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TOTAL SYNTHESIS OF (+)-LINOXEPIN

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Dedicated to Professor Dr. Masakatsu Shibasaki on the occasion of his 70th birthday

Abstract – (+)-Linnoxepin is a lignan-type natural product with a synthetically challenging fused dihydronaphthalene/dihydrooxepine structure. It also has multiple biological activities, including antitumor, antioxidant, and antiviral activities. Here, we review three recent total syntheses of (+)-linnoxepin.

1. INTRODUCTION

Lignans are a family of structurally diverse dimeric phenylpropanoids¹⁻³ with a wide range of biological activities, including antitumor, anti-inflammatory, antioxidant, antiviral, cardiovascular and immunosuppressive activities,⁴⁻¹¹ and their synthesis has attracted much interest.^{2,12,13} Various aryl-dihydronaphthalene-type lignans, represented by **1-6**, have been isolated as natural products.¹⁵⁻²¹ Among them, (+)-linnoxepin (**1**) was isolated from the flower of *Linum perenne* L. by Schmit and co-workers in 2007.¹⁴ It has the characteristic dihydronaphthalene structure, but has a tetra-substituted double bond embedded in a highly strained dihydrooxepine ring system, which presents an interesting synthetic challenge. Recently, Tietze's group,^{22,23} Lautens's group^{24,25} and our group²⁶ have independently reported total syntheses of **1**. This review describes the methodology used by each group.

2. TIETZE'S APPROACH TO (+)-**1**: A PALLADIUM-CATALYZED DOMINO REACTION TO CONSTRUCT THE B, E RINGS OF **1**

In 2013, Tietze's group reported the first total synthesis of racemic linnoxepin (**1**).²² They subsequently synthesized **1** in optically active form.²³ Their synthetic approach is illustrated in Scheme 1. A palladium-catalyzed domino process involving carbopalladation and a Mizoroki-Heck-type reaction was

planned to construct the tetracyclic core structure of **1**. The asymmetric center at C8 would be introduced by enantioselective hydroboration reaction (Scheme 1).

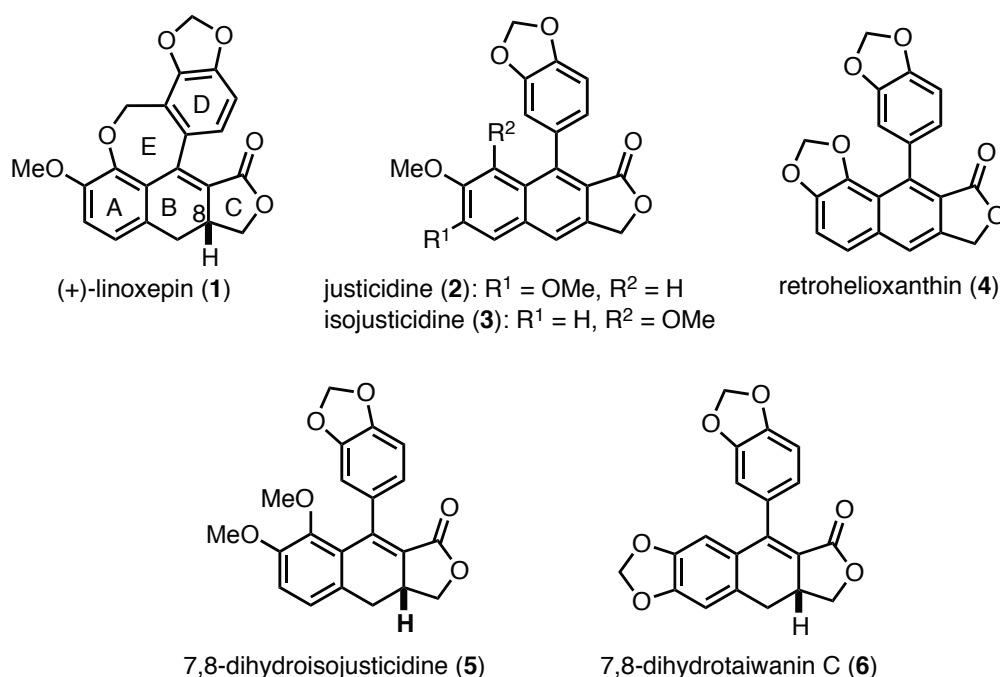
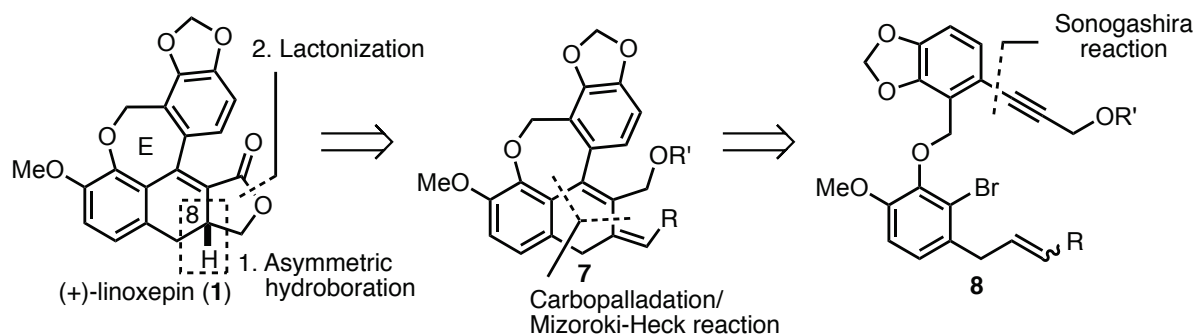


