

SYNTHESIS OF VINCA ALKALOIDS AND RELATED COMPOUNDS XXX¹.

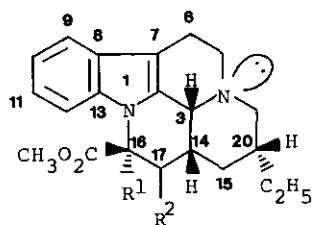
 TOTAL SYNTHESIS OF (+)-TACAMINE, (+)-APOTACAMINE AND
 THEIR 20-EPIMERS

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Abstract - Starting from iminium salt **3** two different methods have
 been elaborated for the synthesis of (+)-tacamine **1**, (+)-apotacamine **2**
 and their 20-epimers.

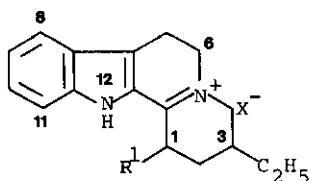
Van Beek et al. reported recently the isolation of two new alkaloids, tacamine **1**²
 and apotacamine **2**³ from the leaves of *Tabernaemontana eglandulosa*.



	R ¹	R ²
1	OH	H
2	Δ ^{16,17}	

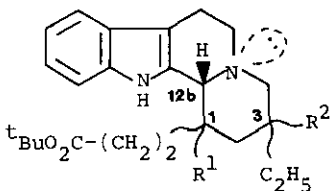
The semisynthesis of the enantiomers of these alkaloids, named pseudovincamine and pseudoapotacamine, using (-)-tabersonine as a starting material has already been published by Le Men et al.⁴

The iminium salt 3 described by Massiot et al.⁵ proved to be a suitable educt in our hands for the total synthesis of racemic 1 and 2. In order to minimize the possibility of dialkylation, 3 (X = ClO₄) was reacted with the bulky tert-butyl acrylate in methylene chloride in the presence of triethylamine affording the perchlorate 4⁶ in a yield of 79 %.



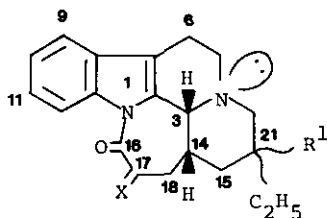
	R ¹
<u>3</u>	H
<u>4</u>	-(CH ₂) ₂ CO ₂ Bu ^t
<u>5</u>	-CH ₂ -C(OH)-CO ₂ C ₂ H ₅

The base, prepared from 4 by aqueous NaOH was reduced with NaBH₄ giving rise to three stereoisomers: 6 (yield 37.5 %)⁷, 7 (yield 10 %)⁸ and 8 (yield 30 %)⁹ after TLC separation. Their indicated configuration is substantiated by ¹H and ¹³C NMR data. It is worth to mention that very recently Lounasmaa et al. described the methyl ester of the fourth possible isomer as the only product arising from their method¹⁰.



	R ¹	R ²
<u>6</u>	β-H	α-H
<u>7</u>	α-H	α-H
<u>8</u>	β-H	β-H

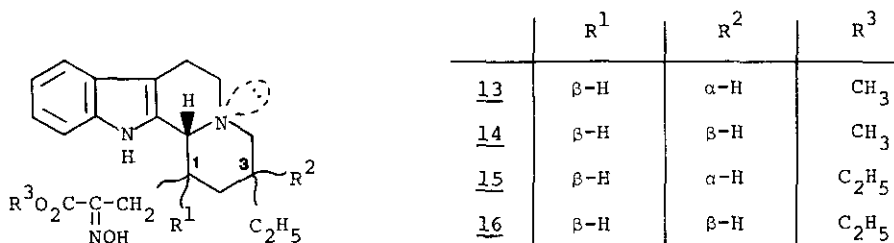
On boiling 6 or 8 in chloroform with POCl₃, lactam 9 (yield 86 %)¹¹ or 10 (yield 80 %)¹² was obtained. Both lactams were transformed to the corresponding isonitroso derivatives (11 and 12) on treatment with tert-butyl nitrite in toluene in the presence of KOBu^t. The seven membered oximes 11 (yield 49 %)¹³ and 12 (yield 53 %)¹⁴ are mixtures of Z and E isomers.



