

HETEROCYCLES, Vol. 60, No. 5, 2003, pp. 1045 - 1048

Received, 23rd January, 2003, Accepted, 28th February, 2003, Published online, 3rd March, 2003

A NOVEL HETEROCUMULENIC PAUSON-KHAND REACTION OF ALKYNILCARBODIIMIDES: A FACILE AND EFFICIENT SYNTHESIS OF HETEROCYCLIC RING-FUSED PYRROLINONES

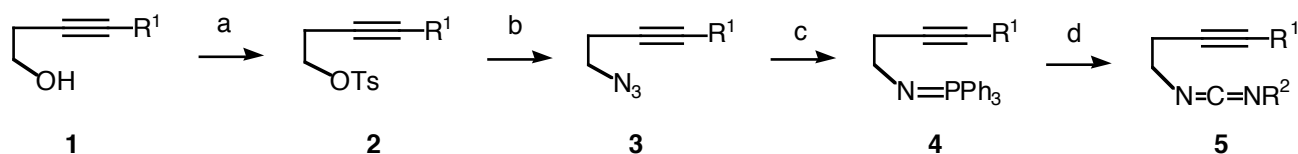
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Abstract – The first examples of the heterocumulenenic Pauson-Khand reaction are described. Alkynylcarbodiimides undergo an intramolecular Pauson-Khand cycloaddition upon heating in the presence of $\text{Mo}(\text{CO})_6$ and DMSO to afford 4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-ones and 1*H*-pyrrolo[2,3-*b*]indol-2-ones.

The Pauson-Khand (P-K) reaction is a three-component reaction, formulated as a formal [2+2+1] cycloaddition of an alkyne, an alkene, and carbon monoxide leading to a cyclopentenone,¹ and has been exploited extensively.² Recently, allenes have also been utilized successfully in the P-K reaction to give 4- and/or 5-alkylidenecyclopentenones.³ Curiously, however, no examples of the use of a heterocumulenenic system in place of the allene component in either inter- or intramolecular P-K reaction have been reported so far, despite the fact that it would constitute a novel and potent method for straightforward construction of five-membered heterocycles or, in some cases, carbocycles in light of the wide applicability and versatility imparted by this metal carbonyls-assisted cycloaddition. In this context, we have succeeded in performing a heterocumulenenic P-K cycloaddition which incorporates a carbodiimide functionality and an internal alkyne moiety together with metal carbonyls. We report here for the first time the heterocumulenenic P-K reaction.⁴ The present P-K reaction also offers a novel and efficient synthetic method for 4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-ones and 1*H*-pyrrolo[2,3-*b*]indol-2-ones.

In order to examine the feasibility of the heterocumulenenic P-K reaction, we initially selected simple alkynylcarbodiimides (**5**) as the key substrates. The carbodiimides (**5**) were readily prepared from commercially available alkynols (**1**) by (a) tosylation, (b) nucleophilic substitution by an azide group, (c) Staudinger reaction of azides (**3**) with triphenylphosphine, and by (d) aza-Wittig reaction of iminophosphoranes (**4**) with isocyanates (Scheme 1).



Scheme 1. Reagents and conditions: (a) TsCl, Et₃N, CH₂Cl₂, 0 °C → rt, 3 h, (b) NaN₃, DMF, rt, 3-12 h (82-86 %, two steps), (c) Ph₃P, benzene, 80 °C, 2 h, (d) R²NCO, benzene, rt, 1 h (27-54 %, two steps).

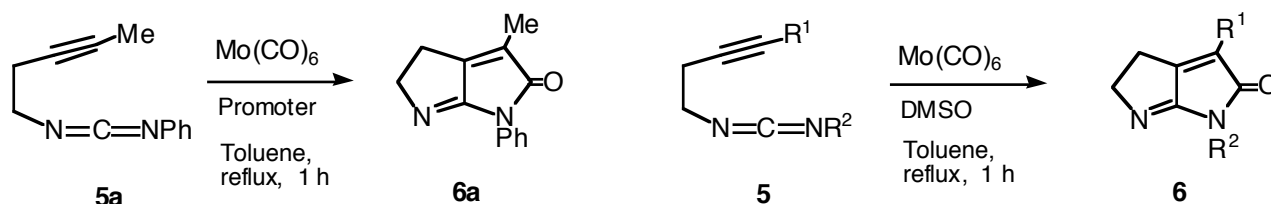


Table 1. Efficiency of promoters in the Pauson-Khand reaction of **5a**^a

Entry	Promoter	Yield of 6a (%) ^b
1	None	0
2	NMO	Trace
3	MeCN	21
4	DMF	51
5	DMSO	60

- a) The reactions were carried out using 1.1 equiv. Mo(CO)₆ and 5 equiv. promoters.
 b) Isolated yield.

