

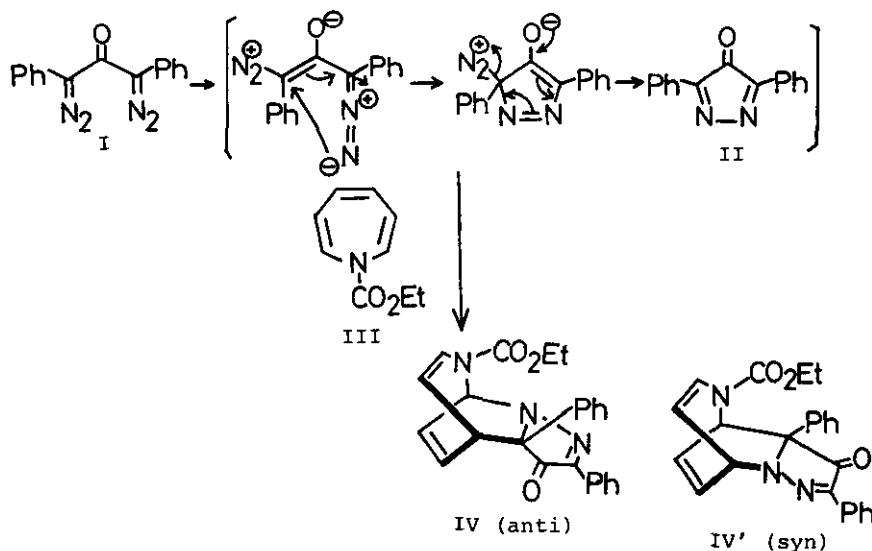
X-RAY STRUCTURE DETERMINATION OF A CYCLOADDUCT OF 1H-AZEPINE  
AND DIAZACYCLOPENTADIENONE

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**Abstract** - The cycloaddition reaction of N-ethoxycarbonyl-azepine with 2,5-diphenyl-3,4-diazacyclopentadienone gave anti-endo[4+2] $\pi$  cycloadduct. The structure was determined by X-ray crystallographic study.

Thermolysis of 1,3-bisdiazo-1,3-diphenyl-2-propanone (I) provided a simple route to 2,5-diphenyl-3,4-diazacyclopentadienone (II) which is served as a convenient heterodienone component in spite of an unstable intermediate. This fascinating intermediate has been shown to act as both a dienophile with 2,3-dimethyl-1,3-butadiene and a diene with norbornene in pericyclic reactions.<sup>1)</sup> These interesting features of the cycloaddition reactions of the diazacyclopentadienone prompted us to examine the mode of cycloaddition of I with seven-membered ring unsaturated compounds; e.g., azepine, diazepine, cycloheptatriene and tropone.



Of those, N-ethoxycarbonylazepine (III) reacted with I to give 1:1 cyclo-

adduct (IV) in a moderate yield. In this communication we present the results of structure confirmation of the adduct (IV).

The  $^{13}\text{C}$ -NMR spectrum in  $\text{C}_6\text{D}_6$  of IV shows five detectable signals of  $\text{sp}^3$  carbons, of which, signals at 38.85, 68.96, and 80.16 ppm are ascribable to the structure of  $[4+2]\pi$  cycloadduct. The UV absorption at  $\lambda_{\text{benzene}}^{\text{max}}$  400 nm ( $\epsilon$  4000) is in good agreement with that expected for the conjugated chromophore ( $\text{N}=\text{C}=\text{C}=\text{O}$ ).

From these data, the structure of IV was considered to be  $[4+2]\pi$  cycloadduct instead of  $[6+4]\pi$  cycloadduct. However, there are some obscurity in regio-chemistry of IV. Assignment of the conjugated carbonyl absorption at  $1725\text{ cm}^{-1}$  by IR spectrum is uncertain for that expected structure. Thus, IV was submitted to single-crystal X-ray analysis.