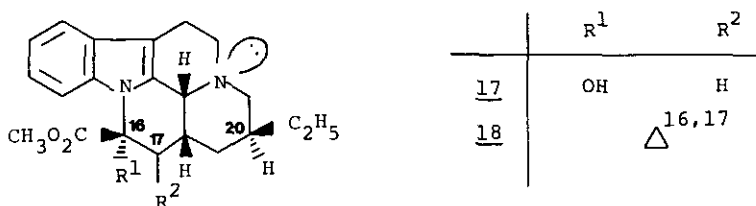
	R ¹	X
<u>9</u>	α-H	H ₂
<u>10</u>	β-H	H
<u>11</u>	α-H	=NOH
<u>12</u>	β-H	=NOH

Using the method described earlier by us¹⁵ oxime 12 was boiled in methanol with concentrated sulphuric acid affording (+)-apotacamine (2, yield 42 %) ¹⁶. Compound 11 was transformed by the same treatment to (+)-20-epi-apotacamine (18, yield 53.5 %) ¹⁷. All the spectroscopic data of 2 were identical with those described in the literature².

It was found that the above reaction sequence can be improved by boiling 11 or 12 in $\text{CH}_3\text{OH}/\text{NaOCH}_3$. As a result the oxime esters 13 (yield 60.8 %) ¹⁸ and 14 (yield 54.8 %) ¹⁹ were formed respectively being again mixtures of Z and E isomers.



When 13 was boiled in diluted acetic acid in the presence of sulphuric acid and sodium pyrosulphite (+)-20-epi-tacamine 17 was formed in 37 % yield²⁰. Under similar conditions 14 could be transformed to (+)-tacamine (1, yield 28 %) ²¹, which proved to be identical, except optical rotation, with the natural product²².



(+)-Apotacamine 2 can also be obtained by boiling oxime 14 in methanol/ H_2SO_4 , in 47 % yield.

Utilizing the reaction sequence successfully used for another model by us²³ a simple and efficient approach to 13 and 14 was found. The enamine obtained from the iminium perchlorate 3 was reacted with ethyl bromopyruvate oxime²⁴ giving rise to the iminium bromide 5 in 55.5 % yield²⁵. The latter salt was reduced by NaBH_4 in ethanol affording ester 15 (yield 80.8 %) ²⁶ which transformed to the corresponding methyl ester 13 (yield 86.6 %) on boiling in methanol in the presence

of base.

When the iminium salt 5 was kept in ethanol in the presence of sodium ethoxide at ambient temperature for half an hour, and reduced by NaBH_4 afterwards, the stereoisomer oxime esters 15 (yield 31.6 %) and 16 (yield 28.2 %) ²⁷ were isolated. Ethyl ester 16 was transformed to the corresponding methyl ester 14, using the above described method, in 81.6 % yield ²⁸.

ACKNOWLEDGEMENTS

The financial help of the Gedeon Richter Pharmaceutical Company and the Hungarian Academy of Sciences is gratefully acknowledged. The authors wish to thank É. Bihátsi for mass spectra.

