

## SYNTHESIS OF STRAINED 8-MEMBERED HETEROCYCLIC ALLENES BY [3,3] SIGMATROPIC REARRANGEMENT AND THEIR REACTIVITIES<sup>1</sup>

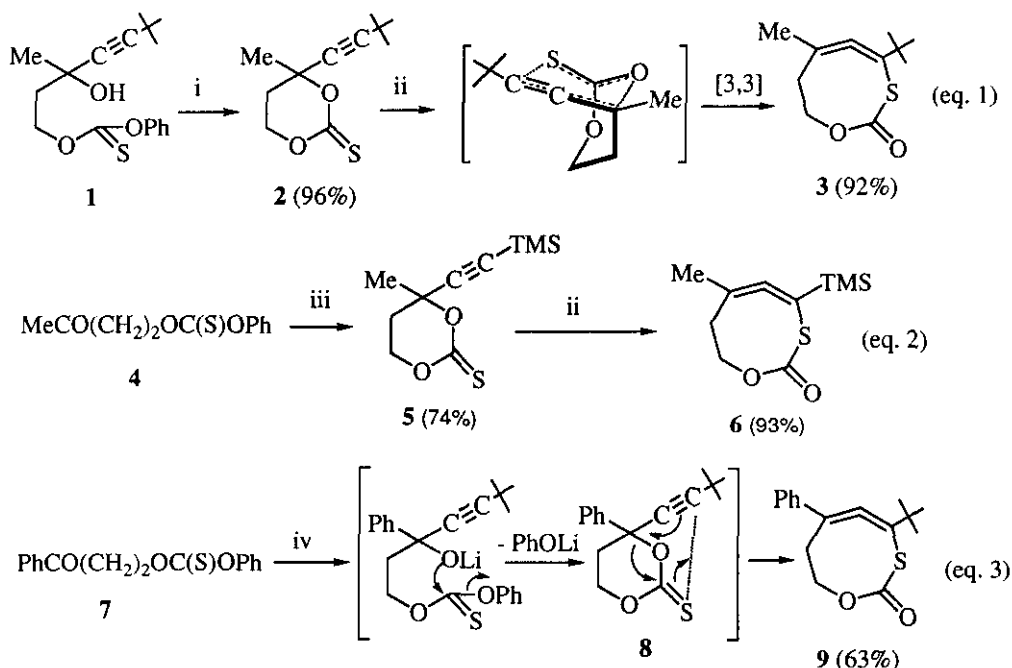
Shinya Harusawa, Hideki Moriyama, Hirofumi Ohishi, Ryuji Yoneda, and Takushi Kurihara\*

*Osaka University of Pharmaceutical Sciences, 2-10-65 Kawai, Matsubara, Osaka 580, Japan*

**Abstract** - A new type of 8-membered heterocyclic allenes were synthesized by the [3,3] sigmatropic ring expansion of 6-membered cyclic thionocarbonates. The MNDO optimized structure indicated the allenyl moiety in cyclic allene (**3**) is bent and strained. Reactivities of the heterocyclic allenes were also examined.

