R ₂ = CHO: 7	R ₁ = H, R ₂ = CH ₂ OAc,	R ₁ = H, R ₂ = CH ₂ OSi(CH ₃) ₂ t-Bu,
R ₁ = H, R ₂ = CH ₂ OH: 8	R ₃ = O, R ₄ = CN: 11	R ₃ = O, R ₄ = CH ₃ : 14
R ₁ = H, R ₂ = CH ₂ OAc: 9	R ₁ = H, R ₂ = CH ₂ OH,	R ₁ = H, R ₂ = CH ₂ OH,
	R ₃ = O, R ₄ = H: 12	R ₃ = O, R ₄ = CH ₃ : 3

5 → 6 + 7: NCS (1.8 eq), Me₂S (1.85 eq) in toluene : DMF (10 : 1), -27° to -20°C, 5 h, and then addition of Et₃N (2 eq). 6 → 7: aq 0.1N KOH : MeOH (1 : 1), rt, 5 h. 7 → 8: NaBH₄ (excess) in MeOH, rt, 3 h. 8 → 9: Ac₂O/Py., rt, 7 h. 9 → 10: BrCN (6 eq), Na₂CO₃ (10 eq) in 10% MeOH-CHCl₃, rt, 1 h. 10 → 11: t-BuOCl (1.1 eq) in CH₂Cl₂ with cat. amount of Et₃N, 0°C to rt, 5.5 h; aq N HCl, 1.5 h, rt; 11 → 12: refluxing in aq N HCl, 5 h. 12 → 13: ClSi(CH₃)₂t-Bu (2 eq), imidazole 2.5 eq in DMF, rt, 4 h. 13 → 14 + 3: 10% Pd-C/H₂, 37% aq H₂CO in dioxane, rt, 4 h.

by X-ray crystallography.

Gardnerine 5 gave a mixture of gardneral 6 (in 13% yield),⁵ 16-epigardneral 7 (in 46% yield)⁶ and the starting alkaloid 5 (16%) by the use of NCS-Me₂S oxidation. 16-Epigardnerine 8⁷ was produced easily from 16-epigardneral 7 by NaBH₄ reduction. 16-Epigardnerine 8 was also formed from gardnerine³ in a one pot reaction (in 48% yield). The acetate 9⁸ was smoothly reacted with BrCN and gave rise to a single ring-opened compound 10⁹ in a quantitative yield. t-Butyl hypochlorite reacted with 10 and, without purification, the resulting chloro derivative was hydrolyzed to give a 2-acylindole derivative. After evaporation of the organic solvent used for extraction the hydrolyzed compound was gradually converted to the 2-acylindole derivative 11¹⁰ in desiccator. After purification by chromatography on SiO₂ column, 11 was obtained in 32% yield as needles. Hydrolysis of 11 with strong acid provided the corresponding secondary amine alcohol 12¹¹ in 80% yield. After converting 12 to the silyl ether 13¹², N-methylation was carried out to provide a mixture of 11-methoxy-16-epiaffinine-O-silylether 14¹³ (in 22% yield) and the desired 11-methoxy-16-epiaffinine 3¹⁴ (in 48% yield).

		M ⁺	M ⁺ -H ₂ O	M ⁺ -CH ₂ OH	base peak
affinine	1 ¹⁵	324		293	152
11-methoxyaffinine ^{a,16}		354(79)	336(10)	326(56)	152(100)
16-epiaffinine	2 ¹⁷	324(1)	306(33)		152(100)
11-methoxy-16-epiaffinine	3 ^a	354(3)	336(61)		152(100)
pelirine	4 ^a	354(3)	336(88)		152(100)

a) These compounds were measured by use of direct inlet method this time.

Table 1. Mass Spectral Data of Affinine Derivatives.

Affinine 1 and 11-methoxyaffinine¹⁶ whose primary alcohol groups are oriented to the indolic side exhibited strong peaks of M⁺ and M⁺-CH₂OH besides the base peak at m/z 152.¹⁷ However, 16-epimeric derivatives, 2 and 3, characteristically revealed a strong peak of M⁺-H₂O and a weak peak of M⁺ besides the base peak as shown in Table 1. Since pelirine 4 showed the latter tendency, we propose its structure to be 10-methoxy-16-epiaffinine 4. The ¹³C-nmr signals of pelirine 4 indicate similar values compared with those of the synthesized model compound 3 except for the signals of aromatic and conjugated carbons as shown in Table 2. ¹H-nmr spectra of compounds 3 and 4 also showed excellent similarity with each other except for the aromatic protons.

Finally, in order to confirm the proposed structure of pelirine 4, X-ray crystallography was used and the deduced structure was proved to be correct.¹⁸ The structure was solved by direct method MULTAN¹⁹ and refined by the full matrix least-squares method to R=0.047.. The ORTEP drawing of 4 is shown in Fig 1.

No	3	4
2	137.8	136.0
3	190.0	200.0
5	57.2	57.1
6	19.5	19.3
7	121.5	119.8
8	123.0	128.7
9	121.5	100.6
10	112.5	154.7
11	160.2	113.2
12	93.6	118.6
13	134.9	132.0
14	43.4	43.5
15	31.7	31.8
16	38.2	38.1
17	67.9	67.9
18	12.1	12.1
19	120.9	120.9
20	135.3	135.2
21	52.8	52.1
N-Me	42.0	42.0
O-Me	55.6	55.9

¹³C-NMR spectra of 3 and 4. Table 2.