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1. For part XXIX: see A. Dancsó, Gy. Kalaus, M. Kajtár-Peregy, L. Szabó and Cs. Szántay, submitted to *Acta Chim. Hung.*
2. T.A. Van Beek, P.P. Lankhorst, R. Verpoorte and A. Baerheim Svendsen, *Tetrahedron Letters*, 23, 4827 (1982).
3. T.A. Van Beek, R. Verpoorte and A. Baerheim Svendsen, *Tetrahedron*, 40, 737 (1984).
4. J. Le Men, C. Caron-Sigaut, G. Hugel, L. Le Men-Olivier and J. Lévy, *Helv. Chim. Acta*, 61, 566 (1978).
5. G. Massiot, F. Sousa Oliveira and J. Lévy, *Bull. Soc. Chim. France*, II, 185 (1982).
6. Compound 4 (X = ClO_4): mp 199-200 °C (methanol); IR (KBr) 3220 cm^{-1} (indole NH), 1720 cm^{-1} (ester CO), 1625 cm^{-1} (C=N).
7. Compound 6: mp 137-138 °C (ether); MS m/z (%) 382 (M^+ , 89.8), 353 (12.7), 326 (55.7), 325 (100.0), 309 (27.6), 267 (31.4), 225 (33.1), 170 (40.8); IR (KBr) 3400 cm^{-1} (indole NH), 1712 cm^{-1} (ester CO); ¹H NMR ²⁹ (CDCl_3) δ (ppm) 0.91 (3H, t, $-\text{CH}_2-\text{CH}_3$), 1.44 (9H, s, $-\text{C}(\text{CH}_3)_3$), 3.30 (1H, m, H-12b), 6.9-7.5 (4H, m, aromatic), 8.9 (1H, s, indole NH); ¹³C NMR (CDCl_3) δ (ppm) 11.4 ($-\text{CH}_2-\text{CH}_3$), 21.5 ($-\text{CH}_2-\text{CH}_2-\text{COOC}(\text{CH}_3)_3$), 21.9 (C_7), 27.4 ($-\text{CH}_2-\text{CH}_3$), 28.1 ($-\text{C}(\text{CH}_3)_3$), 32.0 ($-\text{CH}_2-\text{CH}_2-\text{COOC}(\text{CH}_3)_3$), 32.6 (C_3), 33.2 (C_2), 34.3 (C_1), 54.3 (C_6), 62.6 (C_4), 64.6 (C_{12b}), 80.6 ($-\text{C}(\text{CH}_3)_3$), 109.5 (C_{7a}), 111.2 (C_{11}), 117.8 (C_8), 119.0 (C_9), 119.9 (C_{10}), 127.6 (C_{7b}), 134.7 (C_{12a}), 136.8 (C_{11a}), 174.1 ($-\text{CH}_2-\text{CH}_2-\text{COOC}(\text{CH}_3)_3$).