Recent interests have focused on attempts to synthesize the highly strained cyclic allenes and to study their properties.<sup>2-5</sup> 1-*tert*-Butyl-1,2-cyclooctadiene<sup>3</sup> has been known as the smallest isolable carbocyclic allene. Two groups<sup>4</sup> independently reported synthesis of 6-membered silacyclic allenes bridged by long Si-Si bonds (2.3-2.4 Å). Recently, we reported<sup>1</sup> a synthesis of medium-membered heterocyclic allenes by the [3,3] sigmatropic ring expansion of cyclic thionocarbonates. In continuation of the study for the synthesis of other cyclic allenes, we report herein the synthesis and reactivities of the 8-membered cyclic allenes containing a thiolcarbonate moiety. Although it has been reported that the strained cyclic allenes easily dimerize ([2+2] cycloaddition),<sup>2</sup> we anticipated that bulky alkyl groups on the allene could make possible to isolate the labile 8-membered molecule. Treatment of diolmonothionocarbonate (**1**) with lithium bis(trimethylsilyl)amide [(TMS)<sub>2</sub>NLi] in THF at room temperature afforded the 6-membered thionocarbonate (**2**) (96%) (Scheme 1, eq.1). The [3,3] sigmatropic rearrangement of **2** expectedly proceeded in refluxing benzene for 1.5 h and the pure 8-membered heterocyclic allene (**3**)<sup>6</sup> (92%) was obtained. The <sup>1</sup>H- and <sup>13</sup>C-nmr spectra of **3** showed the four non-equivalent methylene protons and the characteristic allenic *sp* and *sp*<sup>2</sup> carbons at 202.9, 117.2, and 102.7 ppm, respectively. This is a new type of isolable 8-membered cyclic allene containing divalent sulfur and oxygen atoms. It should be noted that the sigmatropic rearrangement has been rarely used to generate the strained cyclic allenes.<sup>2</sup> The structure of **3** was estimated from MNDO calculation (Figure 1). The bond angle on the *sp* carbon, C1-C2-C3, is bent from linearity to 170.1°. The dihedral angle, CH<sub>3</sub>-C1-C3-*t*Bu, is strongly twisted (64.3°) from vertical geometry. The bond angle on the *sp* carbon of **3** corresponds to that of carbon skeletal nine-membered cyclic allene estimated from MNDO calculation by Johnson *et al.*<sup>7</sup> Two 8-membered cyclic allenes (**6**, **9**) containing a TMS or a phenyl group could be similarly synthesized from ketones (**4**, **7**) (Scheme 1, eq. 2 and 3). Interestingly, treatment of the

ketone (7) with lithium acetylide directly afforded **9** (63%) via the sigmatropic rearrangement of the 6-membered intermediate (**8**).

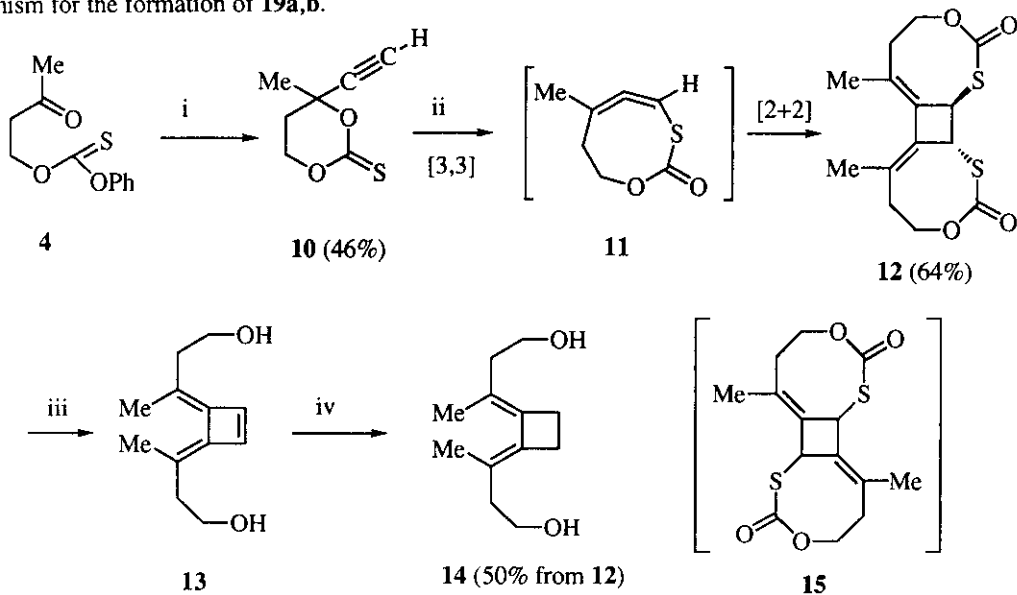


**Scheme 1.** Reagent and conditions: i,  $(\text{TMS})_2\text{NLi}$  (1.0 eq), THF, room temperature for 1.5 h; ii, benzene, reflux for 1.5 h; iii,  $\text{TMSC}\equiv\text{CLi}$ , THF,  $-78^\circ\text{C}$  for 20 min.; iv,  $t\text{BuC}\equiv\text{CLi}$ , THF,  $-78^\circ\text{C} \rightarrow$  room temperature for 25 min

