

EFFECTIVE CATALYTIC ASYMMETRIC SYNTHESIS OF S-(-)-3-METHOXY-CARBONYL-4-(3,4-METHYLENEDIOXYPHENYL)BUTANOIC ACID. A SIMPLE AND EFFECTIVE ROUTE TO CHIRAL LIGNANS¹⁾

Kazuo Achiwa

Faculty of Pharmaceutical Sciences, University of Tokyo,
Bunkyo-ku, Tokyo 113, Japan

Abstract - Effective catalytic asymmetric synthesis of S-2, a key intermediate for the synthesis of chiral lignans, was described. Thus, BPPM-Rh⁺ in the presence of triethylamine gave S-2 in 78% optical yield.

Chiral pyrrolidinephosphine-rhodium catalysts have been proven to be practically useful for the preparation of chiral α -amino acids (83-91% optical yields)^{2,3)}, salsolidine (45%)⁴⁾, α -hydroxy esters (78.5%)^{5,6)}, R-(-)-pantolactone (80.5-86.7%)^{7,8)}, β -amino acids (53-55%)⁹⁾, α -methylsuccinic acid (94.2%)¹⁰⁻¹²⁾ and β -methylaspartic acid (58.2%)¹³⁾, and also the mechanistic studies¹⁴⁾ on these asymmetric hydrogenations suggested that the β,γ -unsaturated acid is one of the most suitable substrates for the asymmetric hydrogenations catalyzed by chiral pyrrolidinephosphine-rhodium complexes to obtain the high optical yields.

I wish to describe here an effective asymmetric hydrogenation of (E)-1¹⁵⁾, a β,γ -unsaturated acid derivative, to give S-2, a key intermediate for the synthesis of chiral lignans, steganacin, steganagin and podophyllotoxin, antileukemic agents¹⁶⁾. The reaction sequences are shown in Scheme I.

In a typical experiment, the asymmetric hydrogenation of (E)-1 (1 mmole) was run in methanol (3 ml) under an initial hydrogen pressure of 50 atm at 50°C for 20 h in the presence of [Rh(COD)BPPM]⁺ClO₄⁻ (BPPM-Rh⁺) (0.01 mmole). After removal of the solvent, the residue was treated with 3 ml of 0.5N-NaOH and the mixture was filtered to remove the catalysts. Then, the filtrate was acidified with HCl and ethereal extract gave 2, [α]_D²⁰ -27.3° (c 2.53, methanol) in a 91% isolated yield. The absolute configuration and optical purity of 2 were determined by converting (-)-2 ([α]_D²⁰ -26.3° (methanol)) into S-3¹⁷⁾, [α]_D²⁰ -2.6° (c 1.058, chloroform), on LiAlH₄ reduction. Therefore, the specific rotation of pure S-2 was calculated to

be $[\alpha]_D -35^\circ$ (methanol).

Scheme I.