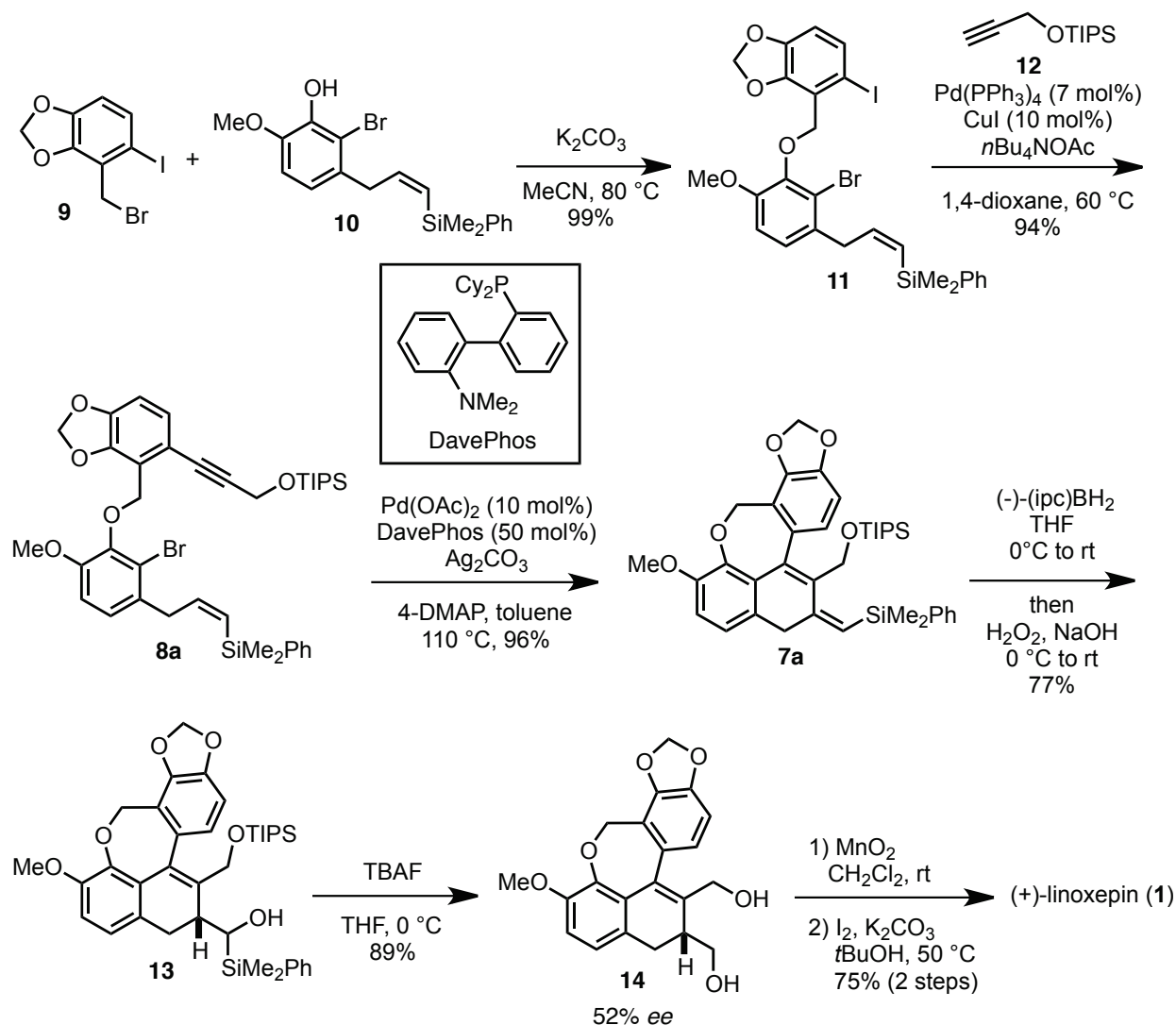
Figure 1. Structures of representative aryl(dihydro)naphthalene lignans, **1-6**