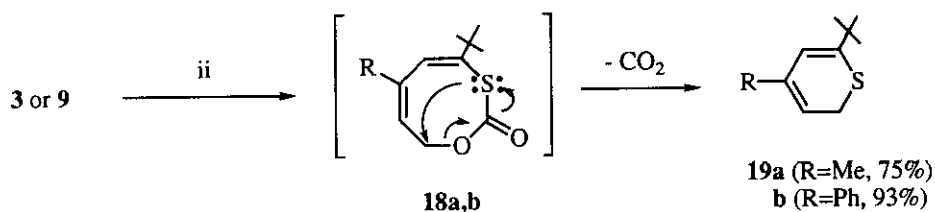
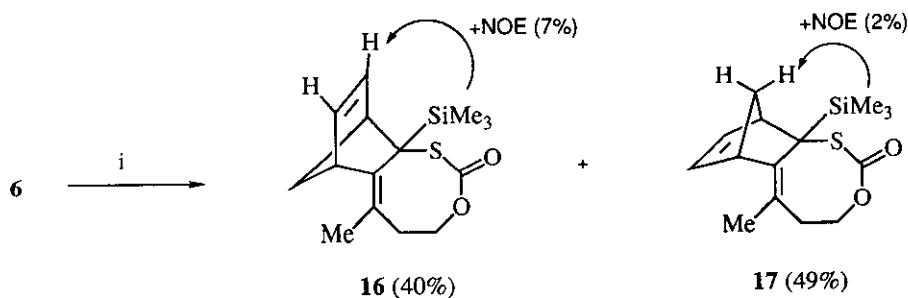
We next investigated a synthesis of the cyclic allene (**11**) lacking a bulky substituent (Scheme 2). Refluxing the cyclic thionocarbonate (**10**), prepared from **4** and ethynylmagnesium bromide, in benzene afforded the [2+2]dimer (**12**) (64%, mp  $220\text{--}231^\circ\text{C}$ ). The formation of **12** is reasonably explained by the dimerization of the intermediate (**11**). We could not initially distinguish between structure (**12**) (head-to-head dimer) and other possible dimerization product (**15**) (head-to-tail dimer). Thus,  $\text{SmI}_2$  reduction<sup>1</sup> of **12** followed by catalytic reduction of a cyclobutene derivative (**13**) were carried out to yield a diol (**14**)<sup>8</sup> (50% from **12**). This sequence clearly demonstrates that **12** is a head-to-head dimer. Further, X-ray analysis of **12** confirmed unequivocally the *trans*-configuration of the four-membered ring (Figure 2).

Reactivities of the heterocyclic allenes were also examined (Scheme 3). Diels-Alder cycloaddition of **6** with cyclopentadiene under uncatalyzed thermal conditions afforded two stereoisomeric products (**16**) (40%) and (**17**) (49%), whose structures were determined by  $^1\text{H}$  NOE experiments. Intriguing point is the formation of 2*H*-

thiopyran derivatives (**19a**) (75%), (**b**) (93%)<sup>9</sup> by treatments of **3** or **9** with DBU, respectively. Isomerization of the double bond and subsequent CO<sub>2</sub> elimination of the resulting cyclic dienes (**18a, b**) may be a plausible mechanism for the formation of **19a, b**.



**Scheme 2.** Reagents and conditions: i, CH≡CMgBr, THF, 0°C; ii, benzene, reflux for 2 h; iii, SmI<sub>2</sub> - HMPA, <sup>t</sup>BuOH; iv, H<sub>2</sub> / Pd-C, EtOH



**Scheme 3.** Reagents and conditions: i, cyclopentadiene, benzene, 60-70°C for 31 h; ii, DBU (1.0 eq), benzene, room temperature <5 min

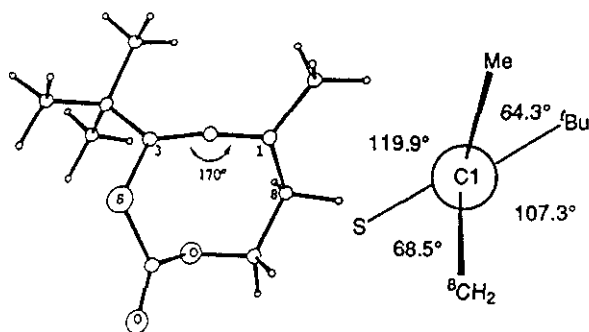


Figure 1. MNDO Estimated Structure of **3** with Dihedral Geometry around the Allenic Moiety