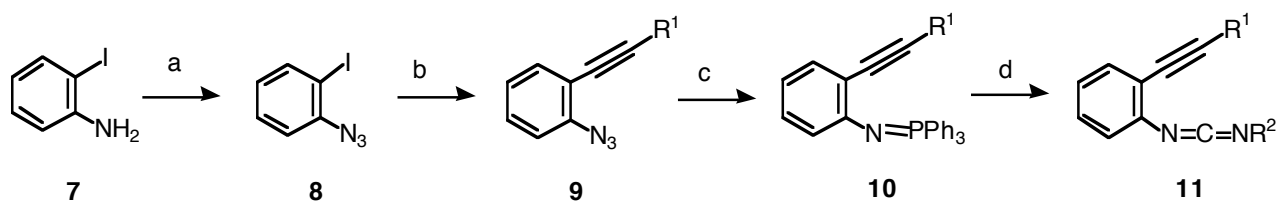
Table 2. Heterocumulenic Pauson-Khand reaction of alkyne carbodiimides (**5**) to give pyrrolopyrrolones (**6**)^a

Entry	R ¹	R ²	Yield of 6 (%) ^b
a	Me	Ph	60
b	Me	<i>p</i> -Tol	53
c	Me	<i>p</i> -ClC ₆ H ₄	57
d	Et	Ph	42
e	Et	<i>p</i> -Tol	47
f	Et	<i>p</i> -ClC ₆ H ₄	49

- a) The reactions were carried out using 1.1 equiv Mo(CO)₆ and 5 equiv. DMSO.
 b) Isolated yield.

With alkyne carbodiimides (**5**) in hand, we attempted the P-K reaction of **5a** first using dicobaltoctacarbonyl. Disappointingly, all attempts to carry out the P-K reaction under various conditions such as thermal and/or promoter [*N*-methylmorpholine oxide (NMO),⁵ DMSO⁶]-assisted conditions were unsuccessful.[†] However, the reactions using molybdenumhexacarbonyl² in the presence of some promoters indeed proceeded to yield the expected P-K product, 4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-one (**6a**).⁷ The representative results are shown in Table 1. Among the promoters tested under refluxing toluene for 1 h, dimethyl sulfoxide⁸ was found to provide the best yield of **6a** (60 %, Entry 5). The Mo(CO)₆-mediated P-K reaction procedure with the aid of DMSO under the same reaction conditions was also applied to some other alkyne carbodiimides (**5b-f**). The P-K reactions worked well to give 1*H*-pyrrolo[2,3-*b*]pyrrolin-2-ones (**6b-f**) (Table 2),[‡] suggesting that this P-K methodology seems to be promising for the synthesis of variously substituted 4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-ones.

In order to probe the generality of this carbodiimide-P-K methodology, we next investigated the reaction of alkyne carbodiimides (**11**) in which both carbodiimide and alkyne functionalities were each connected with an *ortho*-phenylene group. Scheme 2 illustrates the preparation of alkyne carbodiimides (**11**) from commercially available *o*-iodoaniline (**7**) via (a) diazotization and azide-replacement, (b) Heck coupling with an alkyne, followed by (c) Staudinger reaction with triphenylphosphine and (d) aza-Wittig reaction



Scheme 2. Reagents and conditions: (a) i, NaNO_2 -aq. HCl, 5-10 °C, ii, NaN_3 , H_2O , 5 °C, 2 h (84 %), (b) HCCR^1 , $\text{PdCl}_2(\text{PPh}_3)_2$, CuI, Et_3N , CH_2Cl_2 , rt, 1 h (99 %), (c) Ph_3P , benzene, rt, 1.5 h (71-95 %), (d) R^2NCO , benzene, rt, 3-12 h (60-76 %).

with isocyanates. Treatment of carbodiimide (**11a**) with $\text{Mo}(\text{CO})_6$ in the presence of a promoter under the same conditions above afforded 3-phenyl-1-propyl-1*H*-pyrrolo[2,3-*b*]indol-2-ones (**12a**) as the expected P-K product (Table 3). Pentamethylene sulfide actually promoted the P-K reaction but was less effective with either $\text{Mo}(\text{CO})_6$ or $\text{Co}_2(\text{CO})_8$ contrary to our expectation (Entry 5).⁹ Again, DMSO was found to be a good promoter in this reaction. The results of the DMSO-promoted $\text{Mo}(\text{CO})_6$ -P-K reactions of **11a-f** are summarized in Table 4.^z Although the $\text{Mo}(\text{CO})_6$ -mediated P-K reaction of **11** having an unsubstituted acetylene ($\text{R}^1 = \text{H}$) failed to afford compound (**12**) ($\text{R}^1 = \text{H}$), the trimethylsilyl group in **12d-f** could be removed to provide 3-unsubstituted 1*H*-pyrrolo[2,3-*b*]indol-2-ones (**12**).