Scheme 1. Synthetic approach of **1** of Tietze's group

Alkylation of benzyl bromide **9** with phenol **10** in the presence of K₂CO₃ gave aryl ether **11** in 99% yield. Sonogashira coupling reaction of aryl iodide **11** with TIPS-protected propargylic alcohol **12** was carried out in the presence of Pd(PPh₃)₄ and CuI to give **8a**, a domino reaction precursor, in 94% yield. Then, the carbopalladation/Mizoroki-Heck domino reaction was efficiently conducted by treatment of **8a** with a catalytic amount of Pd(OAc)₂ and DavePhos as a ligand in the presence of Ag₂CO₃ and DMAP in toluene at 110 °C, affording tetracyclic **7a** in 96% yield. For the enantioselective synthesis of (+)-linoxepin (**1**), they examined asymmetric hydroboration reaction of trisubstituted alkene **7a** with Brown's chiral boranes.

In the case of (-)-(ipc)BH₂, asymmetric hydroboration reaction proceeded to afford alcohol **14** with 52% *ee* on treatment with hydroxysilane **13** and TBAF. However, it proved difficult to increase the enantioselectivity, though (-)-(ipc)₂BH and Masamune's chiral borane²⁷ were examined as alternative reagents. Total synthesis of (+)-linoxepin (**1**) was achieved from **14** by oxidation of the allylic moiety with MnO₂ followed by oxidation of the resulting aldehyde to the lactone with iodine in the presence of K₂CO₃ in 75% yield (Scheme 2).

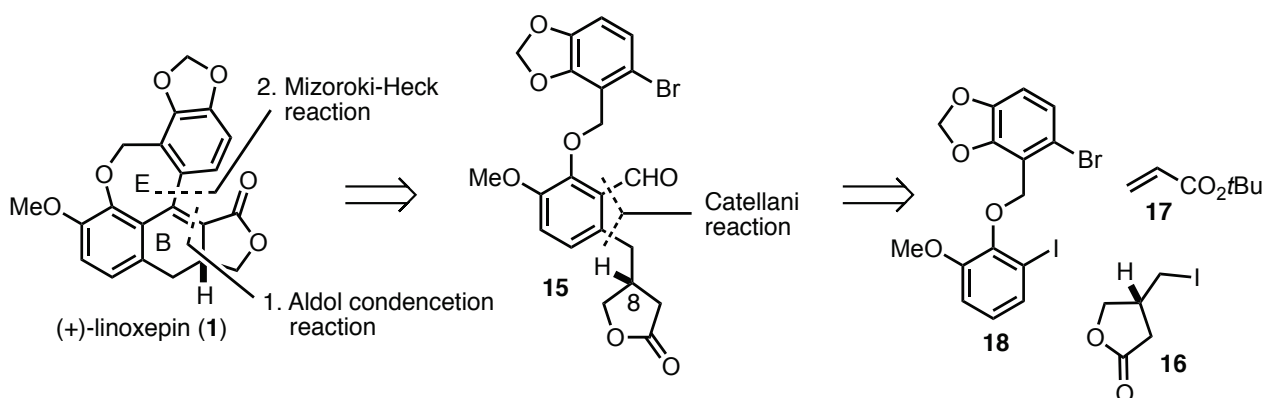


Scheme 2. Synthesis of (+)-linoxepin (**1**) by Tietze's group

The key feature of the above synthesis of (+)-**1** is the palladium-catalyzed domino reaction, involving carbopalladation and Mizoroki-Heck-type reaction of **8a** using DavePhos as a ligand. The stereogenic center at C8 was constructed by asymmetric hydroboration reaction utilizing (-)-(ipc)BH₂ with 52% *ee*.