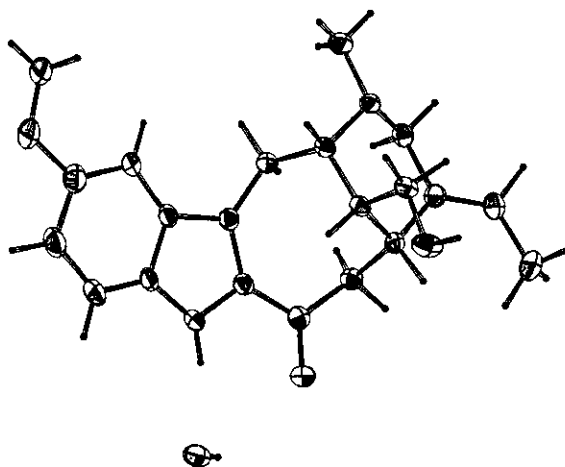


Fig. 1. ORTEP of Pelirine 4

The CD spectrum of 4 showed a curve similar to that of 3 derived from gardnerine 5 having known absolute configuration.²⁰ Therefore, pelirine 4 has the common indole alkaloid configuration.

REFERENCES AND NOTES

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- 2) J.A. Weisbach and B. Douglas, *Chem. Ind. (London)*, 1965, 623.
- 3) S. Sakai, *Heterocycles*, 1976, 4, 131.
- 4) D. G.I. Kingston and O. Ekundayo, *J. Nat. Prod.*, 1981, 44, 509.
- 5) mp 278-280°C, $\nu_{\text{C=O}}$ 1695cm⁻¹, δ (CDCl₃) 9.14 (1H, d, J=0.9Hz), Mass m/z(%): 322(M⁺, 100), 293(M⁺-29, 42), 279(24), 212(15), 199(70), 198(50).
- 6) mp 276-278°C, $\nu_{\text{C=O}}$ 1715cm⁻¹, δ 9.64(1H, d, J=0.9Hz).
- 7) mp 284-287°C, C₂₀H₂₄N₂O₂:0.5Me₂CO, $\delta_{\text{D}}^{18} = -4.3^\circ$ (MeOH), m/z(%): 324(M⁺, 100), 323(M⁺-1, 99), 293(40), 199(69), 198(47).
- 8) mp 234-237°C, C₂₂H₂₆N₂O₃, δ_{OCOMe} 1.97(s).
- 9) amorphous, $\delta_{\text{C3-OMe}}$ 3.22(s), $\delta_{\text{C11-OMe}}$ 3.79(s), δ_{OCOMe} 1.94(s).
- 10) mp 219-220°C, C₂₃H₂₅N₃O₄, m/z(%): 407(M⁺, 28), 202(100), 160(64), $\delta_{\text{C11-OMe}}$ 3.88(s).
- 11) amorphous base, $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 343, 297, 264, 225.
- 12) amorphous, $\delta_{\text{Si-tBu}}$ 0.86(9H, s), $\delta_{\text{C11-OMe}}$ 3.87(3H, s).
- 13) amorphous, $\delta_{\text{Si-tBu}}$ 0.76(9H, s), 2.47(3H, s, N-Me), 3.84(3H, s, C11-OMe), m/z(%): 454(M⁺, 19), 397(14), 297(29), 296(100), 295(19).
- 14) mp 132-135°C, Mass m/z: Calcd for C₂₁H₂₆N₂O₃, 354.1942, Found 354.1933, $\nu_{\text{C=O}}$ 1620cm⁻¹, $\lambda_{\text{max}}^{\text{MeOH}}$ nm(log ϵ): 342(4.32), 262(3.79), 232(sh, 4.13), 215(sh, 4.34), δ : 9.02(NH, br s), 3.27(C5-H, br dd, J=10, 7Hz), 3.37(C6-H, dd, J=14.3, 7.3), 3.51(C6-H, dd, J=14.3, 10.4), 7.56(C9-H, d, J=8.9), 6.83(C10-H, dd, J=8.9, 2.1), 6.77(C12-H, d, J=2.1), 2.63(C14-H, dd, J=11.9, 7.3), 3.30(C14-H, dd, t.like, J=11.9, 11.8), 3.08(C15-H, br dd, J=11.8, 7.3), 1.97(C16-H, m), 3.59(C17-H, dd, J=10.7, 4.3), 3.66(C17-H, J=10.7, 5.2), 1.70(C18-H₃, dd, J=6.7, 1.8), -5.47(C19-H, br q, J=6.7), 3.01(C21-H, br d, J=13.7), 3.68(C21-H, br d, J=13.7), 2.55(N-Me, s), 3.87(OMe, s).
- 15) R.H. Burnell and J.D. Medina, *Canad. J. Chem.*, 1971, 49, 307.
- 16) S. Sakai, A. Kubo, K. Katano, N. Shinma, and K. Sasago, *Yakugaku Zasshi*, 1973, 93, 1165.
- 17) J. Naranjo, M. Pinar, M. Hesse, and H. Schmid, *Helv. Chim. Acta*, 1972, 55, 752.
- 18) X-ray data: pale yellow plates, C₂₁H₂₆O₃N₂:H₂O, mp 130-131°C. Crystals of 4 belong to monoclinic space group P2₁ with cell constant of a = 19.0108(42)Å, b = 7.7596(13)Å, c = 6.8517(19)Å. A total of 2254 unique independent intensities were measured within the range of 3° < 2 θ < 155° on a 4-circle diffractometer (Rigaku AFC-5) using Cu K α radiation (λ = 1.54Å). The structure was solved by the direct method using MULTAN 80 (UNICS III system) and refined anisotropically (isotropically for H) by the least-squares method using the 1789 reflections with $F_o \geq 3\sigma(F_o)$.
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