8. Compound 7: mp 108-109 °C (ether); MS m/z (%) 382 (M⁺, 95.4), 353 (16.4), 326 (68.6), 325 (100.0), 309 (34.3), 267 (39.2), 225 (42.3), 170 (49.7); IR (KBr) 3460 cm⁻¹ (indole NH), 1700 cm⁻¹ (ester CO); ¹H NMR (CDCl₃) δ (ppm) 0.91 (3H, t, -CH₂-CH₃), 1.45 (9H, s, -C(CH₃)₃), 3.10 (1H, m, H-12b), 6.90-7.40 (4H, m, aromatic), 9.56 (1H, s, indole NH); ¹³C NMR (CDCl₃) δ (ppm) 11.4 (-CH₂-CH₃), 22.0 (C₇), 27.2 (-CH₂-CH₃), 28.1 (-C(CH₃)₃), 29.4 (-CH₂-CH₂-COOC(CH₃)₃), 32.0 (-CH₂-CH₂-COOC(CH₃)₃), 35.5 (C₃), 37.4 (C₁), 38.2 (C₂), 52.4 (C₆), 61.2 (C₄), 63.2 (C_{12b}), 81.4 (-C(CH₃)₃), 109.6 (C_{7a}), 111.1 (C₁₁), 117.8 (C₈), 118.9 (C₉), 121.1 (C₁₀), 126.9 (C_{7b}), 135.1 (C_{12a}), 136.4 (C_{11a}), 174.6 (-CH₂-CH₂-COOC(CH₃)₃).
9. Compound 8: Oil; MS m/z (%) 382 (M⁺, 95.3), 381 (35.1), 353 (16.7), 326 (53.0), 325 (100.0), 309 (31.9), 267 (35.1), 253 (15.8), 225 (34.3), 170 (43.1); IR (KBr) 3310 cm⁻¹ (indole NH), 1710 cm⁻¹ (ester CO); ¹H NMR (CDCl₃) δ (ppm) 0.85 (3H, t, -CH₂-CH₃), 1.45 (9H, s, -C(CH₃)₃), 4.05 (1H, m, H-12b), 6.90-7.50 (4H, m, aromatic), 8.32 (1H, s, indole NH); ¹³C NMR (CDCl₃) δ (ppm) 11.8 (-CH₂-CH₃), 18.0 (C₆), 27.3 (-CH₂-CH₃), 28.1 (-C(CH₃)₃), 28.1 (-CH₂-CH₂-COOC(CH₃)₃), 31.8 (-CH₂-CH₂-COOC(CH₃)₃), 33.6 (C₂), 37.8 (C₃), 39.4 (C₁), 53.2 (C₄), 59.9 (C_{12b}), 80.8 (-C(CH₃)₃), 107.1 (C_{7b}), 109.2 (C_{7a}), 111.1 (C₁₁), 117.7 (C₈), 119.2 (C₉), 121.3 (C₁₀), 132.4 (C_{12a}), 136.0 (C_{11a}), 173.6 (-CH₂-CH₂-COOC(CH₃)₃).
10. R. Jokela, S. Schüller, and M. Lounasmaa, *Heterocycles*, 23, 1751 (1985).
11. Compound 9 (HCl salt): mp 283-284 °C (methanol); MS m/z (%) 308 (M⁺, 100.0), 307 (92.0), 280 (22.7), 279 (17.2), 252 (15.8), 251 (11.1), 224 (11.1), 223 (20.4), 169 (11.6), 167 (13.3); IR (KBr) 1690 cm⁻¹ (lactam CO); ¹H NMR (CDCl₃, base) δ (ppm) 0.92 (3H, t, -CH₂-CH₃), 4.55 (1H, m, H-3), 7.20-7.50 (3H, m, H-9, H-10, H-11), 8.45 (1H, m, H-12); ¹³C NMR (CDCl₃, base) δ (ppm) 12.6 (-CH₂-CH₃), 17.1 (C₆), 23.7 (-CH₂-CH₃), 27.3 (C₁₈), 30.2 (C₂₁), 32.5 (C₁₅), 34.8 (C₁₄), 34.9 (C₁₇), 48.4 (C₂₂), 51.2 (C₅), 58.5 (C₃), 117.0 (C₁₂), 117.3 (C₇), 117.6 (C₉), 123.7 (C₁₁), 124.9 (C₁₀), 129.5 (C₈), 131.5 (C₂), 136.2 (C₁₃), 173.0 (C₁₆).
12. Compound 10 (HCl salt): mp 244-245 °C (methanol); MS m/z (%) 308 (M⁺, 100.0), 307 (81.6), 280 (23.3), 279 (19.4), 252 (14.4), 251 (9.4), 224 (11.6), 223 (19.4), 169 (10.0), 167 (13.8); IR (KBr) 1695 cm⁻¹ (lactam CO); ¹H NMR (CDCl₃, base) δ (ppm) 1.35 (3H, t, -CH₂-CH₃), 4.52 (1H, m, H-3), 7.20-7.50 (3H, m, H-9, H-10, H-11), 8.45 (1H, m, H-12); ¹³C NMR (CDCl₃, base) δ (ppm) 11.4

(-CH₂-CH₃), 17.1 (C₆), 27.0 (C₁₈), 27.3 (-CH₂-CH₃), 35.2 (C₁₅), 35.5 (C₁₇), 36.3 (C₂₁), 38.3 (C₁₄), 50.9 (C₂₂), 51.5 (C₅), 58.1 (C₃), 117.0 (C₁₂), 117.4 (C₇), 117.6 (C₉), 123.6 (C₁₁), 124.8 (C₁₀), 129.7 (C₈), 132.4 (C₂), 136.2 (C₁₃), 173.1 (C₁₆).