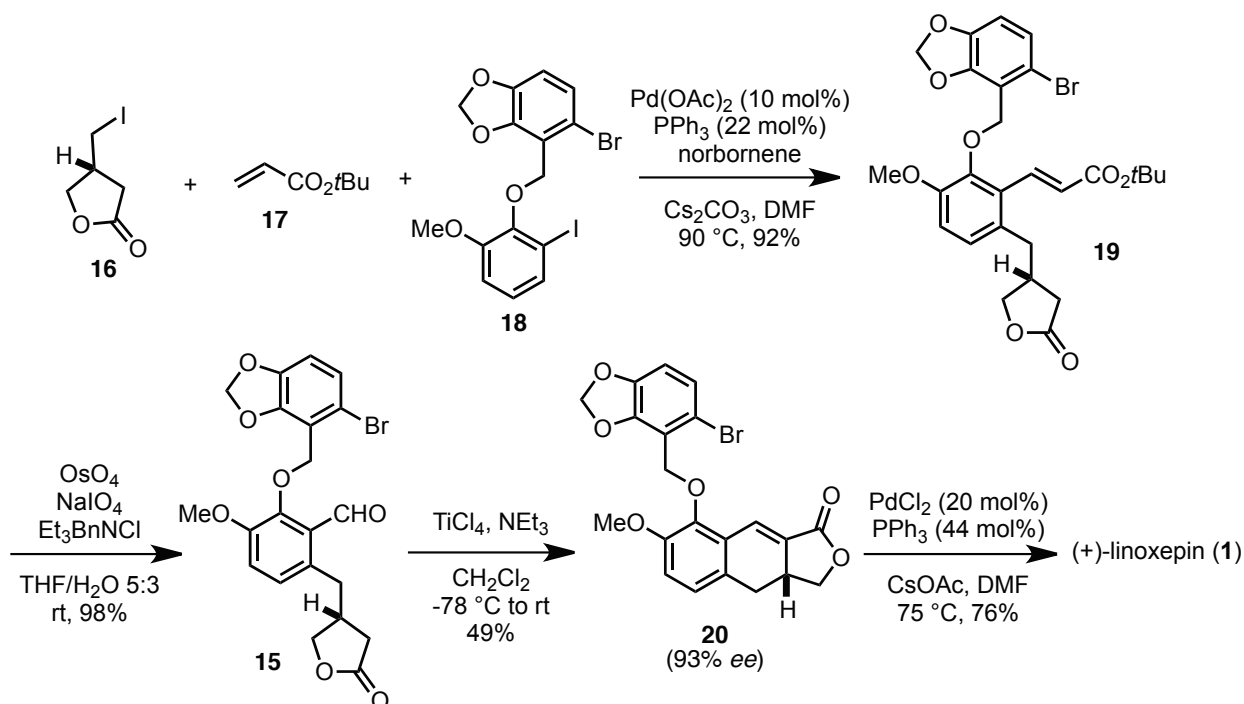
3. LAUTENS'S APPROACH TO (+)-1: CATELLANI REACTION TO CONSTRUCT THE TETRASUBSTITUTED A RING

Lautens's group has reported an enantioselective, protecting-group-free, total synthesis of (+)-linoxetine (**1**) by using palladium-catalyzed Catellani reaction as the key step.^{24,25} Their synthetic approach is illustrated in Scheme 3. They planned to obtain **15** by means of Catellani reaction and then to couple it with three other components: optically pure iodolactone **16**, acrylate **17**, and aryl ether **18**. The unsaturated lactone moiety and the E ring in **1** would be obtained by aldol condensation reaction and subsequent intramolecular Mizoroki-Heck reaction.



Scheme 3. Synthetic approach to **1** by Lautens's group

Palladium-catalyzed Catellani reaction, the key reaction in this synthesis, is a powerful carbon-carbon bond-forming reaction, which involves C-H functionalization at the *ortho*-position of aryl iodide and