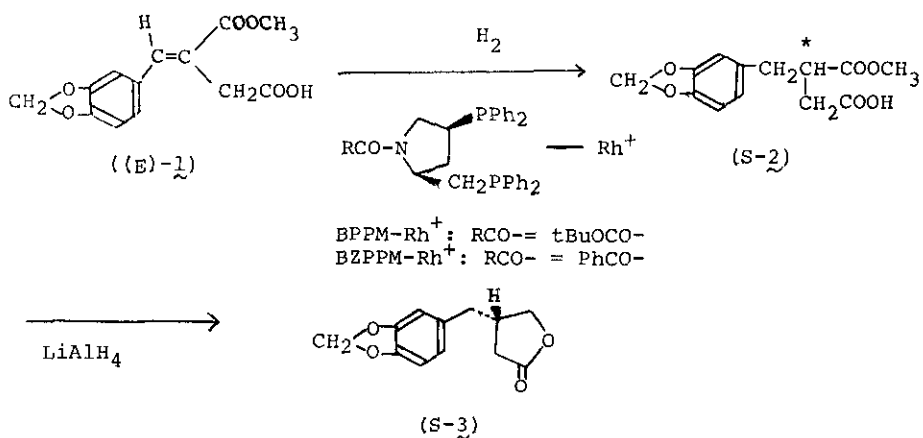


Table I. Catalytic asymmetric hydrogenation of the β,γ -unsaturated acid^{a)}

Chiral reagent (RCO-)	Solvent	$[\alpha]_D^{20}$ (methanol)	Optical y. (conf) ^{d)}
BPPM-Rh ⁺ (t-BuOCO-)	methanol	-20.1°	57% (S)
BPPM-Rh ⁺ (t-BuOCO-)	methanol ^{b)}	-27.3	78 (S)
BPPM-Rh ⁺ (t-BuOCO-)	methanol ^{b,c)}	-27.4	78 (S)
BPPM-Rh ⁺ (t-BuOCO-)	tetrahydrofuran ^{b)}	-8.1	23 (S)
BZPPM-Rh ⁺ (PhCO-)	methanol	-23.5	67 (S)
BZPPM-Rh ⁺ (PhCO-)	methanol ^{b)}	-26.3	75 (S)
BZPPM-Rh ⁺ (PhCO-)	tetrahydrofuran ^{b)}	-10.0	29 (S)

a) All hydrogenations were run with 1 mmole of substrate, 0.01 mmole of $[\text{Rh}(\text{COD})\text{bisphosphine}]^+\text{ClO}_4^-$ in 3 ml of solvent at 50°C for 20 h under an initial hydrogen pressure of 50 atm unless otherwise cited.

b) Triethylamine (0.5 mmole) was added.

c) At 20°C for 30 h.

d) $[\alpha]_D -35^\circ$ (MeOH) was used for pure S-2. See the Text.

Table I indicated clearly that BPPM-Rh⁺ gave the better optical yields than BZPPM-Rh⁺. This fact suggests that the suitable modifications of the N-substituent of the chiral ligands may improve the optical yields of the product. It should be also noted that this hydrogenation offered the practically useful route to chiral lignans¹⁶⁻¹⁹).

Further modifications of chiral catalysts and applications of the catalytic asymmetric hydrogenations catalyzed by pyrrolidinephosphine-rhodium complexes to the synthesis of chiral and biologically active compounds are actively under way.

REFERENCES AND NOTES

1. Asymmetric Reactions Catalyzed by Chiral Metal Complexes. XVII.
2. K.Achiwa, J.Am.Chem.Soc., 1976, 98, 8265.
3. K.Achiwa, Chemistry Letters, 1977, 777.
4. K.Achiwa, Heterocycles, 1977, 8, 247.
5. K.Achiwa, Tetrahedron Letters, 1977, 3735.
6. I.Ojima, T.Kogure, and K.Achiwa, J.C.S. Chem. Comm., 1977, 428.
7. K.Achiwa, T.Kogure, and I.Ojima, Tetrahedron Letters, 1977, 4431.
8. K.Achiwa, T.Kogure, and I.Ojima, Chemistry Letters, 1978, 297.
9. K.Achiwa, and T.Soga, Tetrahedron Letters, 1978, 1119.
10. K.Achiwa, Tetrahedron Letters, 1978, 1475.
11. K.Achiwa, Chemistry Letters, 1978, 561.
12. I.Ojima, T.Kogure, and K.Achiwa, Chemistry Letters, 1978, 567.
13. K.Achiwa, Tetrahedron Letters, 1978, 2583.
14. K.Achiwa, Y.Ohga, Y.Iitaka, and H.Saito, Tetrahedron Letters, 1978, 4683.
15. This stereochemistry was assigned on the basis of the Stobbe reaction mechanism.
16. F.E.Ziegler, K.W.Fowler, and N.D.Sinha, Tetrahedron Letters, 1978, 2767, and the references cited therein.
17. The maximum rotation of R-3 is $[\alpha]_D^{20} + 4.8^\circ$ (c 1.142, CHCl₃); M.Kuhn and A von Wartburg, Helv.Chim.Acta, 1967, 50, 1546. This compound (S-3) thus obtained was also converted to (2R,3S,6R)-podorhizol and (2R,3S,6S)-epi-podorhizol according to the reported conditions¹⁸).
18. E.Brown, J.P.Robin, and R.Dhal, J.C.S. Chem.Comm., 1978, 556.
19. K.Tomioka, H.Mizuguchi, and K.Koga, Tetrahedron Letters, 1978, 4687.

Received, 27th January, 1979