13. Compound 11: mp 227-230 °C (methanol); MS m/z (%) 337 (M⁺, 72.3), 336 (49.3), 320 (49.5), 318 (3.9), 309 (22.5), 308 (27.2), 307 (49.7), 293 (27.2), 292 (100.0), 251 (15.3), 237 (5.5), 223 (20.2), 169 (18.8), 168 (17.2), 167 (16.9); IR (KBr) 3350 cm⁻¹ (OH), 1700 cm⁻¹ (lactam CO), 1600 cm⁻¹ (C=N); ¹H NMR (CDCl₃) δ (ppm) 1.02 (3H, t, -CH₂-CH₃), 4.2 (1H, d, J=6.8 Hz, H-3), 7.10-7.50 (3H, m, H-9, H-10, H-11), 8.35 (1H, m, H-12), 12.0 (1H, broad s, oxime =NOH); ¹³C NMR (CDCl₃) δ (ppm) 12.5 (-CH₂-CH₃), 17.3 (C₆), 23.5 (-CH₂-CH₃), 29.1 (C₂₁), 29.9 (C₁₈), 32.4 (C₁₅), 34.6 (C₁₄), 49.8 (C₂₂), 51.1 (C₅), 57.6 (C₃), 117.5 (C₁₂), 118.0 (C₉), 119.0 (C₇), 124.3 (C₁₀), 125.1 (C₁₁), 130.5 (C₈), 131.8 (C₂), 136.4 (C₁₃), 152.7 (C₁₇), 164.4 (C₁₆).
14. Compound 12: mp 225-231 °C (ether); MS m/z (%) 337 (M⁺, 82.2), 336 (58.4), 321 (26.2), 320 (57.1), 308 (23.1), 307 (38.1), 294 (37.7), 293 (100.0), 223 (17.1), 170 (16.3), 169 (18.2); IR (KBr) 3320 cm⁻¹ (OH), 1688 cm⁻¹ (lactam CO), 1590 cm⁻¹ (C=N); ¹H NMR (DMSO-d₆) δ (ppm) 0.80 (3H, t, -CH₂-CH₃), 4.30 (1H, m, H-3), 7.15-7.55 (3H, m, H-9, H-10, H-11), 8.30 (1H, m, H-12), 11.27 (1H, broad s, oxime =NOH); ¹³C NMR (DMSO-d₆) δ (ppm) 11.0 (-CH₂-CH₃), 16.4 (C₆), 26.3 (-CH₂-CH₃), 32.8 (C₁₈), 34.8 (C₁₅), 36.4 (C₂₁), 37.7 (C₁₄), 49.9 (C₂₂), 50.7 (C₅), 57.4 (C₃), 115.7 (C₁₂), 118.1 (C₇), 118.2 (C₉), 124.0 (C₁₀), 124.7 (C₁₁), 129.8 (C₈), 132.1 (C₂), 134.7 (C₁₃), 149.3 (C₁₇), 162.5 (C₁₆).

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16. Compound 2 (HCl salt): mp 204-205 °C (methanol); MS m/z (%) 336 (M⁺, 68.1), 335 (37.0), 292 (100.0), 276 (19.7), 252 (7.6), 238 (51.3), 168 (6.6); IR (KBr) 1725 cm⁻¹ (ester CO), 1642 cm⁻¹ (C=C); ¹H NMR (CDCl₃, base) δ (ppm) 0.46 (1H, ddd, J=12.5 Hz, J=12.5 Hz, J=12.5 Hz, H-15_α), 0.85 (3H, t, -CH₂CH₃), 1.08 (2H, m, -CH₂-CH₃), 1.45 (1H, m, H-20), 1.72 (1H, ddd, J=12.5 Hz, J=2.5 Hz, J=2.5 Hz, H-15_β), 2.20 (1H, dd, J=11.0 Hz, J=11.0 Hz, H-21_α), 2.35-3.20 (4H, m, H-14, H-21_β, H-6), 3.35 (2H, m, H-5), 4.42 (1H, broad d, J=7.0 Hz, H-3), 6.38 (1H, d, J=7.3 Hz, H-17), 7.05-7.30 and 7.45 (4H, m, aromatic); ¹³C NMR (CDCl₃, base) δ (ppm) 11.3 (-CH₂-CH₃), 16.3 (C₆), 26.9