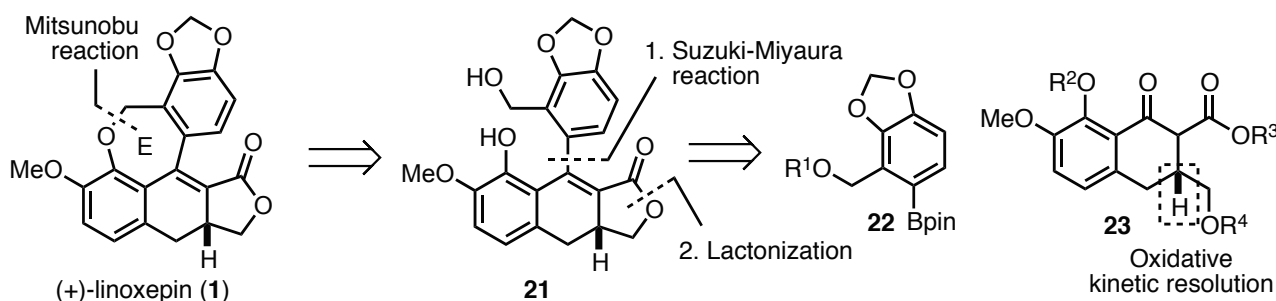
Scheme 4. Lautens's total synthesis of (+)-linoxetine (**1**)

subsequent Mizoroki-Heck-type reaction.^{28,29} This enables two adjacent positions on the aromatic ring to be functionalized in one step. Reaction of aryl iodide **18**, optically active lactone **16**, and *tert*-butyl acrylate (**17**) in the presence of a catalytic amount of Pd(OAc)₂ (10 mol%) and triphenylphosphine (22 mol%) with norbornene and cesium carbonate generated the three-component coupling product **19** in 92% yield. After oxidative cleavage of the unsaturated ester in **19** with OsO₄-NaIO₄, the resulting aldehyde **15** was subjected to aldol condensation reaction using TiCl₄ to provide unsaturated lactone **20** in 49% yield. Finally total synthesis of (+)-linoxepin (**1**) was achieved by constructing the E ring with intramolecular Mizoroki-Heck reaction in the presence of PdCl₂-PPh₃ as a catalyst in 76% yield (Scheme 4).

The key feature of this synthesis is the use of the palladium-catalyzed Catellani reaction to install all of the carbon atoms required for the synthesis of **1**. Moreover, this is the protecting-group-free synthesis in only seven steps from commercially available starting material, and is the first reported application of the Catellani reaction to natural product synthesis.

4. NAGASAWA'S APPROACH TO (+)-**1**; OXIDATIVE KINETIC RESOLUTION TO CONSTRUCT THE TETRALONE CORE STRUCTURE OF THE AB RING

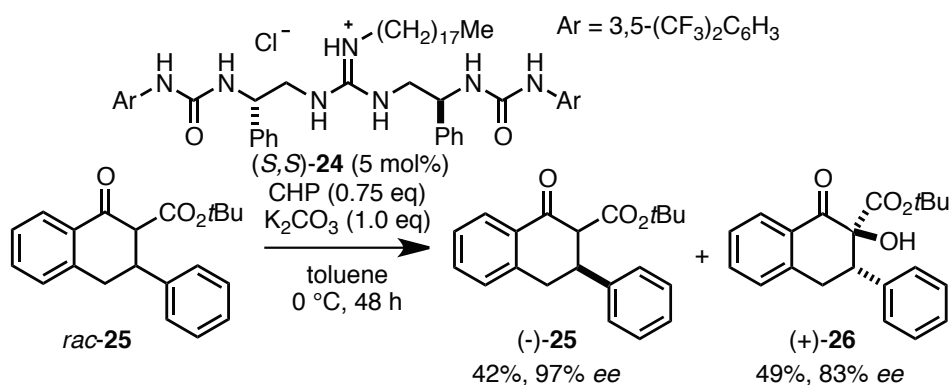
In 2015, our group reported a synthesis of (+)-linoxepin (**1**) based on the organocatalytic oxidative kinetic resolution of β -substituted tetralone.²⁶ The synthetic approach is illustrated in Scheme 5. In our synthesis, we aimed to obtain optically active tetralone **23** by oxidative kinetic resolution in the presence of a guanidine-bisurea bifunctional organocatalyst. Coupling reaction of the two segments, **22** and **23**, would be done under Suzuki-Miyaura conditions, followed by construction of the E ring by Mitsunobu reaction.