Single crystals of IV were obtained from ethanol-acetone solution by slow evaporation at room temperature. Unit cell dimensions were obtained from least-squares refinement of the  $2\theta$  angles of 15 reflections measured on a Syntex  $\text{P}\bar{1}$  automated diffractometer ( $\text{Mo}$ ,  $\lambda=0.71069\text{ \AA}$ ). The crystals are triclinic, space group  $\text{P}\bar{1}$ , there are two molecules in the unit cell of dimensions  $a=9.9201$ ,  $b=12.5030$ ,  $c=9.1718\text{ \AA}$ ,  $\alpha=106.40$ ,  $\beta=105.46$ ,  $\gamma=97.41^\circ$ . Intensity data were collected using the  $\theta$ - $2\theta$  scan technique and of the 3530 independent reflections within the range of the diffractometer ( $2\theta > 55^\circ$ ), 2792 were observed. The orientation of the major part of the molecule was found by direct phasing<sup>2)</sup>, but difficulty was experienced in finding a correct phasing model owing to two overlapping molecules. The middle positions of the both models were used for phase determination (29 atoms). The position of remaining atom was obtained from a difference Fourier map.

Refinements were carried out by the block-diagonal least-squares method<sup>3)</sup> using isotropic temperature factors for the hydrogen with exception of the hydrogen atoms of ethyl group with high temperature factors and anisotropic temperature factors for the remaining atoms. Final R value was 0.053 for the observed reflection.

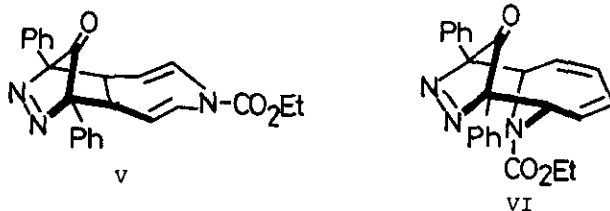
Thus, the molecular structure shown in the Figure establishes that IV has the structure of anti-endo $[4+2]\pi$  cycloadduct from III ( $4\pi$ ) and II ( $2\pi$ ).

The  $\text{C}_{15}-\text{C}_{14}-\text{C}_{22}$  angle is  $4^\circ$  smaller in the diazacyclopentenone than in the cyclopentenone.<sup>4)</sup> This difference might be due to the short bond distance of N-N bond of the diazacyclopentenone as compared with a corresponding cyclopentenone, which gives rise to a higher C=O stretching frequency at  $1725\text{ cm}^{-1}$  in the IR spectrum.



In consideration of the molecular structure of IV, we can consequently explain the PMR spectrum by spin-spin decoupling experiments and study of the solvent-induced shifts of the resonance of olefinic protons. The six methine protons in the PMR spectrum ( $\text{CDCl}_3$  with TMS as internal standard) are ascribed as follows :  $\delta$  3.45 ppm (dd,  $J_{10,11}=J_{7,10}=7\text{Hz}$ ,  $\text{H}_{10}$ ) ; 4.88 (dd,  $J_{11,12}=7\text{Hz}$ ,  $\text{H}_{11}$ ) ; 5.64 (dd,  $J_{8,9}=8\text{Hz}$ ,  $J_{7,8}=6\text{Hz}$ ,  $\text{H}_8$ ) ; 6.39 (d,  $\text{H}_{12}$ ) ; 6.48 (dd,  $\text{H}_9$ ) ; 6.84 (brd,  $\text{H}_7$ ).

It is to be noted that 1H-azepine exhibits the marked polyene character and has been shown to undergo a wide range of cycloaddition reactions wherein the triene system acts as a  $2\pi$ ,  $4\pi$ , or  $6\pi$  component.<sup>5)</sup> Therefore, the adduct is able to consider one of  $[4+2]\pi$  (IV), another type of  $[4+2]\pi$  (V) from azepine( $2\pi$ ) and diazacyclopentadienone( $4\pi$ ), and  $[6+4]\pi$  cycloadduct (VI) from azepine( $6\pi$ ) and diazacyclopentadienone( $4\pi$ ) in thermal conditions. However, in this case, the  $[4+2]\pi$  adduct (IV) is an isolable product and the regiochemistry is anti and endo showing a high degree of reactivity and regioselectivity.



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