- ($-\underline{\text{C}}\text{H}_2-\text{CH}_3$), 31.6 (C_{15}), 33.4 (C_{20}), 37.2 (C_{14}), 51.1 (C_{21}), 51.1 (C_5), 52.1 (C_3), 52.5 ($-\text{COO}\underline{\text{C}}\text{H}_3$), 108.8 (C_7), 112.4 (C_{12}), 118.3 (C_9), 120.2 (C_{10}), 122.0 (C_{11}), 123.6 (C_{17}), 129.0 (C_{16}), 129.1 (C_2), 130.4 (C_8), 134.3 (C_{13}), 163.7 ($-\text{COO}\underline{\text{C}}\text{H}_3$).
17. Compound 18 (HCl salt): mp 212-214 °C (methanol); MS m/z (%) 336 (M^+ , 65.8), 335 (31.7), 293 (41.8), 292 (100.0), 277 (7.7), 276 (17.3), 239 (10.9), 238 (60.7), 168 (7.1); IR (KBr) 1723 cm^{-1} (ester CO), 1639 cm^{-1} (C=C); ^1H NMR (CDCl_3 , base) δ (ppm) 0.92 (3H, t, $-\text{CH}_2\underline{\text{C}}\text{H}_3$), 3.90 (3H, s, $-\text{O}\underline{\text{C}}\text{H}_3$), 4.38 (1H, d, $J=7.1$ Hz, H-3), 6.33 (1H, d, $J=7.1$ Hz, H-17), 7.00-7.55 (4H, m, aromatic); ^{13}C NMR (CDCl_3 , base) δ (ppm) 12.3 ($-\text{CH}_2-\underline{\text{C}}\text{H}_3$), 16.2 (C_6), 22.6 ($-\text{CH}_2-\text{CH}_3$), 28.2 (C_{20}), 28.7 (C_{15}), 34.5 (C_{14}), 47.4 (C_{21}), 51.3 (C_5), 52.1 (C_3), 52.1 ($-\text{COO}\underline{\text{C}}\text{H}_3$), 108.8 (C_7), 112.3 (C_{12}), 118.1 (C_9), 120.1 (C_{10}), 121.8 (C_{11}), 124.0 (C_{17}), 129.0 (C_{16}), 129.3 (C_8), 130.0 (C_2), 163.6 ($-\text{COO}\underline{\text{C}}\text{H}_3$).
18. Compound 13: mp 251-253 °C (methanol); MS m/z (%) 369 (M^+ , 94.0), 368 (61.0), 353 (31.0), 352 (100.0), 340 (7.1), 310 (10.0), 293 (22.1), 292 (36.0), 276 (11.0), 267 (9.0), 253 (28.0); IR (KBr) 3320 cm^{-1} (indole NH, OH), 1715 cm^{-1} (ester CO).
19. Compound 14 (HCl salt): mp 194-196 °C (methanol); MS m/z (%) 369 (M^+ , 19.0), 368 (22.0), 353 (28.1), 352 (100.0), 335 (7.0), 310 (3.0), 293 (14.0), 292 (31.0), 276 (15.0), 267 (6.0), 253 (13.0); IR (KBr) 3400 cm^{-1} (indole NH, OH), 1712 cm^{-1} (ester CO); ^1H NMR ($\text{DMSO}-d_6$, base) δ (ppm) 0.90 (3H, t, $-\text{CH}_2-\underline{\text{C}}\text{H}_3$), 3.65 (1H, m, H-3), 3.75 (3H, s, $-\text{O}\underline{\text{C}}\text{H}_3$), 6.90-7.15 (2H, m, H-10, H-11), 7.20-7.45 (2H, m, H-9, H-12), 10.4 (1H, broad s, indole NH), 11.1 (1H, broad s, oxime =NOH); ^{13}C NMR ($\text{DMSO}-d_6$, base) δ (ppm) 13.0 ($-\text{CH}_2-\underline{\text{C}}\text{H}_3$), 20.6 (C_7), 27.8 ($-\text{CH}_2-\text{CH}_3$), 30.1 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COO}\underline{\text{C}}\text{H}_3$), 31.7 (C_2), 34.2 (C_3), 36.0 (C_1), 51.8 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COO}\underline{\text{C}}\text{H}_3$), 52.8 (C_6), 57.9 (C_4), 62.7 (C_{12b}), 108.4 (C_{7a}), 111.2 (C_{11}), 117.4 (C_8), 118.3 (C_9), 120.5 (C_{10}), 127.1 (C_{7b}), 130.7 (C_{12a}), 136.5 (C_{11a}), 150.3 ($-\text{CH}_2-\underline{\text{C}}(\text{NOH})-\text{COO}\underline{\text{C}}\text{H}_3$), 163.9 ($-\text{CH}_2-\text{C}(\text{NOH})-\underline{\text{C}}\text{OO}\underline{\text{C}}\text{H}_3$).
20. Compound 17: mp 180-181 °C (methanol); MS m/z (%) 354 (M^+ , 100.0), 353 (79.0), 339 (25.3), 336 (3.6), 295 (23.0), 294 (8.6), 293 (36.6), 292 (40.6), 268 (42.3), 252 (63.0), 238 (7.3), 223 (42.3), 196 (29.0); IR (KBr) 3360 cm^{-1} (OH), 1722 cm^{-1} (ester CO); ^1H NMR (CDCl_3) δ (ppm) 0.94 (3H, t, $-\text{CH}_2-\underline{\text{C}}\text{H}_3$), 3.82 (3H, s, $-\text{O}\underline{\text{C}}\text{H}_3$), 4.30 (1H, m, H-3), 4.60 (1H, s, OH), 7.00-7.20 and 7.45 (4H, m, aromatic); ^{13}C NMR (CDCl_3) δ (ppm) 12.6 ($-\text{CH}_2-\underline{\text{C}}\text{H}_3$), 17.0 (C_6), 22.9 ($-\text{CH}_2-\text{CH}_3$), 26.9 (C_{20}), 27.8 (C_{15}), 35.2 (C_{14}), 40.2 (C_{17}), 47.3 (C_{21}),