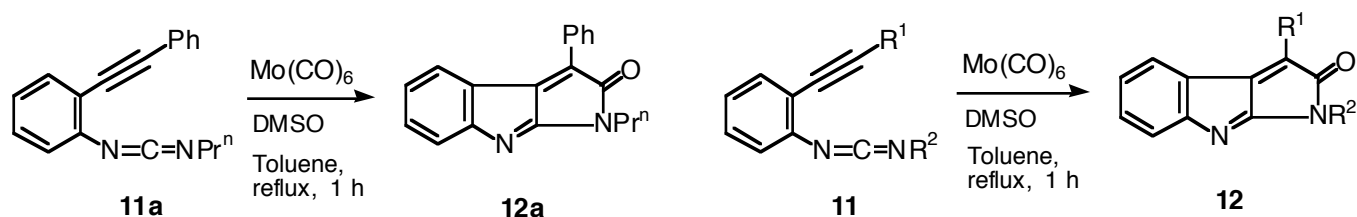


Table 3. Efficiency of promoters in the Pauson-Khand reaction of **11a**^a

Entry	Promoter	Yield of 12a (%) ^b
1	None	0
2	NMO	51
3	DMF	44
4	DMSO	55
5	$(\text{CH}_2)_5\text{S}$	28

- a) The reactions were carried out using 1.1 equiv. $\text{Mo}(\text{CO})_6$ and 5 equiv. promoters.
b) Isolated yield.

Table 4. Heterocumulenic Pauson-Khand reaction of alkynylcarbodiimides (**11**) to give pyrroloindolones (**12**)^a

Entry	R^1	R^2	Yield of 12 (%) ^b
a	Ph	<i>n</i> -Pr	55
b	Ph	Ph	45
c	Ph	<i>p</i> -Tol	48
d	Me_3Si	<i>n</i> -Pr	50
e	Me_3Si	Ph	60
f	Me_3Si	<i>p</i> -Tol	70

- a) The reactions were carried out using 1.1 equiv. $\text{Mo}(\text{CO})_6$ and 5 equiv. DMSO.
b) Isolated yield.

In conclusion, we have developed a heterocumulenic Pauson-Khand reaction for the first time. The synthetic usefulness of this methodology was demonstrated in the facile and efficient synthesis of 4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-ones and 1*H*-pyrrolo[2,3-*b*]indol-2-ones by using alkynylcarbodiimides as the substrates in the intramolecular cycloaddition.

Further studies on the heterocumulenic Pauson-Khand methodology to provide a variety of heterocycles are under way.

† Complexation of the alkyne moiety of **5a/11a** with dicobaltoctacarbonyl seems to be successful, but attempts to promote the cycloaddition resulted in decomposition

≠ The reaction in refluxing toluene gave the P-K products (**6/12**) albeit in relatively low yields (42-70 %) accompanying polymerization or deterioration of **5/11**, while in refluxing benzene (at about 80 °C) the reaction hardly occurred.

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7. A mixture of carbodiimide (**5a**) (69 mg, 0.375 mmol), Mo(CO)₆ (109 mg, 0.412 mmol) and DMSO (0.13 mL, 1.88 mmol) in toluene (5 mL) was heated under reflux (120 °C) for 1 h. After being cooled, the mixture was evaporated and the residue was briefly chromatographed with dichloromethane as eluant on a short silica gel column and further purified by thin-layer silica gel chromatography (50 v/v % AcOEt-hexane) to give 3-methyl-1-phenyl-4,5-dihydro-1*H*-pyrrolo[2,3-*b*]pyrrolin-2-one (**6a**) (48 mg, 60 % yield); mp 94.2-94.8 °C; IR (KBr) 1728, 1540 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 2.03 (t, J = 2.2 Hz, 3H), 2.80 (ddq, J = 4.4, 4.4, 2.2, 2H), 4.36 (dd, J = 4.4, 4.4, 2H), 7.24-7.28 (m, 1H), 7.32-7.46 (m, 2H), 7.68-7.70 (m, 2H); ¹³C NMR (100 MHz, DEPT, CDCl₃) δ 9.62 (q), 24.43 (t), 62.82 (t), 123.44 (dx2), 123.97 (s), 126.23 (d), 128.97(dx2), 134.50(s), 145.80(s), 166.70(s), 174.82(s); HRMS found: *m/z* 212.0956, calcd for C₁₃H₁₂N₂O: M 212.0950.
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