Scheme 5. Synthetic plan for (+)-**1** by Nagasawa's group

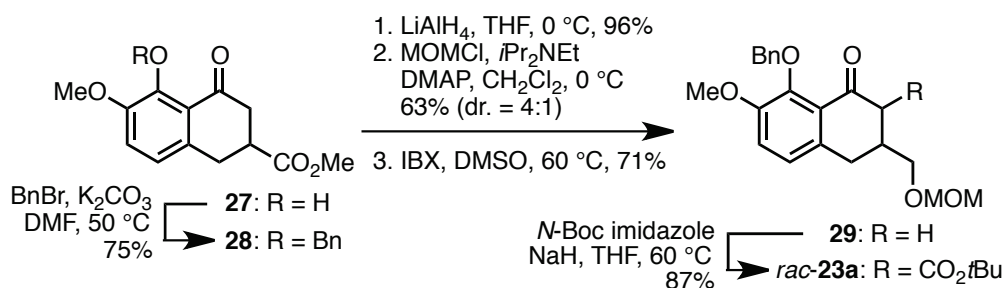
We have recently developed asymmetric α -hydroxylation of tetralone-derived β -ketoesters using guanidine-bisurea bifunctional organocatalyst (*S,S*)-**24** in the presence of cumene hydroperoxide (CHP)

as an oxidant. In the case of *rac*-**25** bearing a substituent at the β -position, kinetic resolution afforded (-)-**25** and (+)-**26** in high yield with high *ee* (Scheme 6). Thus, we planned to apply this reaction for the construction of the asymmetric center at C8 in (+)-**1**.



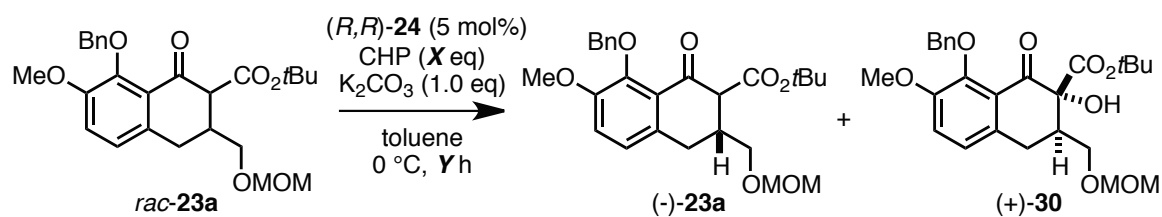
Scheme 6. Oxidative kinetic resolution of *rac*-**25** using (*S,S*)-**24**

Synthesis of the substrate *rac*-**23a** for the oxidative kinetic resolution reaction is shown in Scheme 7. After protection of the phenolic hydroxyl group in **27** with a benzyl group, ester and ketone in benzyl ether **28** were reduced with LiAlH₄ to give the diol. After selective protection of the primary alcohol with a MOM group, the secondary alcohol was oxidized with IBX to give ketone **29**. Then, β -ketoester of *rac*-**23a** was obtained from **29** by reaction with *N*-Boc imidazole in 87% yield.



Scheme 7. Synthesis of β -ketoester *rac*-**23a**

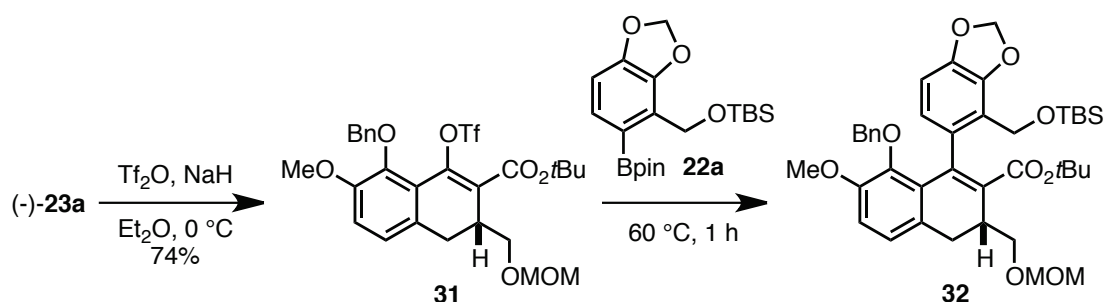
Then, the oxidative kinetic resolution reaction of *rac*-**23a** was investigated (Table 1). Under the previously optimized conditions, i.e., 0.75 equivalent of CHP for 48 hours, the desired (-)-**23a** was obtained in 44% yield with 76% *ee* (entry 1). Enantioselectivity of (-)-**23a** was improved to 99% *ee* by increasing the reaction time to 72 h (entry 3).