- 50.9 (C₅), 54.3 (-COOCH₃), 54.5 (C₃), 81.8 (C₁₆), 106.4 (C₇), 110.5 (C₁₂), 118.5 (C₉), 120.3 (C₁₀), 121.7 (C₁₁), 129.0 (C₈), 130.7 (C₂), 134.5 (C₁₃), 174.5 (-COOCH₃).
21. Compound 1: mp 193-195 °C (methanol); MS m/z (%) 354 (M⁺, 100.0), 353 (85.0), 339 (28.2), 336 (8.5), 295 (33.2), 294 (52.1), 293 (79.1), 292 (48.6), 252 (48.0), 238 (12.6), 223 (34.1), 196 (23.4); IR (KBr) 3350 cm⁻¹ (OH), 1718 cm⁻¹ (ester CO); ¹H NMR (CDCl₃) δ (ppm) 0.83 (3H, t, -CH₂-CH₃), 3.79 (3H, s, -OCH₃), 4.34 (1H, m, H-3), 7.0-7.30 and 7.48 (4H, m, aromatic); ¹³C NMR (CDCl₃) δ (ppm) 11.5 (-CH₂-CH₃), 17.0 (C₆), 26.9 (-CH₂-CH₃), 31.1 (C₁₅), 32.1 (C₁₄), 38.3 (C₂₀), 40.1 (C₁₇), 50.5 (C₂₁), 50.7 (C₅), 54.2 (-COOCH₃), 54.2 (C₃), 81.8 (C₁₆), 106.2 (C₇), 110.4 (C₁₂), 118.4 (C₉), 120.2 (C₁₁), 121.7 (C₁₀), 128.8 (C₈), 130.8 (C₂), 134.5 (C₁₃), 174.3 (-COOCH₃).
22. We are indebted to Dr. T.A. van Beek for the authentic tacamine sample and the MS and NMR spectra.
23. J. Sâpi, L. Szabó, Gy. Kalaus and Cs. Szántay, unpublished results.
24. T.L. Gilchrist, D.A. Lingham and T.G. Roberts, J. Chem. Soc. Chem. Commun., 1089 (1979).
25. Compound 5 (X=Br): mp 152-154 °C (ethanol); IR (KBr) 3465 cm⁻¹, 3350 cm⁻¹ (indole NH, OH), 1702 cm⁻¹ (ester CO), 1625 cm⁻¹ (C=N).
26. Compound 15: mp 256-258 °C (water); MS m/z (%) 383 (M⁺, 90.8), 382 (52.1), 367 (25.4), 366 (100.0), 354 (7.5), 310 (13.2), 294 (14.8), 293 (32.4), 253 (36.0), 225 (23.4), 223 (17.7), 184 (41.2), 170 (72.2), 169 (58.2); IR (KBr) 3290 cm⁻¹ (indole NH, OH), 2780 cm⁻¹ (Bohlmann band), 1695 cm⁻¹ (ester CO), 1610 cm⁻¹ (C=N); ¹H NMR (DMSO-d₆) δ (ppm) 1.17 (3H, t, -CH₂-C(NO₂)-COOCH₂CH₃), 0.85 (3H, t, -CH₂-CH₃), 3.52 (1H, s, H-3), 4.0 (2H, q, -CH₂-C(NO₂)-COOCH₂CH₃), 6.80-7.10 (2H, m, H-10, H-11), 7.20-7.45 (2H, m, H-9, H-12), 10.51 (1H, s, indole NH), 12.08 (1H, s, oxime =NOH); ¹³C NMR (DMSO-d₆) δ (ppm) 11.4 (-CH₂-CH₃), 14.1 (-CH₂-C(NO₂)-COOCH₂CH₃), 21.8 (C₇), 23.5 (-CH₂-C(NO₂)-COOCH₂CH₃), 27.0 (-CH₂-CH₃), 32.6 (C₃), 34.2 (C₁), 34.4 (C₂), 53.1 (C₆), 60.8 (-CH₂-C(NO₂)-COOCH₂CH₃), 62.3 (C₄), 63.7 (C_{12b}), 108.2 (C_{7a}), 111.2 (C₁₁), 117.5 (C₈), 118.3 (C₉), 120.4 (C₁₀), 126.8 (C_{7b}), 134.5 (C_{12a}), 136.4 (C_{11a}), 151.7 (-CH₂-C(NO₂)-COOCH₂CH₃), 164.1 (-CH₂-C(NO₂)-COOCH₂CH₃).
27. Compound 16: mp 190-192 °C (ethanol); MS m/z (%) 383 (M⁺, 71.5), 382 (44.4), 366 (100.0), 310 (11.1), 292 (33.2), 253 (25.4), 225 (17.9), 223 (17.6),