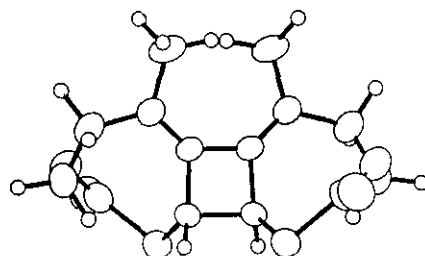


Figure 2. X-Ray Determined Structure of **12**

## REFERENCES AND NOTES

1. This is part 11 of the series entitled "[3,3] Sigmatropic Ring Expansion of Cyclic Thionocarbonates". Part 10, see, S. Harusawa, N. Kase, R. Yoneda, and T. Kurihara, *Tetrahedron Lett.*, 1994, **35**, 1255.
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3. J.D. Price and R.P. Johnson, *Tetrahedron Lett.*, 1986, **27**, 4679.
4. a) Y. Pang, S.A. Petrick, V.G. Young, Jr., M.S. Gordon, and T.J. Barton, *J. Am. Chem. Soc.*, 1993, **115**, 2534. b) T. Shimizu, F. Hojo, and W. Ando, *ibid.*, 1993, **115**, 3111.
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6. **3**, a colorless oil, ir (neat) 1712  $\text{cm}^{-1}$  (CO).  $^1\text{H-Nmr}$  (200 MHz,  $\text{CDCl}_3$ ): 1.16 (9H, s), 1.68 (1H, ddd,  $J = 13.8, 2.6, 2.1$  Hz), 1.82 (3H, s), 2.52 (1H, ddd,  $J = 13.8, 10.6, 5.8$  Hz), 4.42 (1H, ddd,  $J = 12.0, 5.8, 2.1$  Hz), 4.55 (1H, ddd,  $J = 12.0, 10.6, 2.6$  Hz).  $^{13}\text{C-Nmr}$  ( $\text{CDCl}_3$ ): 19.8 (q), 29.0 (t), 29.5 (q), 35.9 (s), 71.3 (t), 102.7 (s), 117.2 (s), 172.0 (s), 202.9 (s). CI-MS  $m/z$ : 213 ( $\text{M}^++1$ ), 184 ( $\text{M}^+-\text{CO}$ ). HR-ms  $m/z$ : Calcd for  $\text{C}_{11}\text{H}_{17}\text{O}_2\text{S}$  213.0948, Found: 213.0946.
7. R.O. Angus, Jr., M.W. Schmitt, and R.P. Johnson, *J. Am. Chem. Soc.*, 1985, **107**, 532.
8. **14**, a colorless oil,  $^1\text{H-nmr}$  ( $\text{CDCl}_3$ ): 1.82 (6H, s), 2.27 (4H, t,  $J = 6.5$  Hz), 2.46 (4H, s), 3.70 (4H, t,  $J = 6.5$  Hz).  $^{13}\text{C-Nmr}$  ( $\text{CDCl}_3$ ): 19.9 (q), 27.1 (t), 38.2 (t), 60.7 (t), 121.7 (s), 138.2 (s).
9. **19b**, a pale yellow oil,  $^1\text{H-nmr}$  ( $\text{CDCl}_3$ ): 1.28 (9H, s), 3.37 (2H, d,  $J = 8.3$  Hz), 5.71 (1H, td,  $J = 8.3, 1.0$  Hz), 6.41 (1H, d,  $J = 1.0$  Hz), 7.21-7.43 (5H, m).  $^{13}\text{C-Nmr}$  ( $\text{CDCl}_3$ ): 26.7 (t), 29.8 (q), 37.9 (s), 111.1 (d), 117.5 (d), 126.8 (d), 127.8 (d), 128.9 (d), 139.8 (s), 141.5 (s), 151.9 (s).

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