Table 1. Investigation of oxidative kinetic resolution of *rac*-**23a**

entry	CHP (X eq)	time (Y h)	(-)- 23a		(+)- 30		<i>s</i> ^[a]
			yield [%]	ee [%]	yield [%]	ee [%]	
1	0.75	48	44	76	49	89	39
2	1.50	48	50	91	34	87	45
3	0.75	72	37	99	52	77	39

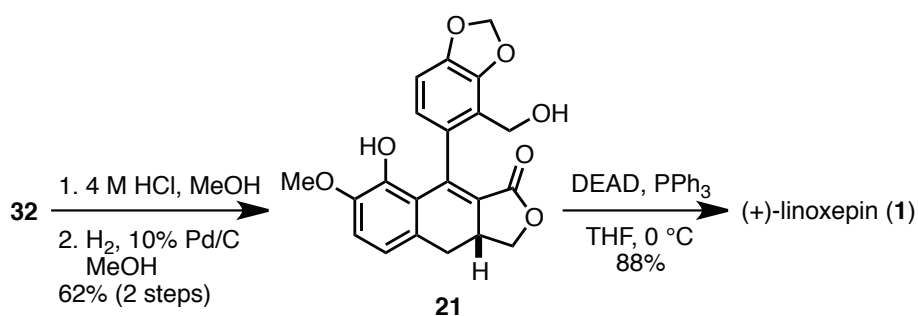
[a] The selectivity factor (*s*) was calculated as follows. $s = k_{\text{fast}}/k_{\text{slow}} = \ln[1-C(1+ee(+)-\mathbf{30})] / \ln[1-C(1-ee(+)-\mathbf{30})] = \ln[(1-C)(1-ee(-)-\mathbf{23a})] / \ln[(1-C)(1+ee(-)-\mathbf{23a})]$; $C = ee(-)-\mathbf{23a} / (ee(-)-\mathbf{23a} + ee(+)-\mathbf{30})$.

Optically active tetralone (-)-**23a** was then reacted with triflic anhydride in the presence of sodium hydride to give vinyl triflate **31** in 74% yield. Next, coupling reaction of **31** and **22a** was investigated under Suzuki-Miyaura reaction conditions in the presence of palladium catalyst (Table 2). The best result was obtained by utilizing a catalytic amount of Pd(PPh₃)₄ (5 mol%) with solid KOH as a base, and **32** was obtained in 47% yield (entry 5).

Table 2. Synthesis of **32** and investigation of Suzuki-Miyaura coupling with **31**

entry	catalyst (5 mol%)	base (5 eq)	solvent (0.2 M)	32 [%]
1	Pd(PPh ₃) ₄	2 M Na ₂ CO ₃ aq	1,4-dioxane	41
2	Pd(PPh ₃) ₄	Na ₂ CO ₃ (solid)	1,4-dioxane	trace
3	Pd(PPh ₃) ₄	2 M Na ₂ CO ₃ aq	toluene	trace
4	Pd(OAc) ₂	2 M Na ₂ CO ₃ aq	1,4-dioxane	trace
5	Pd(PPh ₃) ₄	KOH (solid)	1,4-dioxane	47

Total synthesis of (+)-linoxepin (**1**) from **32** was completed as follows. The MOM, TBS, and *tert*-butyl ester groups in **32** were removed under acidic conditions, and then deprotection of Bn ether with hydrogen in the presence of 10% Pd/C provided **21** in 62% yield in two steps. Finally, the E ring was constructed under Mitsunobu reaction conditions by utilizing diethyl azodicarboxylate (DEAD) and triphenylphosphine to give (+)-**1** in 88% yield (Scheme 8).



Scheme 8. Total synthesis of (+)-**1** by Nagasawa and co-workers

Thus, (+)-**1** was synthesized in 11 steps from the known tetralone **27** by utilizing organocatalytic oxidative kinetic resolution of *rac*-**23a**. This approach should be applicable to a variety of natural products containing tetralone structures with asymmetric centers at the β -position.

CONCLUSION

In this article, we have reviewed three recent syntheses of (+)-linoxepin (**1**) via distinct approaches, i.e., palladium-catalyzed domino reaction for the construction of the A,B,D,E ring system, palladium-catalyzed three-component coupling-type Catellani reaction, and oxidative kinetic resolution with an organocatalyst. These three approaches should be applicable to synthesize a range of lignan-type natural products and their derivatives, which should facilitate detailed structure-activity relationship studies of these multi-functional molecules.