184 (33.8), 170 (56.9), 169 (50.1); IR (KBr) 3300 cm^{-1} (indole NH, OH), 1703 cm^{-1} (ester CO), 1610 cm^{-1} (C=N), $^1\text{H NMR}$ (CDCl_3) δ (ppm) 0.87 (3H, t, $-\text{CH}_2-\text{CH}_3$), 1.30 (3H, t, $-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 4.23 (1H, s, H-12b), 4.24 (2H, q, $-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 7.00-7.30 (3H, m, H-10, H-11, H-12), 7.45 (1H, broad s, indole NH), 7.35-7.55 (1H, m, H-9), 9.32 (1H, broad s, oxime =NOH); $^{13}\text{C NMR}$ (CDCl_3) δ (ppm) 11.4 ($-\text{CH}_2-\text{CH}_3$), 14.0 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 16.9 (C_7), 27.1 ($-\text{CH}_2-\text{CH}_3$), 30.1 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 32.8 (C_2), 36.1 (C_3), 37.6 (C_1), 50.5 (C_4), 51.2 (C_6), 57.2 ($\text{C}_{12\text{b}}$), 62.1 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 108.7 ($\text{C}_{7\text{a}}$), 111.5 (C_{11}), 117.8 (C_8), 119.4 (C_9), 121.6 (C_{10}), 127.0 ($\text{C}_{7\text{b}}$), 131.1 ($\text{C}_{12\text{a}}$), 136.5 ($\text{C}_{11\text{a}}$), 148.1 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$), 166.5 ($-\text{CH}_2-\text{C}(\text{NOH})-\text{COOCH}_2\text{CH}_3$).

28. All products gave satisfactory elemental analysis.
29. All NMR measurements were performed on a Jeol FX-100 instrument. A detailed discussion of the NMR data will be published later on.

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