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REFERENCES

1. D. A. Whiting, *Nat. Prod. Rep.*, 1985, **2**, 191.
2. J. Y. Pan, S. L. Chen, M. H. Yang, J. Wu, J. Sinkkonen, and K. Zou, *Nat. Prod. Rep.*, 2009, **26**, 1251.
3. J. Zhang, J. Chen, Z. Liang, and C. Zhao, *Chem. Biodivers.*, 2014, **11**, 1.
4. H. Lu and G.-T. Liu, *Planta Med.*, 1992, **58**, 311.
5. T. Hirano, M. Gotoh, and K. Oka, *Life Sci.*, 1994, **55**, 1061.
6. L. U. Thompson, M. M. Seidl, S. E. Rickard, L. J. Orcheson, and H. H. Fong, *Nutr. Cancer*, 1996, **26**, 159.
7. E. L. Ghisalberti, *Phytomedicine*, 1997, **4**, 151.
8. J. L. Charlton, *J. Nat. Prod.*, 1998, **61**, 1447.
9. D. D. Kitts, Y. V. Yuan, A. N. Wijewickreme, and L. U. Thompson, *Mol. Cell. Biochem.*, 1999, **202**, 91.
10. L. Kangas, N. Saarinen, M. Mutanen, M. Ahotupa, R. Hirsinummi, M. Unkila, M. Perala, P. Soininen, R. Laatikainen, H. Korte, and R. Santti, *Eur. J. Cancer Prev.*, 2002, **11**, 48.
11. S. Yamauchi, T. Ina, T. Kirikihira, and T. Masuda, *Biosci. Biotechnol. Biochem.*, 2004, **68**, 183.
12. R. S. Ward, *Chem. Soc. Rev.*, 1982, **11**, 75.
13. J. S. Sun, H. Liu, X. H. Guo, and J. X. Liao, *Org. Biomol. Chem.*, 2016, **14**, 1188.
14. T. J. Schmidt, S. Vossing, M. Klaes, and S. Grimme, *Planta Med.*, 2007, **73**, 1574.
15. A. Mohagheghzadeh, T. J. Schmidt, and A. W. Alfermann, *J. Nat. Prod.*, 2002, **65**, 69.
16. N. Vasilev, P. Nedialkov, I. Ionkova, and S. Ninov, *Archiv. Pharm.*, 2004, **59**, 528.
17. N. Vasilev and I. Ionkova, *Pharm. Biol.*, 2005, **43**, 509.
18. B. Konuklugil, I. Ionkova, N. Vasilev, T. J. Schmidt, J. Windhovel, E. Fuss, and A. W. Alfermann, *Nat. Prod. Res.*, 2007, **21**, 1.
19. N. Vasilev, R. Ebel, R. Edrada, E. Fuss, A. W. Alfermann, I. Ionkova, A. Petrova, M. Repplinger, and T. J. Schmidt, *Planta Med.*, 2008, **74**, 273.
20. A. Mohagheghzadeh, S. Dehshahri, and S. Hemmati, *Z. Naturforsch. C*, 2009, **64**, 73.
21. T. J. Schmidt, S. Hemmati, M. Klaes, B. Konuklugil, A. Mohagheghzadeh, I. Ionkova, E. Fuss, and A. W. Alfermann, *Phytochemistry*, 2010, **71**, 1714.
22. L. F. Tietze, S. C. Dufert, J. Clerc, M. Bischoff, C. Maass, and D. Stalke, *Angew. Chem. Int. Ed.*, 2013, **52**, 3191.
23. L. F. Tietze, J. Clerc, S. Biller, S. C. Dufert, and M. Bischoff, *Chem. Eur. J.*, 2014, **20**, 17119.
24. H. Weinstabl, M. Suhartono, Z. Qureshi, and M. Lautens, *Angew. Chem. Int. Ed.*, 2013, **52**, 5305.

25. Z. Qureshi, H. Weinstabl, M. Suhartono, H. Liu, P. Thesmar, and M. Lautens, *Eur. J. Org. Chem.*, 2014, 4053.
 26. M. Odagi, K. Furukori, Y. Yamamoto, M. Sato, K. Iida, M. Yamanaka, and K. Nagasawa, *J. Am. Chem. Soc.*, 2015, **137**, 1909.
 27. S. Masamune, B. M. Kim, J. S. Petersen, T. Sato, S. J. Veenstra, and T. Imai, *J. Am. Chem. Soc.*, 1985, **107**, 4549.
 28. M. Catellani, F. Frignani, and A. Rangoni, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 119.
 29. E. Motti, M. Rossetti, G. Bocelli, and M. Catellani, *J. Organomet. Chem.*, 2004, **689**, 3741.
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